



Measuring the electric dipole moment of the electron using polar molecules in a parahydrogen matrix

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ABSTRACT

This paper outlines the groundwork we have done in preparation for our proposed experiment to measure the electric dipole moment of the electron in barium monofluoride molecules within a cryogenic parahydrogen matrix. We describe the experimental setup we use to grow and characterize the cryogenic crystals and produce a molecular source of barium monofluoride. The technique we present has the potential to improve the sensitivity to the electric dipole moment of the electron by increasing the number of molecules under measurement without affecting their spin coherence times. If successful, this approach would represent a significant advancement in precision measurements using molecules in cryogenic solid matrices.

1. Introduction

Studying the electric dipole moment of the electron (eEDM) is a promising path to explore physics beyond the Standard Model [1–4]. A non-zero eEDM measurement would indicate a violation of time-reversal and CP symmetry, potentially shedding light on unresolved phenomena like matter-antimatter asymmetry, dark matter, and dark energy.

The most common experimental method for measuring the eEDM is to observe changes in the precession rate of electron spins in an electric field. The sensitivity of this measurement depends on the strength of the electric field experienced by the electrons, their spin coherence time, and the number of electrons examined. Currently, the most stringent

limit for the eEDM is $d_e \leq 4.1 \times 10^{-30}$ e cm, obtained using electrons confined in HfF^+ molecular ions [5].

To improve upon this limit, the PHYDES Collaboration plans to use heavy polar molecules, such as barium monofluoride (BaF), that have a high internal effective electric field (≈ 10 GV/cm) and embed them in a cryogenic parahydrogen (p-H_2) crystal matrix. By trapping the BaF molecules in a matrix, we can increase the number of molecules in the measurement volume by several orders of magnitude, thus improving the sensitivity to the eEDM effect [6]. Furthermore, the unique properties of the parahydrogen matrix, such as a large intermolecular distance and the absence of permanent electric moments [7], minimize the interaction with the guest molecules, potentially granting them a

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narrow transition linewidth and long coherence time. Recently, coherence times as long as 0.1 s for the hyperfine transitions of rubidium atoms embedded in p-H₂ have been reported in literature [8].

2. The PHYDES experiment

Current efforts within the PHYDES collaboration have focused on producing and characterizing BaF-doped p-H₂ crystals in preparation for the eEDM measurement. To produce BaF molecules for our experiment, we designed a dedicated setup in which solid BaF₂ powder is evaporated in a glow discharge chamber in a commercial ion source. The resulting BaF₂ molecules are then dissociated and ionized by an electron beam, accelerated to 1 keV, and passed through a velocity filter to isolate a beam of BaF⁺. These ions are decelerated and combined with p-H₂ gas to form a solid matrix. In a preliminary test, we achieved a BaF⁺ current of about 20 nA at an energy of 5 eV.

To prevent charge accumulation within the matrix and ensure a high density of molecules, the BaF⁺ ions must be neutralized. We plan to achieve this by injecting electrons into the matrix. These electrons will be photoextracted by UV laser pulses from a gold layer coated on the substrate where we grow the cryogenic crystal.

To investigate the feasibility of our photoextraction and neutralization approach, we studied the dynamics of quasi-free electrons extracted from a gold photocathode and injected into solid parahydrogen. We measured their thermalization length in the matrix and showed that these low-energy electrons rapidly thermalize within the bulk of the matrix, where they would become available for the neutralization process [9].

The cryogenic system for growing p-H₂ crystal matrices (shown in Fig. 1) follows the methods described in [10,11]. High-purity hydrogen gas cooled to around 20 K by a pulse tube refrigerator flows through a copper line containing hydrous ferric oxide, which enhances the conversion to the p-H₂ state [12]. The p-H₂ gas flows at approximately 10 standard cubic centimeters per minute toward a cold sapphire substrate coated with a 10 nm gold layer, where it condenses to form a solid crystal. The substrate, mounted on a copper plate cooled to around 2.9 K using a two-stage helium pulse tube cryocooler, is housed in a six-way cross stainless steel chamber. This chamber is pumped by a turbomolecular system at the bottom and one on the top to expedite hydrogen removal from the upper section of the vacuum chamber. Before starting the growth process, the system is evacuated to a base pressure below 10⁻⁸ mbar and, during crystal growth, the pressure is maintained constantly below 5 × 10⁻⁷ mbar.

The thickness and ortho-hydrogen fraction of the p-H₂ crystal were routinely monitored during growth using infrared absorption spectroscopy, as described in [9]. The crystal growth rate was approximately 100 μm/h with an ortho-hydrogen fraction of about 2%–3%. The crystals were grown to thicknesses of about 500–600 μm.

3. Conclusions

The innovative approach of embedding polar molecules with high internal electric fields in cryogenic crystal matrices shows promise in improving the eEDM detection limits and uncovering physics beyond the Standard Model. In this context, the PHYDES Collaboration is pioneering the combined use of isotopically selected BaF molecules produced in a custom-made molecular source and a p-H₂ host matrix. The advantage of using parahydrogen as a host matrix is that it has a larger intermolecular distance and lattice constant compared to other commonly used cryogenic matrices (such as Ne or Ar), so it provides more free space for the guest molecules and weaker interactions with the surrounding host molecules.

We have built a molecular source producing up to 20 nA of BaF⁺ molecules at an energy of 5 eV and demonstrated the ability to grow optically transparent p-H₂ crystals 500–600 μm thick.

Future work will focus on improving the production and characterization of BaF-doped p-H₂ crystals and developing techniques to manipulate and selectively target the electronic states of BaF molecules within the p-H₂ crystal necessary for the eEDM measurement.

