

Unraveling the molecular mechanisms of vimentin interaction with G-quadruplex repeats

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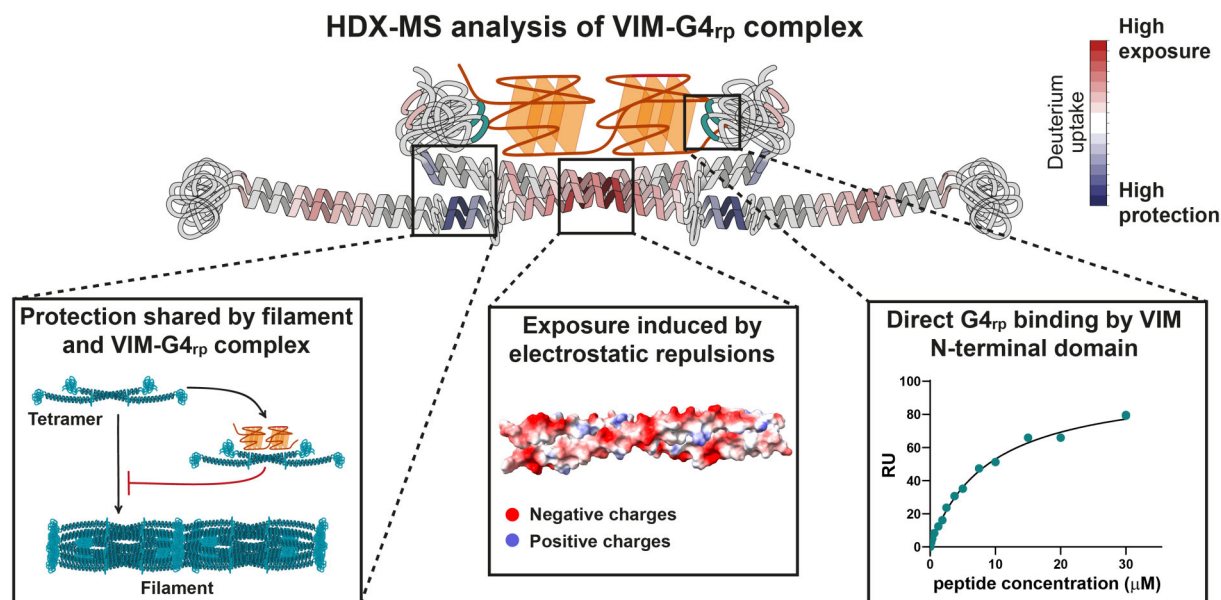
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Abstract

Vimentin is a cytoskeletal protein that exists as soluble tetramers or polymerized filaments. Its tetrameric form selectively interacts with DNA non-canonical secondary structures called G-quadruplex repeats. Here, we applied optimized experimental conditions to preferentially trap the protein at selected oligomerization states. All of them were analyzed by hydrogen–deuterium exchange mass spectrometry to map the vimentin domains that undergo structural changes upon DNA binding and compare them to those associated with its polymerization. G-quadruplex repeats induced unique perturbations of vimentin structure. Among them, the exposure of coil-1B was the most relevant and it had been associated with a reduction in the protein α -helical content. In addition, few protein structural changes were observed both upon G-quadruplex repeats binding and vimentin polymerization, such as the stabilization of coil-1A and coil-2A. This suggests that DNA complex formation impairs the contribution of these regions to vimentin oligomerization. Finally, a peptide from the protein N-terminal domain (residues 11–30) has been identified as a specific G-quadruplex binder, underscoring its direct role in stabilizing the vimentin–DNA complex. Overall, these findings provide detailed insights into the molecular mechanisms that allow DNA to finely modulate the composite structural equilibrium of vimentin and, accordingly, its functions at nuclear level.

Graphical abstract



Introduction

The eukaryotic cytoskeleton is a complex system comprising three major components: microfilaments, intermediate filaments (IF), and microtubules [1]. These highly ordered com-

ponents are formed by protein monomers that non-covalently associate to generate linear filaments or tubules, with thickness ranging from 7 to 25 nm [2]. Due to the rapid assembly and disassembly of these proteins, the architecture and

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composition of the cytoskeleton are highly dynamic, thus enabling it to regulate multiple cellular features, including cell shape, division, and motility [3, 4]. Although cytoskeletal proteins are primarily localized in the cytoplasm, it is now well established that some of them are also present in the nucleus, where they participate in chromatin organization, chromosome condensation, transcription regulation, DNA replication, and repair [5–8].

Within this heterogeneous group of proteins, the type III intermediate filament protein vimentin (VIM) is among the earliest cytoskeletal components expressed during embryogenesis. After this initial phase, VIM is gradually replaced by tissue-specific IFs, and its postnatal expression becomes largely restricted to a few cell types, such as mesenchymal cells [9]. VIM participates in the fine-tuning of a variety of physiological and pathological processes, including axon regeneration, immune response, and pathogen invasion [10–14]. In particular, in cancer cells, VIM overexpression is strongly associated with tumor growth and metastasis development [15]. This outcome is closely related to VIM's role as a key driver of the epithelial-to-mesenchymal transition (EMT), a process in which polarized epithelial cells acquire a motile phenotype with enhanced metastatic potential [16, 17]. As a result, a high expression level of VIM is a validated marker of poor prognosis for oncologic patients [18].

From a structural point of view, the monomeric unit of VIM comprises intrinsically disordered N- and C-terminal domains, along with two central α -helical domains, known as coil-1 and coil-2. Each coil is further divided into two segments, A and B. Two flexible unfolded linkers, named L1 and L12, are located between coil-1A and coil-1B and between coil-1 and coil-2, respectively [19]. VIM is never found as a monomeric protein. Its assembly begins when two monomers align in parallel to form a coiled-coil dimer. Subsequently, two dimers associate laterally with variable relative orientations to generate tetramers, which constitute the soluble cellular form of VIM. Cross-linking analyses identified four main pairing modes: A₁₁, A₂₂, A₁₂, and A_{CN} [20, 21]. Among these, the predominant arrangement in solution is the A₁₁ tetramer, formed by two dimers aligned in an antiparallel orientation with coil-1B in register [22]. Subsequent association of tetramers generates unit-length filaments (ULF), which further anneal longitudinally to form mature intermediate filaments. In cells, the equilibrium between filaments and tetramers is regulated by VIM post-translational modifications, such as phosphorylation and glycosylation at selected sites [23–25].

Although VIM is primarily localized in the cytoplasm, in its soluble tetrameric form it has also been detected in the nucleus of metastatic cancer cells, where it plays specialized functions that extend beyond its conventional cytoskeletal roles [26]. For example, in polyploid giant cancer cells, ChIP-seq analyses demonstrated that nuclear VIM acts as a transcriptional regulator to promote the expression of genes involved in cell invasion and migration [26]. Notably, some of these regulatory functions rely on the assembly of VIM with additional proteins. Recently, VIM has been reported to associate with Ku80, contributing to the regulation of DNA repair through the non-homologous end joining (NHEJ) pathway and thereby underscoring its direct role in maintaining genome integrity [27]. In addition, VIM itself can act as an unconventional DNA-binding protein. The earliest proof of a direct interaction between VIM and DNA comes from *in vitro* studies showing its affinity for highly repetitive G-rich sequences

and, consistently, immunoprecipitation assays performed in mouse fibroblasts revealed its binding to centromeric satellite and telomeric regions [28, 29]. These genomic sequences are known to potentially fold into non-canonical secondary structures called G-quadruplex (G4) [30, 31]. They are characterized by consecutive planar arrays of four guanine residues paired through Hoogsteen hydrogen bonds that interact with each other by π - π stacking interactions. The resulting tetrahelical core is further stabilized by the coordination of monovalent cations, with a preference for potassium ions [32]. As structural elements, G4s are known to orchestrate protein recruitment at genomic level, thereby contributing to the regulation of gene expression [33].

Recently, we proved that the interaction between VIM and selected DNA sites is not dependent on the nucleic acid sequence but on its secondary structure. Notably, VIM tetramers are unable to bind single-stranded, double-stranded or isolated G4. Conversely, it recognizes with high affinity and selectivity DNA domains where at least two G4 units are in close proximity and generates the higher-order arrangement referred to as G4 repeat (G4_{rp}) [34]. Analysis of multiple sequences known to fold into G4_{rps}—including telomeric repeats of different lengths and sequences from oncogenic promoters—revealed that this interaction is independent of both the topology and the number of G4 units within the G4_{rp} [34–37]. Since bioinformatic analyses indicated that G4_{rps} are enriched in promoter regions of EMT-related genes, the VIM–G4_{rp} complex has been proposed as a promising target for the development of novel anticancer therapies [34, 35]. However, the structural features of this complex remain to be fully elucidated.

As far as it concerns the protein, the Cryo-EM structure of VIM filament has been recently reported [38]. Additionally, X-ray crystal structures of selected domains—coil-1A (PDB 1GK7), coil-1B (PDB 3SWK and 3UF1), coil-2B (PDB 1GK4), and fragments such as coil-1A–L1–coil-1B (PDB 3SSU) and coil-2A–coil-2B (PDB 3KLT) [39–42]—are available. In contrast, the high-resolution structure of the soluble full-length tetramer has not been solved yet. Thus, essential structural details to fully understand the VIM structure-function relationship are still missing. However, hydrogen–deuterium exchange mass spectrometry (HDX-MS) has been successfully exploited to monitor VIM structural evolution from tetramers to filaments [43].

Building on this evidence and on the relevance of VIM–DNA complexes in the nuclear environment, here we applied HDX-MS to identify the VIM domains involved in G4_{rp} recognition and to compare the conformational changes induced by DNA binding with those associated with filament assembly. As a nucleic acid model, we selected a telomeric sequence comprising eight telomeric repeats [d(TTAGGG)₈, 2TEL], whose higher-order G4_{rp} architecture and interaction with VIM have already been validated both *in cells* and *in vitro* [28, 34, 36].

