

# 1 **Hydrothermal activity on the CV parent body: new perspectives from the giant**

## 2 **Transantarctic Mountains minimeteorite TAM5.29**

3 Nava J.<sup>1</sup> (*corresponding author*), Suttle M.D.<sup>2,3</sup>, Spiess R.<sup>1</sup>, Folco L.<sup>2</sup>, Najorka J.<sup>3</sup>, Carli C.<sup>4</sup> and Massironi  
4 M<sup>1</sup> [jacopo.nava@phd.unipd.it](mailto:jacopo.nava@phd.unipd.it), [martindavid.suttle@dst.unipi.it](mailto:martindavid.suttle@dst.unipi.it), [richard.spiess@unipd.it](mailto:richard.spiess@unipd.it),  
5 [luigi.folco@unipi.it](mailto:luigi.folco@unipi.it), [j.najorka@nhm.ac.uk](mailto:j.najorka@nhm.ac.uk), [cristian.carli@inaf.it](mailto:cristian.carli@inaf.it), [matteo.massironi@unipd.it](mailto:matteo.massironi@unipd.it).

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7 <sup>1</sup>University of Padova, Department of Geosciences, Via G.Gradenigo 6, 35131 Padova, Italy.

8 <sup>2</sup>Dipartimento di Scienze della Terra, Università di Pisa, 56126 Pisa, Italy.

9 <sup>3</sup>Core research laboratories, Department of Earth Science, The Natural History Museum, Cromwell Rd, London  
10 SW7 5BD, UK.

11 <sup>4</sup>IAPS-INAF, Istituto Nazionale di Astrofisica e Planetologia Spaziali, Roma, Italy

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### 13 **Abstract**

14 TAM5.29 is an extraterrestrial dust grain, collected on the Transantarctic Mountains (TAM). Its  
15 mineralogy is dominated by an Fe-rich matrix composed of platy fayalitic olivines and clasts of  
16 andradite surrounded by diopside-jarosite mantles, chondrules are absent. TAM5.29 records a  
17 complex geological history with evidence of extensive thermal metamorphism in presence of fluids  
18 at  $T < 300^{\circ}\text{C}$ . Alteration was terminated by an impact, resulting in shock melt veins and compaction-  
19 orientated foliation of olivine. A second episode of alteration at lower temperatures ( $< 100^{\circ}\text{C}$ )  
20 occurred post-impact and is either parent body or terrestrial in origin and resulted in the formation  
21 of iddingsite. The lack of chondrules is explained by random sub-sampling of the parent body, with  
22 TAM5.29 representing a matrix-only fragment. On the basis of bulk chemical composition,  
23 mineralogy and geological history TAM5.29 demonstrates affinities to the CV<sub>ox</sub> group with a  
24 mineralogical assemblage in between the Allende-like and Bali-like subgroups (CV<sub>oxA</sub> and TAM5.29  
25 are rich in andradite, magnetite and FeNiS but CV<sub>oxA</sub> lacks hydrated minerals, common in TAM5.29.  
26 Conversely CV<sub>oxB</sub> are rich in hydrated phyllosilicates but contains almost pure fayalite, not found in  
27 TAM5.29). In addition, TAM5.29 has a slightly different metasomatic history, in-between the  
28 oxidised and reduced CV metamorphic grades whilst also recording higher oxidizing conditions as  
29 compared to the known CV chondrites. This study represents the third CV-like cosmic dust particle,  
30 containing a unique composition, mineralogy and fabric, demonstrating variation in the thermal  
31 metamorphic history of the CV parent body(-ies).

### 32 **1. Introduction**

33 The CV (Vigarano-like) carbonaceous chondrites (CCs) are a group of primitive meteorites sharing  
34 approximately equal ratios of chondrules and matrix, as well as the highest abundances of refractory  
35 phases (CAIs [Ca-Al-rich inclusions] and AOAs [ameboid olivine aggregate]) among any chondrite  
36 class (~10 vol%, [McSween, 1977](#); [Brearley and Jones, 1998](#); [Weisberg et al., 2006](#)). Chondrules in CVs  
37 are typically large (averaging 1 mm in diameter) type I, porphyritic olivine subtypes (~90%, [Jones,](#)  
38 [2012](#)) and often surrounded by thick (~400 $\mu\text{m}$ ) accretionary rims ([King and King, 1991](#); [Tomeoka and](#)  
39 [Ohnishi, 2010](#)). Meanwhile, their matrices are rich in fayalitic olivine, Ca,Fe-pyroxenes and andradite  
40 ([Krot et al., 1995-1998](#); [Brearley and Jones, 1998](#); [Weisberg et al., 2006](#)).