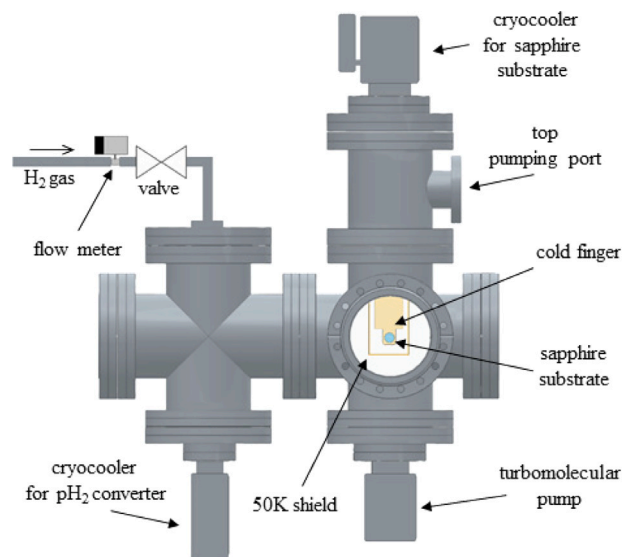


Fig. 1. Schematic illustration of the experimental setup used for crystal growth: hydrogen gas flows through a copper line containing a hydrous ferric oxide catalyst, which is maintained at around 20 K. The substrate on which the p-H₂ is grown is cooled to approximately 2.9 K by another cryocooler. The growth chamber, a six-way cross stainless-steel chamber, is pumped by turbomolecular pumps.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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References

- [1] J. Baron, W.C. Campbell, D. DeMille, J.M. Doyle, G. Gabrielse, Y.V. Gurevich, P.W. Hess, N.R. Hutzler, E. Kirilov, I. Kozyryev, B.R. O’Leary, C.D. Panda, M.F. Parsons, E.S. Petrik, B. Spaun, A.C. Vutha, A.D. West, Order of magnitude smaller limit on the electric dipole moment of the electron, *Science* 343 (6168) (2014) 269–272, <https://www.science.org/doi/abs/10.1126/science.1248213>.
- [2] V. Andreev, D.G. Ang, D. DeMille, J.M. Doyle, G. Gabrielse, J. Haefner, N.R. Hutzler, Z. Lasner, C. Meisenhelder, B.R. O’Leary, C.D. Panda, A.D. West, X. West, Improved limit on the electric dipole moment of the electron, *Nature* 562 (7727) (2018) 355–360, <http://dx.doi.org/10.1038/s41586-018-0599-8>.
- [3] C.J. Ho, J.A. Devlin, I.M. Rabej, P. Yzombard, J. Lim, S.C. Wright, N.J. Fitch, E.A. Hinds, M.R. Tarbutt, B.E. Sauer, New techniques for a measurement of the electron’s electric dipole moment, *New J. Phys.* 22 (5) (2020) 053031, <http://dx.doi.org/10.1088/1367-2630/ab83d2>.
- [4] P. Aggarwal, H.L. Bethlem, A. Borschevsky, M. Denis, K. Esajas, P.A.B. Haase, Y. Hao, S. Hoekstra, K. Jungmann, M.C. Meijknecht, R.G.E. Timmermans, W. Ubachs, L. Willmann, A. Zapara, Measuring the electric dipole moment of the electron in BaF, *Eur. Phys. J. D* 72 (197) (2018) <http://dx.doi.org/10.1140/epjd/e2018-90192-9>.
- [5] T.S. Roussy, L. Caldwell, T. Wright, W.B. Cairncross, Y. Shagam, K.B. Ng, N. Schlossberger, S.Y. Park, A. Wang, J. Ye, E.A. Cornell, An improved bound on the electron’s electric dipole moment, *Science* 381 (6653) (2023) 46–50, <https://www.science.org/doi/abs/10.1126/science.adg4084>.
- [6] A.C. Vutha, M. Horbatsch, E.A. Hessels, Oriented polar molecules in a solid inert-gas matrix: A proposed method for measuring the electric dipole moment of the electron, *Atoms* 6 (1) (2018) <https://www.mdpi.com/2218-2004/6/1/3>.

- [7] T. Momose, H. Hoshina, M. Fushitani, H. Katsuki, High-resolution spectroscopy and the analysis of ro-vibrational transitions of molecules in solid parahydrogen, *Vib. Spectrosc.* 34 (1) (2004) 95–108, <http://dx.doi.org/10.1016/j.vibspec.2003.06.001>.
- [8] S. Upadhyay, U. Dargyte, D. Patterson, J.D. Weinstein, Ultralong spin-coherence times for rubidium atoms in solid parahydrogen via dynamical decoupling, *Phys. Rev. Lett.* 125 (2020) 043601, <https://link.aps.org/doi/10.1103/PhysRevLett.125.043601>.
- [9] A.F. Borghesani, G. Carugno, G. Messineo, J. Pazzini, Electron thermalization length in solid para-hydrogen at low-temperature, *J. Chem. Phys.* 159 (10) (2023) 104501, <http://dx.doi.org/10.1063/5.0163776>.
- [10] B.A. Tom, S. Bhasker, Y. Miyamoto, T. Momose, B.J. McCall, Producing and quantifying enriched para-H₂, *Rev. Sci. Instrum.* 80 (1) (2009) 016108, <http://dx.doi.org/10.1063/1.3072881>.
- [11] A. Bhandari, A.P. Rollings, L. Ratto, J.D. Weinstein, High-purity solid parahydrogen, *Rev. Sci. Instrum.* 92 (7) (2021) 073202, <http://dx.doi.org/10.1063/5.0049006>.
- [12] D.H. Weitzel, W.V. Loebenstein, J.W. Draper, O.E. Park, Ortho-para catalysis in liquid-hydrogen production, *J. Res. Natl. Bur. Stand.* 60 (3) (1958) 221–227.