Materials and methods

Protein expression and purification

The sequence that codifies for the full-length human VIM or its Δ 30 truncated form (VIM Δ 30) was inserted into the pET24a+ vector by GenScript. The plasmid was used to transform BL21(DE3) *Escherichia coli* competent cells

(Invitrogen™). The transformed cells were grown at 37°C in LB media until an OD₆₀₀ of 0.7 was reached, and then the expression of the proteins was induced by adding 1 mM isopropyl-β-D-1-thiogalactopyranoside (IPTG). VIM and VIMΔ30 were extracted and purified following a previously described protocol [44]. Briefly, cells were centrifuged at 7000 × *g* for 20 min at 4°C, resuspended in 10 mM Tris, 2 mM phenylmethylsulfonyl fluoride (PMSF), and 10 mM MgCl₂, pH 7.5. After resuspension, the cells were sonicated, and Triton X-100 and DNase I (Thermo Scientific™) were added up to 0.2% and 50 g/ml, respectively. The inclusion bodies were harvested by centrifugation and resuspended in washing buffer [10 mM Tris, 1 mM ethylenediaminetetraacetic acid (EDTA), 1 mM DTT, 2 mM PMSF, 1.5 M KCl, pH 7.5]. After centrifugation, this procedure was repeated with the same buffer in the absence of KCl. Finally, the pellet was washed with 10 mM Tris, 0.1 mM EDTA, pH 7.5, and the inclusion bodies were resuspended in a denaturing buffer (9.5 M urea in 10 mM Tris, pH 7.5). The solubilized IF proteins were collected by centrifugation at 64 000 × *g* for 1 h at 20°C. VIM and VIMΔ30 were purified from the supernatant through two sequential steps of ion-exchange chromatography, first using an anionic (Q Sepharose FF, Cytiva) followed by a cationic (SP Sepharose FF, Cytiva) column. The elution buffer was 10 mM Tris, 8 M urea, 1 mM DTT, pH 7.5, with a NaCl gradient from 0 to 500 mM NaCl. The quality of the purified proteins was evaluated by sodium dodecyl sulfate–polyacrylamide gel electrophoresis and MS. MS analyses of VIM confirmed that Met1 was not present in the purified product; those of VIMΔ30 showed that the first 30 amino acids were missing. Purified proteins were frozen and stored at –80°C. The day before the use, proteins were renatured following the protocol developed by Herrmann and colleagues [44]. Accordingly, the proteins were dialyzed at room temperature for 1 h against dialysis buffers [5 mM Tris-HCl (pH 8.4), 1 mM EDTA, 0.1 mM EGTA, and 1 mM dithiothreitol] containing a progressively reduced urea concentration (6, 4, 2, and 1 M urea). The last step (buffer with 1 M urea) was performed overnight at 4°C. At the end, the proteins were dialyzed against tetramer buffer (5 mM Tris-HCl, pH 8.4) for 1 h at room temperature. After dialysis, VIM and VIMΔ30 concentrations were determined by measuring the absorbance at 280 nm and using an $\epsilon = 24\,900\text{ cm}^{-1}\text{M}^{-1}$ and an $\epsilon = 21\,890\text{ cm}^{-1}\text{M}^{-1}$, respectively.

Oligonucleotides

Oligonucleotides were purchased from Eurogentec (Liège, Belgium) as RP-HPLC-purified products. Oligonucleotides were dissolved in milliQ H₂O to prepare 100 μM or 1 mM stock solutions. Before use, each sample was heated at 95°C for 7 min in the required buffer and then slowly cooled down at room temperature to equilibrate the system. For double-strand DNA (dsSCR), an equimolar solution of Scr22_F and Coscr22_D was annealed as reported earlier. The DNA sequences used in this work are listed in Table 1.

Peptides synthesis and characterization

Two peptides from the N-terminal domain of VIM (peptide 11–30 and peptide 68–79, Table 2) were synthesized by manual solid-phase peptide synthesis (SPPS) using Fluorenylmethoxycarbonyl (Fmoc) chemistry and Tert-butyl (tBu) pro-

tection strategy on Rink-amide MBHA resin in a 0.1625 mmol scale.

For peptide 68–79, OXYME PURE/N,N'-dicyclohexylcarbodiimide (DIC) activation was performed using a three-fold molar excess (0.4 mmol) of Fmoc-amino acid in DMF for each coupling cycle. The coupling time was 50 min. Fmoc deprotection was performed with 20% v/v piperidine in DMF. The cleavage from the resin was performed by treatment with a mixture containing trifluoroacetic acid (TFA):triisopropylsilane:H₂O (95:2.5:2.5 v/v), followed by precipitation with Et₂O. The crude peptide was purified by preparative RP-HPLC using a C18 column. The column was perfused at a flow rate of 12 ml/min with a mobile phase containing solvent A (0.05% TFA in water) and a 5%–30% linear gradient of solvent B (0.05% TFA in acetonitrile:water, 9:1 v/v) in 30 min. The fractions containing the desired products were collected and lyophilized to constant weight.

The synthesis of the peptide 11–30 presented significant challenges, primarily due to the formation of diketopiperazine (DKP), a cyclic byproduct that can frequently form during peptide chain elongation and that leads to truncated peptide chains [45]. To minimize byproduct formation, for amino acids Tyr11, Arg12, Arg13, Pro18, Gly19, Thr20, Ala21, Ser22, Arg23, and Tyr30, the coupling procedure was repeated twice. Moreover, Fmoc deprotection was performed with 20% v/v piperidine in DMF up to Gly19 and then with 20% v/v piperidine in DMF in the presence of OXYME PURE 1 M. Finally, during peptide purification, the linear gradient of solvent B was adjusted from 10% to 35% in 25 min (Supplementary Table S1). The purified peptides were characterized by LC-MS on Agilent Infinity II system (Supplementary Table S1). Their folding was monitored by circular dichroism (CD) in the far-UV range (190–260 nm) using Jasco (Tokyo, Japan) J-1500 spectropolarimeter. Samples were prepared at 0.5 mM in 10 mM Tris, pH 8.4, with increasing concentrations of 2,2,2-trifluoroethanol (TFE). CD spectra were acquired on a fused quartz cell of 0.02 cm pathlength (Hellma, Mühlheim, Germany).

Dynamic light scattering

Dynamic light scattering (DLS) data were acquired on a Zetasizer Nano ZS (Malvern Panalytical), using a 1 cm path-length cuvette for small volumes. Protein solutions were preliminarily filtered with 0.22 μm PVDF sterile syringe filters (Merck KGaA). Solutions of 8 μM VIM or VIMΔ30 were prepared under various buffer conditions: 5 mM Tris at pH 7.5 or pH 8.4, with 0, 50, or 150 mM KCl in the presence/absence of 4 μM 2TEL. After 30 min incubation at room temperature. DLS data were acquired and analyzed using the Zetasizer software. From the point of maximum intensity of the DLS curve, we derived the hydrodynamic radius (R_h) that represents the radius of a sphere that diffuses at the same rate as the tested analyte.

Transmission electron microscopy

Transmission electron microscopy (TEM) images were acquired for VIM solutions at concentrations ranging from 0.15 to 0.3 mg/ml in 5 mM Tris, pH 8.4, in the presence or absence of 50 mM KCl. For samples in the presence of DNA, VIM was added to the oligonucleotide (two-fold excess of 2TEL versus VIM tetramer) in the KCl-containing reaction buffer, and incubation was performed for 30 min at room temperature. Samples were loaded onto a glow-discharged

Table 1. DNA sequences herein studied

Name	Sequence
2TEL	TTAGGG TTAGGG TTAGGG TTAGGG TTAGGG TTAGGG TTAGGG TTAGGG
2TEL_BIOT	Biotin-TTAGGG TTAGGG TTAGGG TTAGGG TTAGGG TTAGGG TTAGGG TTAGGG TTAGGG
2TEL_F	6FAM-TTAGGG TTAGGG TTAGGG TTAGGG TTAGGG TTAGGG TTAGGG TTAGGG TTAGGG
TEL	AGGGTT AGGGTT AGGGTT AGGGT
TEL_FD	Dabcyl-AGGGTT AGGGTT AGGGTT AGGGT - 6FAM
TEL_BIOT	Biotin-AGGGTT AGGGTT AGGGTT AGGG
Scr22	GGATGT GAGTGT GAGTGT GAGG
Scr22_F	GGATGT GAGTGT GAGTGT GAGG - 6FAM
Coscr22_D	Dabcyl-CCTCAC ACTCAC ACTCAC ATCC
Scr44	GGATGT GAGTGT GAGTGT GAGGGG ATGTGA GTGTGA GTGTGAG G
Scr44_BIOT	Biotin-GGATGT GAGTGT GAGTGT GAGGGG ATGTGA GTGTGA GTGTGAG G

Sequences are reported from 5' to 3'.

Table 2. Peptide sequences herein studied

Name	Sequence
11–30	H-Y ¹¹ RRMFGGPGTASRPSSRSY ³⁰ -NH ₂
68–79	H-V ⁶⁸ RLRSSVPGVRL ⁷⁹ -NH ₂

Sequences are reported from the N- to the C-terminal.

carbon-coated copper grid for 30 s, washed with 2% methylamine tungstate solution (NANO-WTM, nanoprobes) at pH 6.8, and finally stained for 40 s with 2% methylamine tungstate solution (NANO-WTM, nanoprobes) at pH 6.8. Electron micrographs were recorded on a TalosTM F200X G2 TEM (Thermo ScientificTM) operated at 100–200 kV.