41 The CV chondrite group most likely represent members of a single parent body, derived from an  
42 asteroid-sized planetesimal. Thermal remnant magnetism studies reveal a single coherent magnetic  
43 field among constituent chondrules ([Carpornen et al., 2011](#)). This requires the presence of a partially  
44 differentiated structure with a molten core and silicate mantle overlain by a cold chondritic "lid" as  
45 described by the model of Weiss and Elkins-Tanton, ([2013](#)). Furthermore, isotopic signatures of CV

46 chondrites ( $\epsilon^{54}\text{Cr}$  vs.  $\epsilon^{50}\text{Ti}$  and  $\epsilon^{54}\text{Cr}$  vs.  $\Delta^{17}\text{O}$ ) fall within the carbonaceous supergroup, requiring an  
47 outer solar system origin or late accretion history (Warren, 2011). In addition to a partially  
48 differentiated interior, the CV chondritic “lid” attests to a protracted episode of parent body  
49 evolution with evidence for aqueous alteration (variously characterised by oxidation, hydration and  
50 replacement [e.g. phyllosilicate-rich chondrule rims, Tomeoka and Tanimura, 2000 or oxidized metal  
51 and mobilized Na and K, Krot et al., 2004]) as well as recording the highest grade shock deformation  
52 among the carbonaceous chondrites (commonly S2-S3 shock stages, Scott et al., 1992) and thermal  
53 metamorphism ( $\sim 300^\circ\text{C}$ , Krot et al., 1995; 1998).

54 The CV class is divided into two subgroups: a reduced ( $\text{CV}_{\text{red}}$ ) and an oxidized group ( $\text{CV}_{\text{ox}}$ ) –  
55 dependent on their ratio of metal/magnetite and on the Ni content of sulfide phases (Mcsween,  
56 1977). The oxidised CVs are further classified into Bali-like ( $\text{CV}_{\text{oxB}}$ ) and Allende-like ( $\text{CV}_{\text{oxA}}$ ) subtypes,  
57 with the Bali-like population containing high abundances of hydrated minerals (up to 4.2 vol%,  
58 Howard et al., 2010) (including Fe-phyllosilicate) and nearly pure fayalite ( $\text{Fa}_{90}$ ) (e.g. Kaba:  $\text{Fo}_{100}=20.9$   
59 vol% and  $\text{Fo}_{90}=23$  vol%; Mokoia:  $\text{Fo}_{100}=19.3$  vol% and  $\text{Fo}_{90}=13.3$  vol%; Howard et al., 2010), while the  
60 Allende-like population lack hydrated minerals altogether and do not contain pure fayalite  
61 (Weisberg et al., 2006; MacPhearson & Krot, 2014). Further differences are observed in the  
62 abundance and speciation of secondary minerals formed by metasomatic alteration (MacPhearson &  
63 Krot, 2014). Fe-rich olivine and diopside-hedenbergite are found in all CVs.  $\text{CV}_{\text{ox}}$  and  $\text{CV}_{\text{red}}$  groups can  
64 be distinguished as  $\text{CV}_{\text{ox}}$  (especially the Allende-like members) contain nepheline, sodalite,  
65 andradite, magnetite and Fe-Ni-sulfides, whereas, these minerals are rare or absent in the  $\text{CV}_{\text{red}}$   
66 group that instead contain kirschteinite (MacPhearson & Krot, 2014).

67 These distinct petrographies require different alteration histories operating on the theoretical  
68 pristine CV material. The oxidised subgroup is enriched in alkali elements and Fe-rich silicates,  
69 attesting to Fe-alkali-halogen metasomatism (Krot et al., 1995; 1998). However, the formation of  
70 fayalite remains unresolved. Different processes have been invoked such as: alteration of low-Ca  
71 pyroxene into fayalitic olivine (Housley and Cirlin, 1983), hydrothermal growth of fayalite (Krot et al.,  
72 2004) or the formation of fayalite through dehydration of phyllosilicates (Brearley, 1999). At present,  
73 the most accredited mechanism of formation of fayalite is development during thermal  
74 metamorphism by growth from an amorphous precursor phase whilst in the presence of aqueous  
75 fluids (Abreu and Brearley, 2011). Thermodynamic constraints on the formation of fayalite were  
76 delineated by Zolotov et al. (2006) who suggested that the presence of Ca-Fe-silicates (fayalite,  
77 nepheline, sodalite, diopside-hedenbergite, andradite, grossular, kirschteinite and phyllosilicates)  
78 are clear evidence of alteration from primary minerals held inside CAIs (i.e. melilite, anorthite, Al-Ti-  
79 diopside, hibonite, spinel and perovskite) (Krot et al., 1995) with dissolution of anorthite and albite  
80 in chondrule mesostasis releasing CaO and  $\text{SiO}_2$  for secondary mineral growth (Krot et al., 1995).