Fluorescence anisotropy

Fluorescence anisotropy measurements were performed on a Spark[®] plate reader (Tecan Trading AG, Switzerland). Measurements were performed using an Optiplat 96-well black (PerkinElmer) with the following parameters: 485 nm excitation wavelength, 535 nm emission wavelength, bandwidth of 20 nm, and integration time of 40 μ s. The instrument G factor was determined prior to anisotropy measurements. Labeled oligonucleotide 2TEL_F was previously annealed and then used at a concentration of 10 nM in 5 mM Tris-HCl (pH 8.4), 50 mM KCl. Analysis was performed with increasing concentrations of vimentin (ranging from 0 to 150 nM). After mixing, the solution was equilibrated for 5 min before data acquisition. Experiments were performed in triplicate. Acquired data were fitted according to a one-binding site model to derive the dissociation constants of VIM for G4_{rp} folded oligonucleotide 2TEL_F.

Hydrogen–deuterium exchange mass spectrometry

For HDX, a stock solution of 18 μ M VIM in 5 mM Tris, pH 8.4, in the presence or absence of 50 mM KCl and with or without 9 μ M DNA (molar ratio VIM tetramer:oligonucleotide = 1:2) were incubated for 30 min at room temperature. After incubation, samples were cooled at 1°C (ice/water mixture) or 20°C (Thermocontrol Biosan CH-100) for 5 min to reach the selected temperature for the exchange reaction. Then, for exchange kinetics, stock solutions were diluted with 99% D₂O buffer at the same composition to a final D₂O content of 90%. The exchange reaction proceeded for 10 s, 1 min, 10 min, or 100 min at 1°C or 20°C. At the end of each incubation, samples were di-

luted 1:1 with ice-cold quenching buffer (3.8 M guanidine-HCl, 0.3% formic acid, pH 1.9) to pH 2.3 and flash-frozen in liquid nitrogen and kept at –80°C until use. Deuterium labeling was done in biological and technical replicates to allow the acquisition of a triplicate for each VIM condition ([Supplementary Excel File 1](#)).

Maximally deuterated controls were prepared as previously described, with slight modifications [46]. Briefly, stock solutions were lyophilized, then they were resuspended in 7 M guanidine-HCl, 50 mM DTT, and heated at 70°C for 5 min. After cooling at room temperature, they were diluted with the required D₂O buffer to a final 90% D₂O concentration and incubated at 40°C for 10 min. They were then quenched and flash-frozen into liquid nitrogen ([Supplementary Excel File 2](#)). The percentage of deuterium uptake (%D) at different incubation times was calculated as: %D = (m_t – m₀)/(m₁₀₀ – m₀), where m_t is the mass of the peptide at incubation time t, while m₀ and m₁₀₀ correspond to the experimentally determined mass of the non-deuterated and maximally deuterated peptide, respectively.

Labeled samples were thawed immediately before injection into the WatersTM HDX Manager. First, samples were loaded onto a column with immobilized pepsin (WatersTM, Enzymate BEH Protein Pepsin Column) at 15°C and eluted with 0.2% formic acid in water as a mobile phase at a flow rate of 100 μ l/min for 1 min followed by 2 min at 200 μ l/min to improve DNA removal during trapping [47]. Peptides generated by the pepsin cleavage were captured on a C4 trap column (WatersTM Acquity UPLC Protein BEH C4) and resolved with an Acquity UPLC HSS T3 column (1.8 μ m, 1.0 \times 50 mm, WatersTM) eluted with mobile phases A (0.1% formic acid in water) and B (0.1% formic acid in acetonitrile), applying a linear gradient from 5% to 35% of B over 8 min at a flow rate of 80 μ l/min. All fluidics, valves, and columns were kept at 1°C, except for the pepsin column. To minimize peptide carryover, after each analysis the pepsin column was washed with a solution of 1.5 M guanidine-HCl, 0.8% formic acid, and 4% acetonitrile and a run with a sawtooth gradient was carried out after each sample.

Eluted peptides were analyzed with an Xevo G2-S QToF Mass Spectrometer (Waters, Milford, MA, USA). All samples were acquired in resolution mode with the instrument scanned in the range 280–2000 *m/z*. Positive ion electrospray ionization was used with a capillary voltage at 1.5 kV, sampling cone at 30 V, source temperature of 80°C, and desolvation temperature of 175°C. A lock-mass correction was performed using the leucine enkephalin peptide. For peptide identification, non-deuterated VIM samples were prepared and analyzed accordingly to the HDX-MS experimental setting used for the deuterated ones. Only for these control samples, data-independent acquisition in MS^E mode was run by applying collision energy ramping from 22 to 35 kV, while data acquisition of deuterated samples was performed in the MS-only mode.

HDX-MS data analysis

Peptides were identified using PLGS (Waters Protein Lynx Global Server 3.0.3) on replicates of undeuterated control samples. MS data of deuterated samples were analyzed using DynamX 3.0 (Waters). Labeled peptides were filtered with a minimum of 0.3 products per amino acid, a minimum of 1 consecutive products, and a maximum mass error for the precursor ion of 5 ppm, detected in at least half of the controls. HDX data were analyzed automatically by Dynamix followed by manual refinement of the data. HDX data for each single peptide in the different conditions are reported in the supplementary data (Supplementary Excel Files 3 and 4). TriPLICATE experiments were analyzed for each time point. Heatmaps were made using HD-eXplosion, with a defined significance threshold of a difference in deuterium uptake ≥ 0.4 Da and a *P*-value ≤ 0.01 in Welch's *t*-test ($n = 3$) [48, 49]. Significance thresholds were calculated for each comparison of protein states according to the method of Hageman and Weis [48]. Since computed significance thresholds ranged between 0.32 and 0.39 Da, a 0.4 Da threshold was used in all heatmaps.

For those single peptide (or a few peptides) showing a variation of deuterium uptake between two states at a single incubation time and with a difference lower than 0.5 Da, an additional *t*-test was performed with DECA and only those with *P*-value ≤ 0.01 were considered (Supplementary Excel File 5 and PDF File 6) [50].

Circular dichroism

CD spectra were acquired on a Jasco J-815 spectropolarimeter equipped with a Peltier temperature controller, using a 1 cm path-length quartz cuvette.

For differential CD spectra, a cuvette divided into two compartments of 0.5 cm pathlength was used, one loaded with DNA in 5 mM Tris, 50 mM KCl, pH 8.4, and the other with VIM in 5 mM Tris, pH 8.4. For spectra acquisition in the far UV (200 to 280 nm) DNA was used at 0.25 μ M and protein at 0.5 μ M, while for measurements in the near UV (230–330 nm) 4 μ M DNA and 8 μ M protein were used. Under both conditions, the VIM tetramer:DNA molar ratio was 1:2. Spectra were acquired before and after mixing the contents of the two compartments with the following parameters: scanning speed 100 nm/min; bandwidth 2 nm; data interval 0.5 nm; response 2 s; and temperature 25°C. Spectra were subtracted from the baseline and reported as CD signal (millideg).

CD melting analyses were performed in 10 mM Li₃(PO)₄, 50 mM KCl, pH 7.5, with 2 μ M TEL, either in the absence

or presence of 80 μ M peptide 11–30 or 68–79. Before data acquisition, samples were subjected to a heating–cooling cycle in the 30°C–90°C range at a rate of 5°C/min. CD spectra were acquired in the 30°C–90°C range at a heating rate of 1°C/min and melting temperatures (*T*_m) were determined from the first derivative of the readings at a fixed wavelength by using GraphPad Prism (GraphPad Software, San Diego, CA, USA).

Fluorescence melting assay

The thermal stability of fluorescently labeled DNA was determined in a Roche LightCycler, using an excitation source at 465 nm and recording the fluorescence emission at 510 nm. As intramolecular G4 forming model, we used TEL_FD; for the double-stranded model DNA (dsDNA), the two complementary strands Scr22_F and Coscr22_D, which were previously annealed at an equimolar concentration. Both models were designed to locate a fluorophore (6-FAM) and a quencher (Dabcyl) nearby when folded into the expected secondary structure, thus resulting in a quenching of the FAM fluorescence signal. Reaction mixtures (20 μ l final volume) were assembled on 96-well plates. They contained 0.5 μ M DNA in 10 mM Li₃(PO)₄, 50 mM KCl, pH 7.5, and increasing concentrations of the tested peptides (0–80 μ M). For competition assays, 0.5 μ M TEL_FD with 0, 5, or 40 μ M peptide 11–30 was mixed with increasing concentrations of Scr22 in the above buffer. TEL_FD samples were preliminarily heated and cooled in the 25°C–95°C range at a 0.1°C/s rate.

Data were derived from heating and annealing steps run at a 1°C/min rate. *T*_m values were determined from the first derivatives of the slower melting profiles using the Roche LightCycler software. Each curve was repeated three times. For each oligonucleotide, ΔT_m was calculated as the variation of the *T*_m induced by the presence of the peptide.

Surface plasmon resonance

Surface plasmon resonance (SPR) measurements were performed on a Biacore X100. Before use, all buffers and solutions were previously filtered on a 0.22 μ m (MCE Membrane Millipore™). A streptavidin-coated sensor chip was first conditioned by flushing for 1 min with a solution of 1 M NaCl and 50 mM NaOH in 50% isopropanol, and finally extensively washed with running buffer (5 mM Tris, 50 mM KCl, pH 8.4 with 0.025% Tween 20).

For peptide–DNA binding analyses, previously annealed 5'-biotinylated oligonucleotides were immobilized on one cell of the chip surface by flowing a 50 nM DNA solution at a 10 μ l/min flow rate until the required response units (RU) were reached (RU 200 for 2TEL_BIOT and Scr44_BIOT; RU 400 for TEL_BIOT). A second cell was left blank as control. Sensorgrams were acquired using serial dilutions of peptide (0–40 μ M), which were injected at a 10 μ l/min flow rate for 100 s. After each run, a 30 s regeneration step was performed with 10 mM glycine, pH 2.5, followed by a 100 s stabilization period with running buffer (5 mM Tris, 50 mM KCl, pH 8.4 with 0.025% Tween 20).

To analyze the DNA binding of tested proteins, the chip was functionalized with 2TEL_BIOT up to RU 200 for kinetic analyses or up to RU 1000 for steady-state analyses. For single-cycle kinetic measurements, proteins were tested in the concentration range 0–400 nM by injecting samples for 60 s at a flow rate of 30 μ l/min. Dissociation was performed with

40 s injection of 0.05% SDS, with a subsequent stabilization period of 60 s with running buffer. For steady-state analyses, serial dilutions of each protein (0–400 nM) were injected at a 10 μ l/min flow rate for 400 or 540 s. After each run, a 40 s regeneration step was performed with 0.05% SDS, followed by a 60 s stabilization period with running buffer. Running buffer was used both at 50 and 150 mM KCl.

In the steady-state analyses, the RUs detected at the equilibrium were converted into relative RU (rRU) based on the maximal value predicted by the experimental RU variation and plotted as a function of peptide or protein concentrations. These data were fitted to derive peptide concentration required to produce 50% of the rRU variation (EC_{50}) or to calculate the protein dissociation constant (K_D). The K_D values for the $G4_{rp}$ -protein tetramer complexes were derived by fitting the data according to a 1:1 binding model. Binding curves were obtained from at least three independent experiments.

For single-cycle kinetic measurement, BiaEvaluation Software was used for data analysis. Curves were acquired in triplicate and fitted according to a 1:1 binding model, thus deriving K_a , K_d , and K_D values.

Results

Optimized experimental conditions reduce vimentin heterogeneity in solution

The interaction of VIM with $G4_{rp}$ occurs when the protein is in its tetrameric form [34]. However, in solution, VIM can exist at variable polymerization degrees, ranging from soluble tetramers up to mature filaments, a condition that challenges structural studies. Therefore, the first aim was to establish the appropriate experimental conditions to minimize filament formation and reduce sample heterogeneity.

In vitro, VIM is known to be stable as a tetramer at high pH and low ionic strength, whereas, at pH 7.5 and in the presence of monovalent cations, tetramers rapidly assemble laterally into ULF on a timescale ranging from seconds to minutes [44, 51]. Subsequently, ULFs undergo an end-to-end annealing process to form filaments over timeframes ranging from minutes to hours [52]. Based on this knowledge, to control the distribution of VIM species in our samples, we modulated both pH (pH 8.4 or 7.5) and ionic strength by using 0–150 mM KCl since potassium cations are the most abundant at the intracellular level and the most efficient in stabilizing $G4$ structures [32, 53].

The degree of VIM polymerization was initially evaluated by DLS, from which we derived the hydrodynamic radius (R_h) of the particles under investigation. For VIM in 5 mM Tris at pH 7.5 and 8.4, this value ranged between 15 and 18 nm, consistent with the previously reported hydrodynamic radius of the tetramer (around 15 nm) (Fig. 1A and Supplementary Fig. S1) [52]. As expected, particle size increased with the cation concentration, with a more pronounced increment at pH 7.5 (Fig. 1A and Supplementary Fig. S1) [51]. Therefore, we chose to work at pH 8.4 to minimize VIM polymerization.

In defining the optimal ionic composition of the reaction mixture, avoiding monovalent cations would, in principle, be preferable to prevent protein polymerization. However, this was not feasible in our study since VIM selectively binds $G4_{rp}$, whose formation depends on the coordination of monovalent cations. Thus, we assessed the behavior of the protein in a buffer at low ionic strength (5 mM Tris, 50 mM KCl, pH

8.4). Under this condition, TEM images showed short, not yet fully formed filaments (Fig. 1B), which supported a reduced VIM polymerization, in agreement with previously reported SAXS studies [54]. Moreover, CD and fluorescence anisotropy data acquired in this buffer confirmed that our selected $G4_{rp}$ model sequence [d(TTAGGG)₈, 2TEL] was properly folded into a $G4_{rp}$ and bound VIM with a $K_D = 12.0 \pm 2.3$ nM (Supplementary Fig. S2). Noteworthy, TEM images of VIM-2TEL samples (hereafter referred to as VIM- $G4_{rp}$) showed dispersed particles similar in size to the protein tetramers observed in the absence of monovalent cations (Fig. 1B). This result was further confirmed by DLS (Supplementary Fig. S1), and it is fully compatible with the previously reported competition between the protein binding to $G4_{rp}$ and the filament assembly [34]. Consequently, all the experiments in the presence of salt were performed in 50 mM KCl.

HDX-MS maps the tripartite structure of vimentin arranged into tetramers

To map the VIM domains involved in $G4_{rp}$ binding and evaluate whether complex formation induces specific protein structural rearrangements, we conducted HDX-MS experiments to monitor hydrogen-to-deuterium exchange, which depends on protein solvent accessibility and hydrogen bonding network (Supplementary Fig. S3) [55].

For VIM in 5 mM Tris, pH 8.4, with exchange reaction performed at 1°C, we identified 125 peptides (Supplementary Fig. S4A), including fragments from the coil-2B domain, previously reported as resistant to pepsin proteolysis [43]. This result enabled the coverage of the vast majority of VIM (91.6%). Data analysis showed several peptides with a high percentage of deuterium incorporation, which indicates the presence of unfolded domains (Fig. 2A). Regions with deuterium uptake higher than 80% were located at sequences 2–107, 408–466, and 246–289, corresponding to the intrinsically disordered N-terminal and C-terminal domains, and to the linker between coil-1 and coil-2, respectively. This picture aligns with the VIM tripartite structure. Worth noting, within the N-terminal domain, the region 87–96 was slightly protected at the shortest exchange time (Fig. 2), consistent with the predicted presence of a short alpha helix, known as pre-coil domain (pcd), covering the last 20 amino acids of the N-terminal region [56].

When the exchange reaction was performed at 20°C, the results were comparable to those acquired at 1°C (90 shared peptides, 90.5% protein coverage, Supplementary Fig. S5A). However, as expected, the higher temperature increased the deuterium exchange rate in the coiled domains of the protein (Supplementary Fig. S6A). Since a fast deuterium exchange might hide some structural changes associated with VIM-DNA complex formation, most of the following experiments have been carried out at 1°C. Still, some analyses were performed at both temperatures (1°C and 20°C) to rule out temperature-induced conformational rearrangements.

The initial steps of vimentin polymerization induced by monovalent cations are associated with the stabilization of coil-1A and coil-2A

Having validated the experimental approach for mapping the entire protein, we then compared the structural behavior of VIM under a range of experimental conditions, i.e. the presence of cations and DNA. By HDX-MS, we expected to observe altered HDX patterns at those protein domains directly

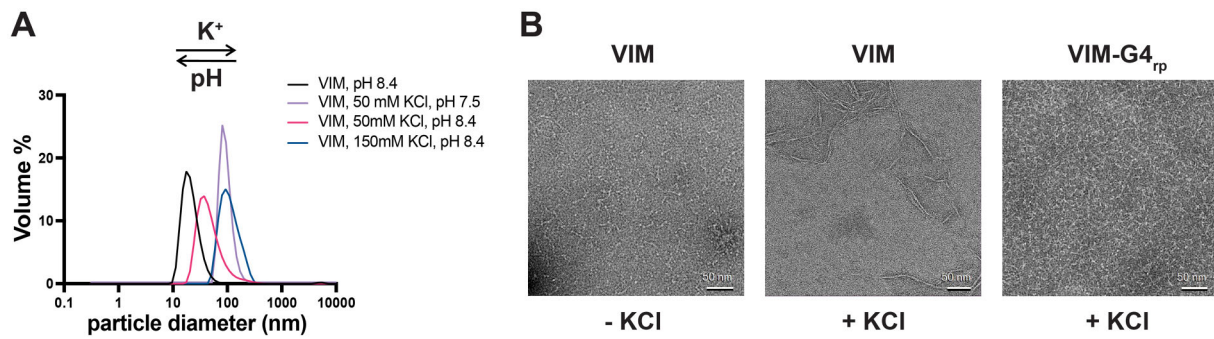


Figure 1. Effect of pH, salt, and DNA on vimentin polymerization. **(A)** DLS of 8 μ M VIM in 5 mM Tris, pH 8.4 (black line) with the addition of 50 or 150 mM KCl (red and blue lines, respectively) and in 5 mM Tris, pH 7.5 with 50 mM KCl (violet line) after 30 min of incubation with the cation. **(B)** TEM images of 8 μ M VIM in 5 mM Tris (left panel), pH 8.4; after 30 min of incubation with 50 mM KCl (central panel) or with 4 μ M 2TEL (VIM-G4_{rp}) and 50 mM KCl (right panel).

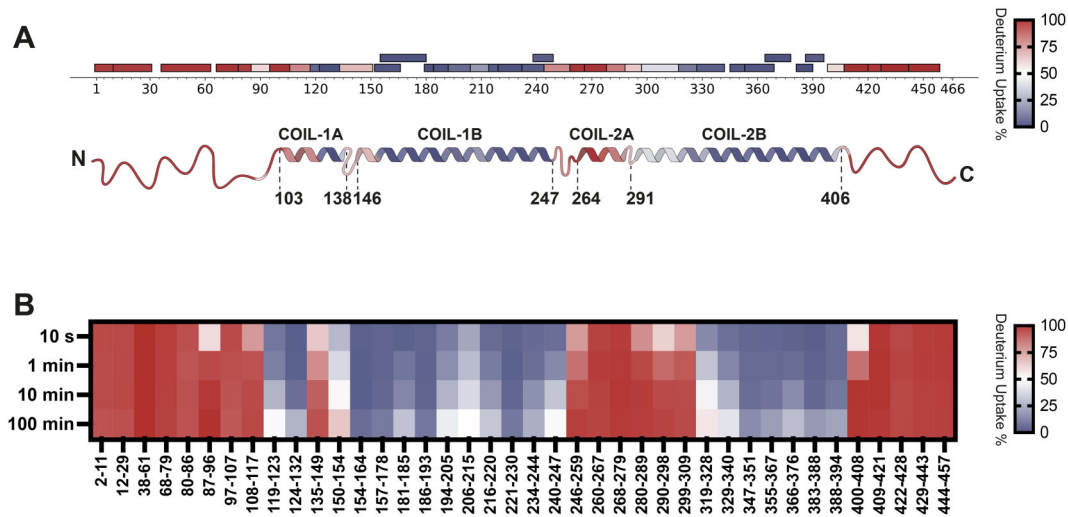


Figure 2. Mapping of vimentin tripartite structure by HDX-MS. Heatmaps showing the percentage of deuterium uptake by VIM peptides after exchange reactions performed in 5 mM Tris, pH 8.4, at 1°C **(A)** for 10 s (top panel) aligned to the schematic tripartite structure of VIM monomer (bottom panel) and **(B)** as a function of the exchange reaction time. Low to high deuterium uptake % is shown from blue to red.

involved in the molecular interactions (e.g. protein–protein interactions during filament elongation or protein–DNA complex formation) or included in concomitant VIM conformational changes.