81 Among micrometeorite collections only two particles have previously been recognized as CV-like  
82 (Genge, 2010 and van Ginneken et al., 2012), although neither were the subject of a detailed  
83 investigation into their parent body processing, a task that we discuss in this work with a more  
84 detailed characterisation of the CV-like TAM5.29 particle. Furthermore, the classification of these  
85 samples as a CV-like reveals some issues. The particle described by van Ginneken et al. (2012)  
86 contains wüstite instead of Fe-Ni metal and the magnetite composition does not fall in the CV  
87 magnetite field. On the other hand, the sample described by Genge (2010) shows olivines with  
88 rounded morphologies and a high density of crystal defects, which are not typical characteristics of  
89 CVs. For this reason, van Ginneken et al. (2012) and Genge (2010) also suggested an affinity with CO  
90 chondrites. In addition TAM5.29 has andradite inclusions, diopside crystals and oriented petrofabric

91 of olivine that were not detected in the two other CV-like particles described by Genge (2010) and  
92 van Ginneken et al. (2012), making TAM5.29 different and unique with an enhanced metasomatic  
93 history and with an undoubted metamorphosed CV-like mineralogy.

94 Micrometeorites from anhydrous CC groups are relatively rare, although up to the 50% of the  
95 incoming micrometeorite flux have oxygen isotopic compositions related to the CO/CV/CK  
96 anhydrous supergroup (Suavet et al., 2010). Thus, the characterisation of unique micrometeorites  
97 and those derived from rare parent bodies is crucial for our collective understanding of the near-  
98 Earth dust complex and the diversity of the asteroid belt.

99 Here we provide a detailed characterisation of the third CV-like cosmic dust particle with unique  
100 mineralogy and fabric, expanding knowledge in the compositional range of micrometeorites as well  
101 as investigating the thermal metamorphism and hydrothermal history of the CV parent body-(ies).  
102 Furthermore, given that micrometeorites originate from the asteroid belt (Genge et al., 2008; van  
103 Ginneken et al., 2012) and cometary sources (Noguchi et al., 2015), the study of this sample can be a  
104 useful support to recent and upcoming space missions to C-type asteroids (e.g. NASA-Dawn mission  
105 at [1] Ceres, Hayabusa 2 at [162173] Ryugu, Osiris-Rex at [101955] Bennu) giving possible insights  
106 into the surface and subsurface composition and geological events. In particular, TAM5.29 with its  
107 particular metasomatic history adds knowledge on aqueous alteration and hydrothermalism, that  
108 are known to characterise C-type asteroids, and the relative products: secondary mineral formation  
109 in particular hydrous minerals like phyllosilicates, the formation of opaque phases and post accretion  
110 processing of organic matter, especially aliphatic and aromatic hydrocarbons.

## 111 **2. Methods**

### 112 **2.1 Sample collection**

113 TAM5.29 (~300  $\mu\text{m}$  x ~600  $\mu\text{m}$ ) was recovered from Frontier Mountain (72°59'S–160°20'E; at  
114 ~2800m above sea level and ~600m above the local icesheet surface), within Victoria Land,  
115 Antarctica. TAM5.29 was collected by the Italian Programma Nazionale delle Ricerche in Antartide  
116 (PNRA) during the 2003 expedition (Folco et al., 2008). Frontier Mountain consists of igneous rocks  
117 belonging to the Granite Harbour Igneous Complex. The top of the mountain is characterised by flat-  
118 glacially eroded surfaces, created by an overriding icesheet in the past (Folco et al., 2008). On these  
119 flat surfaces numerous joints and weathering pits are found. These are filled with loose fine-grained  
120 bedrock detritus in which thousands of micrometeorites accumulated during the last ~1-2 million  
121 years along with a relatively small component of background terrestrial sediment (Folco et al., 2008).

### 122 **2.2 Petrography and major element analysis**

123 Particle TAM5.29 was embedded in epoxy resin, sectioned and polished. The resulting thick section  
124 was used for whole-section petrographic analysis using a field-emission scanning electron  
125 microscope (ESEM-FEG-STEM FEI Quanta 450), equipped with microanalytical EDS (Energy Dispersive  
126 Spectrometry) Bruker, QUANTAX 400 XFlash Detector 6|10, which has a 129 eV spectral resolution  
127 and an intermediate size 10  $\text{mm}^2$  detector plate, capable of rapid-data collection and semi-  
128 quantitative results at the Centro Interdipartimentale di Scienza e Ingegneria dei Materiali (CISIM),  
129 University of Pisa. All analyses were performed under high-vacuum and at a fixed working distance  
130 of 10.0 mm (the optimal sample-to-pole-piece distance to maximise X-ray counts at the EDS detector  
131 on this instrument). Operating conditions are standardized for our lab and use an electron beam  
132 accelerating voltage of 15 kV and an unmonitored beam current. Spectra were acquired with  
133 an acquisition time of 30s maintaining a dead time of approximately 10%. EDS data reported are

134 therefore uncalibrated and standard-less. Weight totals were determined using the Bruker's  
135 "*interactive oxides*" and are quoted as weight normalized values in Table S1-S2. Elemental detection  
136 limits for this instrument are on the order of 0.2-0.5 wt%.