At first, we compared the deuterium uptake of VIM in 5 mM Tris, pH 8.4, in the presence and absence of 50 mM KCl. Under both conditions, the protein coverage was good. In addition, 88 peptides were shared by the two datasets, providing a protein coverage of 87% (Supplementary Figs S4A and B and S5B). The resulting differential heatmap indicated that, upon salt addition, some regions were more exposed to the solvent (shown in red), while others were protected (shown in blue) (Fig. 3A). In detail, areas showing the most significant increment in deuterium uptake upon the addition of 50 mM KCl were localized in the pre-coil domain (aa 86–96) and in the region of coil-2B (aa 329–394). Additional exposed regions were found within coil-1A (aa 124–132) and coil-1B (aa 194–206). Conversely, protected regions were found within coil-1A (aa 108–117) and coil-2A (aa 268–289). Previously reported HDX data on protein mutants trapped at single protein assemblies indicated that the transition from VIM tetramers to ULFs was associated with the stabilization of regions of coil-1A and coil-2A and the exposure of the pre-coil domain

[43]. Consequently, the herein-identified pattern is compatible with an enrichment in ULF in 50 mM KCl.

Noteworthy, the observed differential exchanges did not share the same kinetics at all sites. This was nicely highlighted by monitoring the exchange reactions at time points ranging from 10 s to 100 min. While some exchange variations were detected only at the shortest tested times (i.e. aa 86–96, 268–289), at the longest tested times only regions with incremented deuterium uptake emerged. This picture mirrors the above-described distribution between flexible and stably folded domains in VIM.

G4 repeats induce distinctive vimentin structural changes upon binding

HDX analyses of the VIM-G4_{rp} complex were performed on samples containing a two-fold excess of 2TEL with reference to VIM tetramers. Based on their concentrations and on the complex K_D determined in 50 mM KCl buffer, this condition corresponds to \sim 97% of bound protein, thus allowing us to properly attribute the corresponding HDX profile to the VIM-G4_{rp} complex. By comparing it to the deuterium uptake of VIM in 50 mM KCl buffer (Fig. 3B), we identified

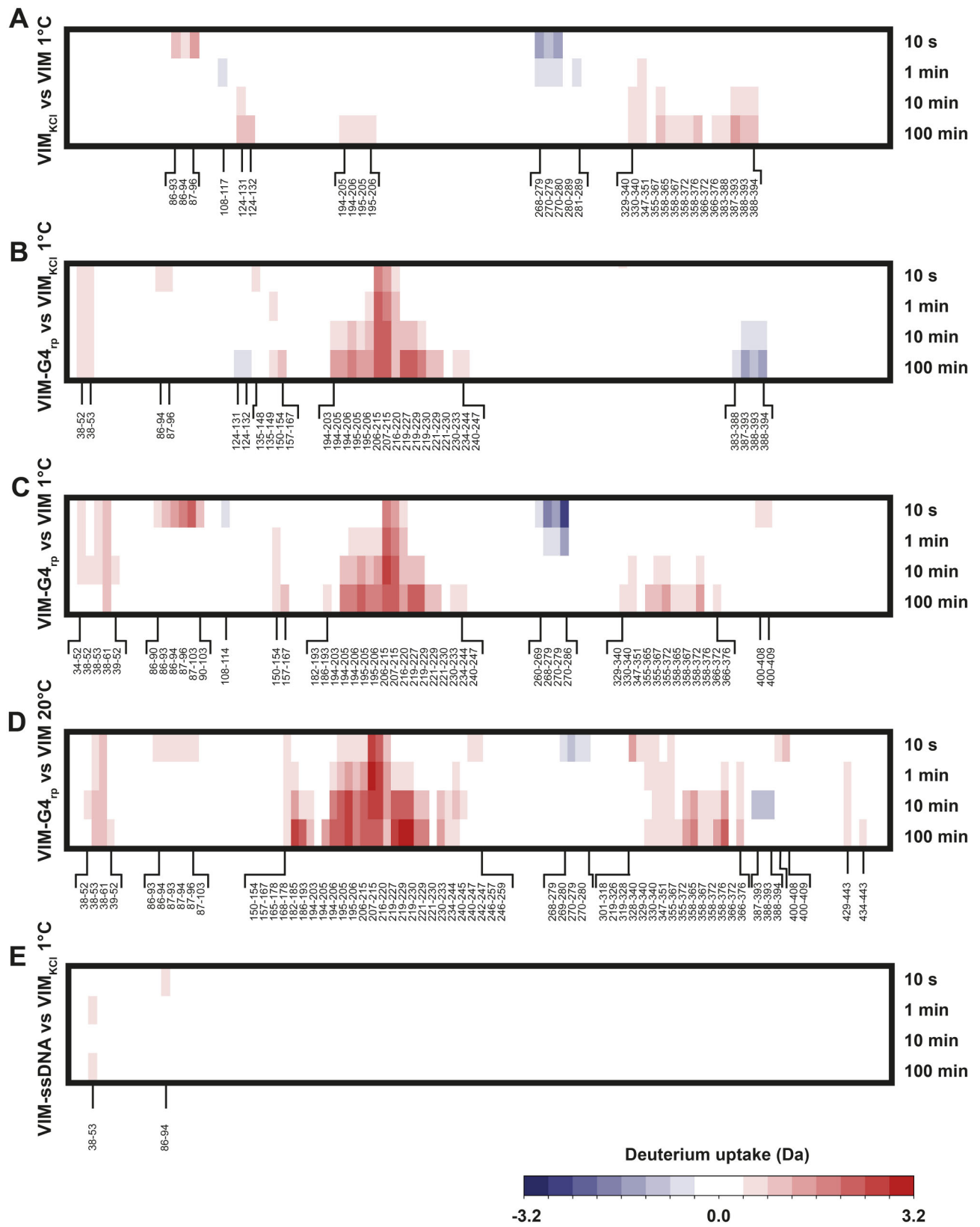


Figure 3. Heatmaps of the differential exposure/protection pattern of vimentin fragments under different experimental conditions. **(A)** VIM in 5 mM Tris, 50 mM KCl, pH 8.4 compared with VIM in the absence of KCl, exchange reaction at 1°C; **(B)** VIM-G4_{rp} (VIM tetramer:2TEL molar ratio 1:2) in 5 mM Tris, 50 mM KCl, pH 8.4 compared with VIM in the same buffer, exchange reaction at 1°C; **(C)** VIM-G4_{rp} (VIM tetramer:2TEL molar ratio 1:2) in 5 mM Tris, 50 mM KCl, pH 8.4 compared with VIM in the absence of KCl, exchange reaction at 1°C; **(D)** VIM-G4_{rp} (VIM tetramer:2TEL molar ratio 1:2) in 5 mM Tris, 50 mM KCl, pH 8.4 compared with VIM in the absence of KCl, exchange reaction at 20°C; **(E)** VIM-ssDNA (VIM tetramer:Scr44 molar ratio 1:2) in 5 mM Tris, 50 mM KCl, pH 8.4 compared with VIM in the same buffer, exchange reaction at 1°C. Red and blue refer to exposure and protection, respectively.

94 shared peptides that provided a protein coverage of 84% (Supplementary Figs S4B and C and S5C). Nevertheless, it is worth reminding that, while in 50 mM KCl VIM is prevalently arranged into ULFs (with eventually a small fraction of filaments), the addition of 2TEL shifts the protein equilibrium toward its tetrameric form. Thus, to properly consider this concomitant modulation of VIM assembly, we also compared the HDX profile of the VIM-G_{4rp} complex to the one of VIM in the absence of KCl, where the protein retains its tetrameric arrangement (Fig. 3C). Here, the protein coverage was 85%, with 97 shared peptides (Supplementary Figs S4A and C and S5D).

The comparison of the two differential heatmaps (VIM-G_{4rp} versus VIM-KCl and VIM-G_{4rp} versus VIM) revealed some shared patterns corresponding to DNA-induced exposure in coil-1B (aa 150–167 and 194–247) and N-terminal domain (aa 38–53) (Fig. 3B and C). These regions can be identified as those affected by the presence of the nucleic acid. Most of the peptides identified in the 150–247 region shared an increased deuterium uptake in the presence of 2TEL. However, they are generally enriched in negatively charged residues, suggesting that they do not interact directly with the nucleic acid while electrostatic repulsions can drive their exposure to the solvent. Among them, the largest difference in deuterium uptake was observed for peptide 206–215 (Fig. 4A–D). Notably, the kinetic profile of this process overlapped in the presence and absence of KCl, while DNA caused the formation of a fast-exchanging fraction. This supports that the VIM-G_{4rp} complex formation induces changes distinct from those associated with protein oligomerization.

Besides the earlier discussed shared pattern, the heatmaps of VIM-G_{4rp} versus VIM-KCl and VIM-G_{4rp} versus VIM (Fig. 3B and C) also highlighted significant variations. In particular, the VIM-G_{4rp} versus VIM panel (Fig. 3C) presented a wider and more pronounced deuterium exchange at the pre-coil domain (aa 86–103), two newly exposed sites within coil-2B (aa 329–376 and 400–409), as well as two protected sites in coil-1A (aa 108–114) and one in coil-2A (aa 260–286). Among them, only those occurring at amino acids 400–409, which are enriched in acidic amino acids, were selectively induced by DNA binding, while the others mirrored the typical pattern observed in VIM upon potassium addition (Fig. 3A and C). This addresses them as VIM domains influenced by KCl and, to a larger extent, by DNA, although not fully overlapping since largest variations were associated with the presence of DNA (Fig. 4E and F).

Finally, the comparison of VIM-G_{4rp} versus VIM-KCl (Fig. 3B) highlighted one site within coil-1A (aa 124–132) and one in coil-2B (aa 383–394) that were selectively protected only in the presence of the nucleic acid. Since they are exposed when VIM tetramers tend to polymerize (Fig. 3A), this can reflect the competition between VIM-G_{4rp} complex formation and higher-order protein assembly.

As reported earlier for VIM, the VIM-G_{4rp} complex data, referring to exchange reactions performed at 1°C, were also compared with those run at 20°C. The new heatmap reporting the effect of the temperature on VIM-G_{4rp} (101 shared peptides, 83.7% protein coverage, Supplementary Fig. S5E) showed an evenly distributed increment of the deuterium uptake at higher temperature (Supplementary Fig. S6B). Notably, this did not occur at the N- and C-terminal domains likely as

a result of their already fast exchange rate at the lowest tested temperature.

Due to the comparable exchange pattern of VIM and VIM-G_{4rp} at the two tested temperatures, the differential exchanges of VIM-G_{4rp} versus VIM tetramer at 1°C and 20°C (107 shared peptides, 86.0% protein coverage, Supplementary Fig. S5F) were comparable (Fig. 3C and D). An exception was represented by the coil-2B domain (aa 387–394), whose protection was detected only at 20°C (Fig. 3D). By following the exchange of peptide aa 388–393 under different conditions (Fig. 4G), we verified very slow exchange rates at 1°C, which prevented us from detecting significant variation in the presence/absence of DNA. Conversely, the implemented exchange rates at 20°C highlighted a protection driven by the G_{4rp}, similar to the one observed from the comparison of VIM-G_{4rp} versus VIM-KCl (Fig. 3B). Worth noting, this region was already identified as a structural anchor in VIM tetramers [43]. Thus, this result indicates that this feature is further amplified by G_{4rp}.

Finally, to validate whether the observed protein changes induced by 2TEL were selectively related to the DNA folding, we analyzed VIM in the presence of a scrambled G-rich sequence (Scr44) that does not fold into G_{4s} (or other secondary structures) in 5 mM Tris, 50 mM KCl, pH 8.4. When we compared the HDX profile of VIM acquired in the presence or absence of ssDNA in KCl buffer (90 shared peptides, 81% protein coverage, Supplementary Figs S4B and D and S5G), no relevant differences were detected in the deuterium uptake between the two samples (Fig. 3E). This result confirms that the presence of ssDNA does not affect the protein polymerization promoted by KCl and further supports that the binding of VIM to DNA relies on specific nucleic acid folding requirements and drives unique protein structural rearrangements.

The interaction between a G-quadruplex repeat and vimentin reduces the protein secondary structure content

HDX data analysis revealed an altered exposure or protection of some VIM domains upon 2TEL binding. As previously discussed, these variations can reflect either direct protein–DNA interactions or long-range conformational changes within the protein. To get insight into the secondary structure rearrangements of VIM and G_{4rp} upon binding, we acquired differential circular dichroism spectra.

Under the experimental conditions applied for HDX-MS, the far-UV spectrum of VIM displayed two minima at 208 and 222 nm, which are indicative of an α -helical structure (Supplementary Fig. S7A) [57]. This observation is consistent with the enrichment of this secondary structure in VIM.

The VIM spectra acquired before and after incubation with G_{4rp} exhibited the same signature, albeit with reduced intensity (Fig. 5A). Since the chiroptical contribution of DNA was negligible in this region (Supplementary Fig. S7B), we were able to quantify a decrease in the VIM α -helical content from 57% to 50% in the presence of 2TEL. This variation can be associated with the exposure of coil-1B detected by HDX-MS.

In the near UV range, CD spectra are a combination of signals arising from the secondary structure of 2TEL and the tertiary structure of VIM. However, under our experimental conditions, the chiroptical contribution of VIM above 250 nm was not significant (Supplementary Fig. S7C). Conversely, the

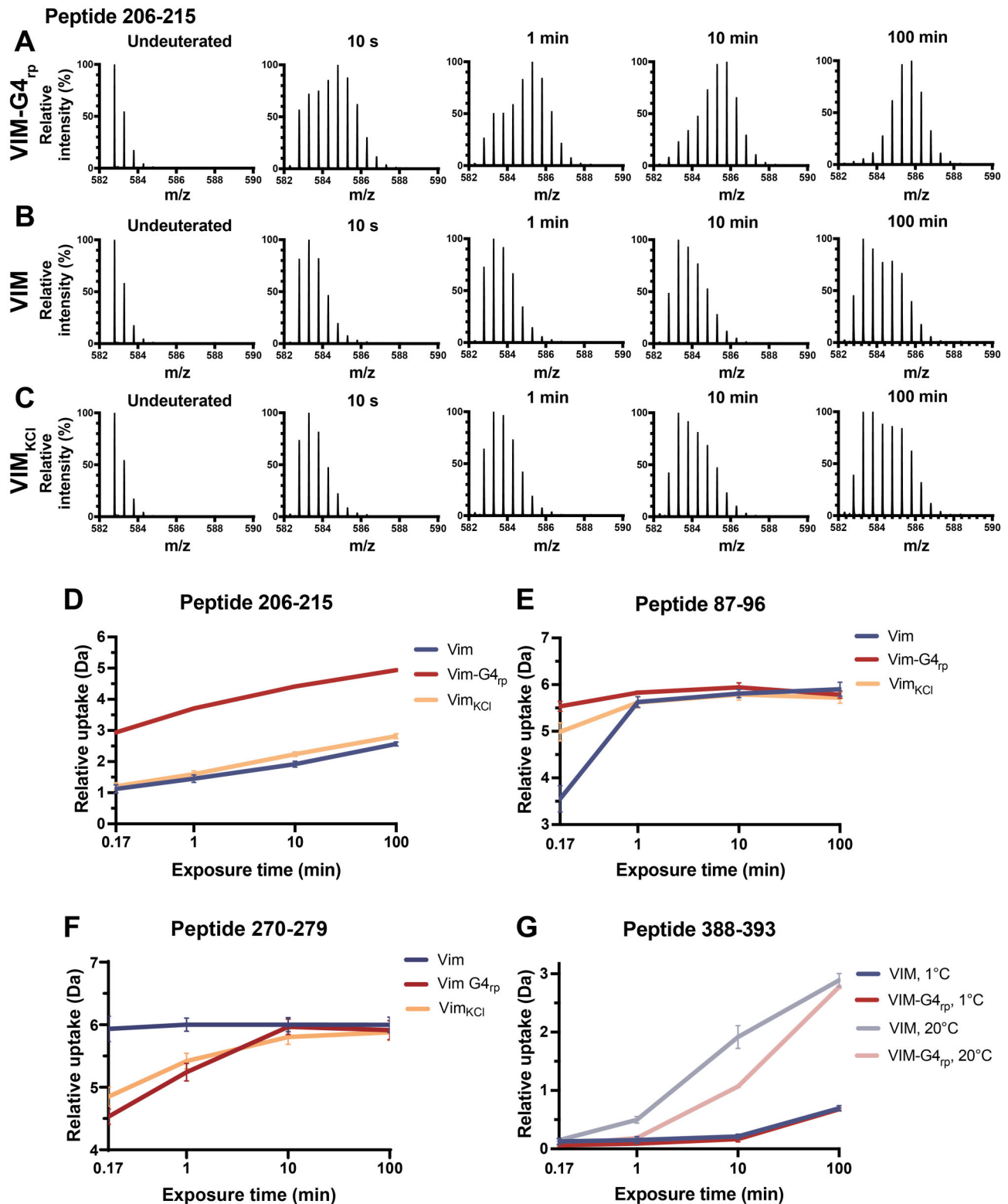


Figure 4. Time-dependent deuterium uptake by selected vimentin peptides under different experimental conditions. MS spectra of peptide 206–215 at different exchange times from (A) VIM-G4_{rp} (VIM tetramer:2TEL molar ratio 1:2) in 5 mM Tris, 50 mM KCl, pH 8.4; (B) VIM in 5 mM Tris, pH 8.4; and (C) VIM_{KCl} in 5 mM Tris, 50 mM KCl, pH 8.4. Relative deuterium uptake as a function of exchange reaction time for peptide 206–215 (D), 87–96 (E), and 270–279 (F) obtained from VIM (red), VIM-G4_{rp} (blue), and VIM_{KCl} (orange); (G) relative deuterium uptake for peptide 388–393 obtained from VIM with exchange reaction performed at 1°C (blue) or 20°C (light blue) and from VIM-G4_{rp} with exchange reaction performed at 1°C (red) or 20°C (pink).

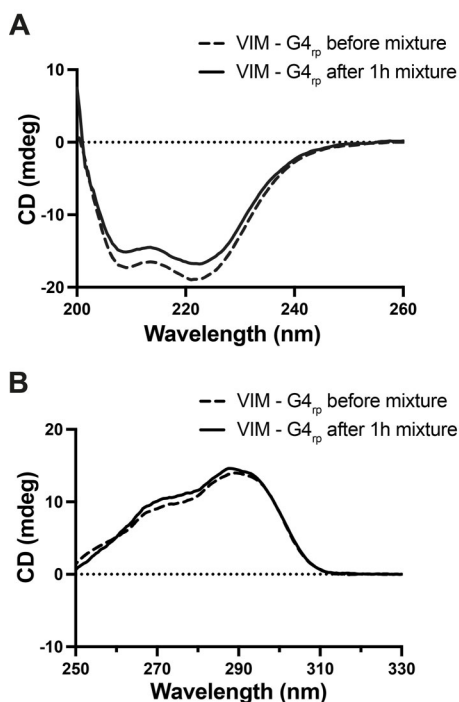


Figure 5. Variation of VIM and 2TEL secondary structure upon complex formation. Differential CD spectra acquired in 5 mM Tris, 50 mM KCl, pH 8.4, 25°C before mixture (dashed line) and 1 h after mixture (solid line) of (A) 0.5 μ M VIM and 0.25 μ M 2TEL (G_{4rp}) and (B) 8 μ M VIM and 4 μ M 2TEL (G_{4rp}).

spectrum of 2TEL presented a profile consistent with its folding into a G_{4rp} (Supplementary Fig. S7D) [36]. Noteworthy, this spectrum did not change in the presence of VIM (Fig. 5B), indicating that 2TEL retained its secondary structure upon protein binding.

The same analysis was performed also with TEL, a related sequence that folds into a single G4 unit and does not bind VIM [34]. In this case, a much more limited variation was detected in the near-UV range, which parallels the one recorded in the absence of nucleic acid (Supplementary Fig. S8). This output further supports a unique interaction pattern of VIM with 2TEL.

The N-terminal domain of vimentin is directly involved in the interaction with G-quadruplex repeats

So far, reported literature data indicate the N-terminal domain of VIM as potentially involved in DNA binding [34, 58, 59]. The presented HDX results did not reveal a clear N-terminal protection pattern associated with the presence of nucleic acid, although an unusual behavior emerged for peptides 11–29 and 12–29. Indeed, in the VIM– G_{4rp} samples, their signals were significantly reduced (Supplementary Fig. S9), thus impairing their proper evaluation across all analyzed datasets.

To address whether this effect was related to DNA binding, we synthesized peptide 11–30 (Table 2) and analyzed its interaction with DNA oligonucleotides folded into different secondary structures. As a control, we also synthesized peptide 68–79 derived from the protein N-terminal domain, whose deuterium uptake was not affected by the presence of DNA. Both peptides were produced by manual solid-phase synthe-

sis to minimize the formation of side products and to obtain them at a high purity degree (Supplementary Table S1). As predicted from their amino acid composition, neither of them folded into any defined secondary structure in 10 mM Tris, pH 8.4 (Supplementary Fig. S10). Only in 100% TFE, a modest α -helical contribution was detected.

The potential interaction of these VIM fragments with DNA was preliminarily evaluated on two simple models: a telomeric sequence that folds into a single G4 (TEL_FD) and a dsDNA. By a fluorescence melting assay, we monitored the variation of DNA T_m induced by increasing concentrations of the tested peptides (Supplementary Fig. S11A and B). While none of them induced a significant thermal stabilization of the dsDNA, peptide 11–30 exhibited a G4-stabilizing effect, which was more pronounced than the control peptide 68–79. The denaturation profile of TEL_FD-peptide 11–30 complex at two different molar ratios was also measured in the presence of increasing concentrations of a single-stranded oligonucleotide (Scr22) used as a potential competitor. The addition of ssDNA did not affect the thermal stability of the G4-peptide complex (Supplementary Fig. S11C), further supporting the preferential interaction of peptide 11–30 with G-quadruplex DNA structures.

We also monitored the telomeric G4 thermal stabilization at a fixed peptide concentration by CD. This analysis confirmed that peptide 11–30 increased the T_m of the G4 folded TEL from $59.3 \pm 0.2^\circ\text{C}$ to $66.1 \pm 0.5^\circ\text{C}$ (Supplementary Fig. S11D). In contrast, peptide 68–79 at the same concentration did not affect the DNA melting profile ($T_m = 59.4 \pm 0.3^\circ\text{C}$). This highlights that the labeling of TEL with fluorophore and quencher groups did not affect the binding of the tested peptides.

Based on this positive output, we focused on peptide 11–30 and quantified its recognition of different DNA secondary structure arrangements by SPR (Supplementary Fig. S12). In agreement with previously described results, the interaction of this peptide with the single-stranded oligonucleotide (Scr44_BIOT) was too weak to be properly quantified (Fig. 6A), whereas, in the presence of a single telomeric G4 (TEL_BIOT), we derived an $EC_{50} = 15.1 \pm 2.3 \mu\text{M}$. Noteworthy, when the target sequence was a G_{4rp} (2TEL_BIOT), the recognition efficiency further increased ($EC_{50} = 8.9 \pm 1.6 \mu\text{M}$). Consistently, the same G_{4rp} sequence was not efficiently bound by the control peptide 68–79 (Fig. 6B).

To determine the contribution of peptide 11–30 to VIM– G_{4rp} interaction, we expressed a protein variant lacking the first 30 amino acids (VIM Δ 30, Supplementary Fig. S13). First, we evaluated the polymerization ability of this truncated protein. DLS analysis showed that in 5 mM Tris at either pH 8.4 or pH 7.5, the Rh size of VIM and VIM Δ 30 were comparable, measuring ~ 18 nm under both conditions (Fig. 7A and Supplementary Fig. S14). Only for VIM Δ 30 was this value essentially retained upon salt addition (21–25 nm at both tested pHs). This highlighted its inability to polymerize in line with the previously reported critical role of the N-terminal in VIM filament formation [60].

Subsequently, we compared the G_{4rp} binding of VIM and VIM Δ 30 by SPR (Fig. 7B and C and Supplementary Fig. S15). In 50 mM KCl, for VIM we derived a $K_D = 5.46 \pm 1.08$ nM, consistent with the 12 nM value previously determined by fluorescence anisotropy. Comparably, VIM Δ 30 exhibited a K_D of 22.51 ± 7.15 nM. However, SPR kinetic analysis revealed faster association–dissociation rates for the

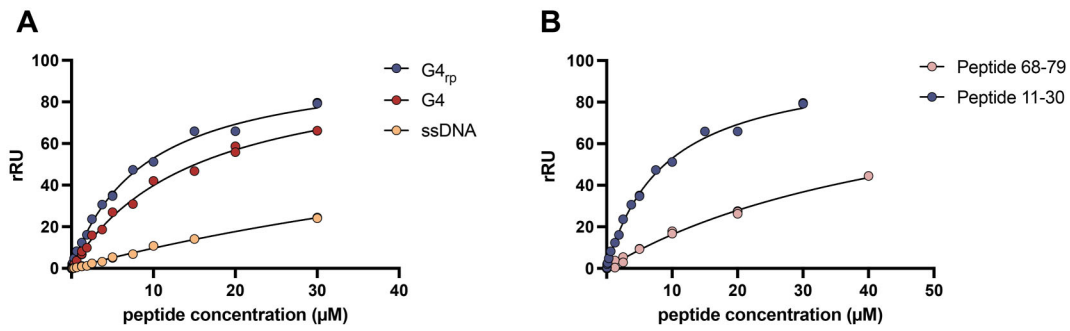


Figure 6. Interaction of vimentin-derived N-terminal peptides with DNA. Plots of the SPR response (rRU) at the steady state in 5 mM Tris, 50 mM KCl, pH 8.4 with 0.025% Tween 20, as a function of: **(A)** peptide 11–30 concentrations on chips functionalized with $G4_{rp}$ (2TEL_BIOT; blue), G4 (TEL_BIOT; red), and ssDNA (Scr44_BIOT; orange); and **(B)** peptide 11–30 (blue) and peptide 68–79 (pink) concentrations on chips functionalized with $G4_{rp}$.

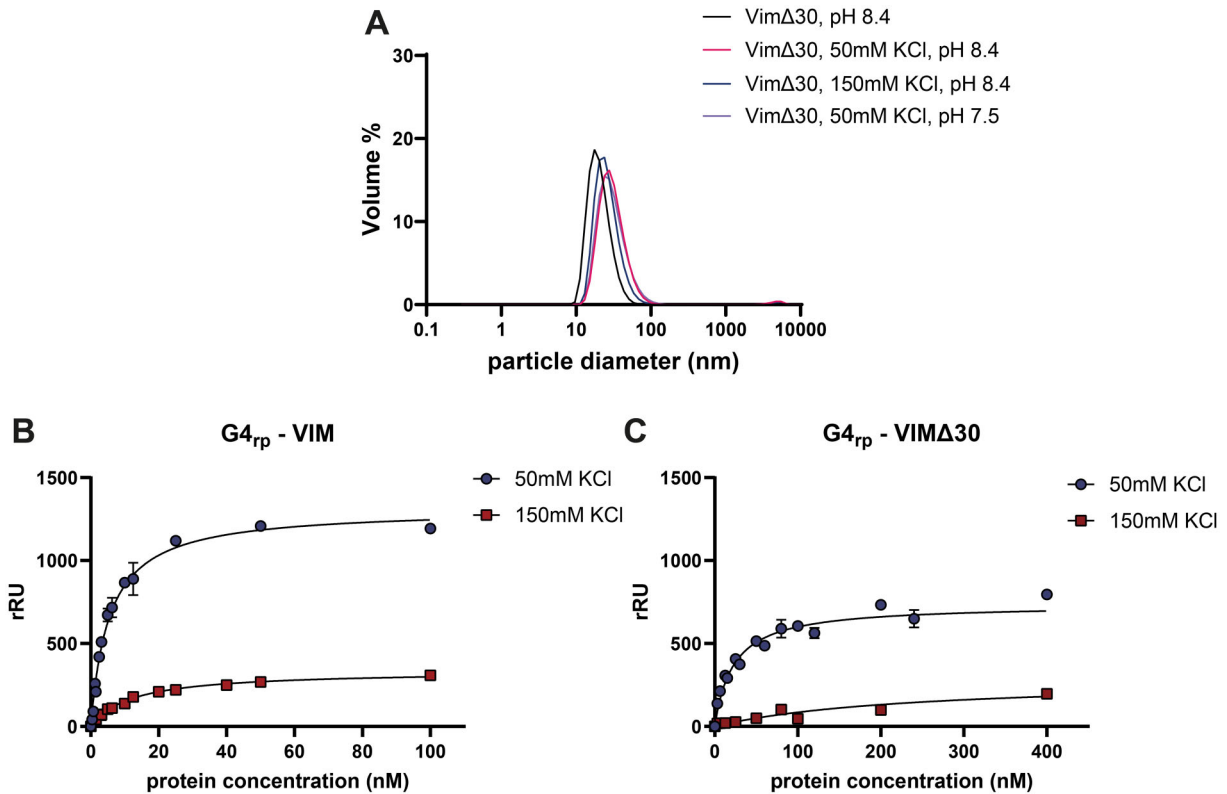


Figure 7. Comparative analysis of $VIM\Delta30$ and full-length vimentin in polymerization and $G4_{rp}$ binding. **(A)** DLS of 8 μ M $VIM\Delta30$ in 5 mM Tris, pH 8.4 (black line), after addition of 50 mM KCl (red line) or 150 mM KCl (blue line), and in 5 mM Tris, pH 7.5 with 50 mM KCl (violet line), measured after 30 min of incubation. Steady-state SPR response (rRU) in 5 mM Tris, pH 8.4, with 0.025% Tween 20, as a function of VIM **(B)** or $VIM\Delta30$ **(C)** concentration, in the presence of 50 mM KCl (blue) or 150 mM KCl (red), on $G4_{rp}$ -functionalized chips.

truncated protein than those of the full-length VIM, which might point to a relevant contribution of charge–charge interactions (Supplementary Fig. S16A and B). To verify this hypothesis, we increased the ionic strength up to 150 mM KCl. Under this condition, the $G4_{rp}$ interaction by VIM was preserved ($K_D = 12.04 \pm 3.04$ nM), while the one exerted by $VIM\Delta30$ was strongly reduced, as further confirmed by kinetic measurements (Supplementary Fig. S16). These results addressed the $VIM\Delta30$ - $G4_{rp}$ binding process as largely electrostatically driven. Accordingly, in the full-length VIM, the 1–30 N-terminal region is therefore expected to provide additional interactions to the DNA binding. This effect can be am-

plified within the VIM tetramer, where up to four N-terminal domains can cooperatively participate in DNA recognition. This would explain why the EC_{50} of peptide 11–30 against $G4_{rp}$ was in the micromolar range, while the binding affinity of the full-length protein was in the nanomolar range. In addition, HDX analysis identified other protected regions (coil-1A aa 108–114, coil-2A 260–286, and coil-2B aa 387–394), which might suggest the potential presence of multiple DNA-binding sites. However, this technique does not allow for discrimination between domains directly involved in DNA binding and regions protected as a result of long-range allosteric rearrangements.

Discussion

In the past, the interest in cytoskeletal proteins has mainly focused on the structural functions associated with their polymerized forms. More recently, these proteins have been proven to play active roles in processes such as chromatin organization and transcriptional regulation. Among them, VIM appears as unique since it selectively recognizes the non-canonical DNA secondary structures known as G₄_{rp}, which are enriched at oncogene promoters and telomeres [28, 34, 35, 36]. Noteworthy, this binding is supported by VIM tetramers and not by its filaments. To dissect the structural basis of this dual behavior, here we applied HDX-MS to map the VIM regions involved in DNA binding and compare them to those involved in protein filament formation.

As far as it concerns the structural rearrangements occurring during the initial steps of VIM polymerization, we identified the protection of regions within coil-1A (aa 108–117) and coil-2A (aa 268–289), along with an increased exposure of the pre-coil domain (aa 86–96). These results align with the previously reported VIM HDX data, although we did not map a protection in the C-terminal region of coil-2B [43]. This difference reasonably reflects a reduced filament formation under our experimental conditions, which, in agreement with SAXS data, favored ULF [54]. Consistently, our experimental setup enabled the detection of the characteristic fingerprint associated with ULF formation by the wild-type protein, which was previously identified only using VIM mutants [43]. Additionally, we also detected fragments of coil-2B, which was identified as resistant to pepsin proteolysis.

The same exposure of the pcd region, as well as the protection of coil-1A and coil-2A, was preserved and even amplified when VIM–G₄_{rp} was compared with the protein tetramer. Moreover, exchange experiments performed at 20°C also highlighted a G₄_{rp}-mediated protection at coil-2B (aa 387–394), a feature associated with the tetrameric form of the protein. Overall, these variations further support the selective interaction of VIM tetramers with G₄_{rp}. A remark is deserved for the N-terminal domain of VIM, previously reported to be involved in DNA recognition, although its intrinsic disordered state challenges for structural information [34, 58, 59]. Nevertheless, HDX-MS analyses identified the 11–30 region as a DNA-binding peptide able to recognize G4 modules with structure-based selectivity.

It is interesting to compare this VIM–G₄_{rp} interaction model to those derived for other characterized G4-binding proteins, i.e. nucleolin, hnRNP A1, POT1, and DHX36 [61]. They typically recognize individual G4 elements through modular domains such as RGG motifs, RRM, or KH domains that are either intrinsically disordered regions enriched in arginine–glycine repeats, or folded elements presenting conserved aromatic residues arranged on β -sheet surfaces [62–65]. These domains establish cation– π , π – π stacking interactions, and hydrogen bonds with guanine tetrads or adjacent loops. While the G₄_{rp} recognition by VIM is based on a similar molecular mechanism involving the positively charged and intrinsically disordered 11–30 region, herein collected data highlighted an overall distinct binding strategy. Indeed, the single peptide exhibits a lower affinity compared with the full-length protein, indicating that additional regions cooperatively interact with the DNA. Furthermore, the binding proceeds through an induced-fit mechanism involving a conformational switch within the VIM α -helical domain

(aa 182–247), located at the central position of the soluble tetramer. We can reasonably associate this DNA-driven structural transition with the reduction of the VIM α -helix content detected by differential CD. Indeed, this protein domain is enriched in acidic amino acids, and, consequently, it likely rearranges to minimize the electrostatic repulsion versus the negatively charged backbone of G₄_{rp}. These findings highlight the capability of HDX-MS to detect weak interactions and multiple conformational rearrangements that occur within dynamic biomolecular complexes.

The interaction of VIM with DNA also revealed a functional dichotomy: upon nucleic acid binding, the protein loses its ability to assemble into filaments, at a variance with its established role in cytoskeletal organization. Data collected here helps to explain the molecular mechanism behind this competition. Indeed, the recent cryo-EM model of VIM filaments revealed an ordered helical framework that generates a central cavity where the protein N-terminal domains are located and interact with each other [38]. As a result, in filaments, the VIM N-terminals are not accessible to DNA. As well, when tetramers are bound to G₄_{rp}, the structural perturbation of the central α -helical domains likely worsens their proper ordered alignment, ultimately impairing filament elongation. By using the truncated VIM Δ 30, we confirmed the direct contribution of the N-terminal region to both VIM polymerization and DNA recognition. This supports the strict interconnection of these two processes, further highlighting the challenge of disentangling their biological contributions in cellular models. Moreover, as suggested by our results, deletion of VIM residues 1–30 *in cell* is not adequate to selectively silence the biological functions associated with the VIM–DNA complexes.

Beyond the functions related to its filaments, at nuclear level VIM serves as a hub for multiple protein interactions, among which those with lamin B and Ku80 are functional in DNA-dependent processes such as chromatin organization and DNA damage signaling [27, 66]. These proteins bind VIM at the C-terminal domain or at coiled-coil regions, respectively [27, 67]. Notably, none of them recognize the DNA-binding interface identified in this work, thus enabling concurrent engagement of VIM with both nucleic acids and protein partners. These findings, merged with the unique selectivity of VIM for G₄_{rp}, support specific functional roles of the VIM–G₄_{rp} complex. In particular, based on the reported function of VIM as a transcription factor at genes associated with cancer cell migration and invasion, and on the already identified enrichment of G₄_{rp}-forming sequences in the promoter of genes involved in EMT, it is feasible to address this complex as a regulatory element to control this process [26, 34, 35]. Moreover, a possible involvement of the VIM–G₄_{rp} interaction in the DNA damage response cannot be excluded. The successful identification of the domains directly involved in these processes paves the way for the rational design of strategies to selectively modulate the formation of the VIM–G₄_{rp} complex and to gain a deeper understanding of its physiological relevance.

In addition, the data presented in this study provide interesting information to improve the individual targeting of both VIM and G4, which are two validated pharmaceutical targets, although the protein is defined as “undruggable” and no G4 selective binders have been approved so far [15, 68]. Looking at VIM-targeting, G₄_{rp} can serve as a promising scaffold to engineer molecules capable of selectively recognizing the

tetrameric VIM. Such an approach would provide a tool to discriminate between distinct structural states of the protein and would complement the already developed aptamers [69]. On the other side, the herein identified peptide 11–30 can be added to the group of those already reported as selective G4 binders [70]. Noteworthy, most of them share anticancer activities, thus suggesting that peptide 11–30 may also exhibit similar properties, making it an interesting candidate for further investigation.

Finally, it is worth mentioning that the cytoplasmic fraction of VIM has been reported to colocalize also with various transcripts, including mRNAs and lncRNAs [71, 72]. Currently, it remains unclear whether these interactions are direct or protein-mediated. Since RNA can also fold into G4, our results offer a compelling molecular rationale for a direct engagement, thus opening an intriguing perspective for future investigation.

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Supplementary data

Supplementary data is available at NAR online.

Conflict of interest

None declared.

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Data availability

The data underlying this article are available in the article and in its online supplementary material.

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