137 Mineral phases were identified by Raman measurements at the Department of Geosciences,  
138 University of Padova, using a Thermo Scientific™ DXR™ Raman Microscope using a 532 nm laser  
139 excitation source. Analyses were performed using a 50x and 100x long working distance objective  
140 with  $\sim 2.5 \text{ cm}^{-1}$  spectral resolution,  $\sim 1 \text{ }\mu\text{m}$  spatial resolution and 25  $\mu\text{m}$  pinhole operating at a  
141 minimum of 1 mW to a maximum of 5 mW of power. Low power (1 to 5 mW) coupled with short  
142 exposure times of 3-4 s was essential to avoid damage to minerals and carbonaceous phases. To  
143 minimise noise, each spectrum was acquired 10 to 15 times. Spectra were recorded in the  
144 frequency range from 100 to 3500  $\text{cm}^{-1}$ . Spectral fitting was carried out using the Thermo Scientific™  
145 OMNIC™ Spectra Software. Rectangular areas were analysed with the Raman *point-by-point*  
146 mapping technique – again each spectrum in the map was collected 10-15 times using an exposure  
147 time of 3 to 4 s and spectra were obtained from a grid of points spaced 2  $\mu\text{m}$  along X axis and 2  $\mu\text{m}$   
148 along Y axis.

149 EBSD analysis was performed on a CamScan 2500 SEM (Department of Geosciences, University of  
150 Padova) equipped with a LaB6 source, a NordlysNano EBSD detector of Oxford Instruments and  
151 Channel 5.12 EBSD acquisition- and post-processing software. The sample was for one-hour Syton  
152 polished to remove surface damage related to conventional diamond polishing, and was then carbon  
153 coated (few nanometers of thickness) to improve conductivity. Operation conditions were 25 mm  
154 working distance, 15 kV beam acceleration and 10 nA probe current. Considered the small grain size  
155 of the olivine crystals, for EBSD mapping in automated mode a 0.1  $\mu\text{m}$  step-size in X and Y directions  
156 was applied during acquisition of a 600 x 400 data grid. EBSD does not discriminate between solid-  
157 solutions of olivine, garnet and Clinopyroxene (Cpx), therefore forsterite and diopside reflector files  
158 of the HKL database were used to index olivine and Cpx, whereas andradite of the American  
159 Mineralogist Crystal Structure Database ([Downs and Hall-Wallace, 2003](#)) was used to index  
160 andradite. Indexing of the EBSD-patterns was accepted when at least 6 Kikuchi bands were detected.  
161 EBSD data were processed using the Oxford Instruments HKL software package Channel 5.12,  
162 generating crystallographic orientation maps, band contrast maps, phase maps and pole figure plots  
163 after noise reduction. Latter was done by removing isolated misindexed data points using a  
164 wildspike correction, whereas all non-indexed points were infilled to a six nearest-neighbour  
165 crystallographic orientation by extrapolation. In the Channel 5.12 software package grain detection  
166 in EBSD maps is based on crystallographic orientation, using a misorientation angle of at least 10°  
167 between two adjacent pixels to identify grain boundaries. Grain orientation data from the entire  
168 map were plotted onto lower hemisphere equal area projections as one point per grain to avoid  
169 grain size related bias during contouring.

170 We determined bulk mineralogy through  $\mu\text{XRD}$  diffraction methods, using a Rigaku Rapid II micro-  
171 diffraction system, equipped with a 2D curved imaging plate detector, at the Department of Earth  
172 Science, The Natural History Museum in London. This was employed to collect in-plane diffraction  
173 pattern data from the TAM5.29 sample. A Cu X-ray source with an incident beam monochromator  
174 provided  $\text{K}\beta$  filtered Cu-K $\alpha$  radiation (1.5418 Å). This was collimated by a pinhole system to a beam  
175 spot of 100 $\mu\text{m}$ . Analysis ran for 10 hours, after which the 2D diffraction image was converted to a 1D  
176 XRD pattern following automated removal of the background signal and integration of the Debye

177 rings. Data were collected on cross section samples using a constant  $\omega$  angle of  $16^\circ 2\theta$  and a rotating  
178  $\phi$  axis to maximise the number of crystallites and the randomness of their orientations in the X-ray  
179 beam. Low-angle ( $<16^\circ 2\theta$ ) diffraction peaks could not be collected because the polished resin block,  
180 which holds the particles prevented the stage from rotating in the  $\omega$  plane by  $<16^\circ 2\theta$ . Peak positions  
181 in the converted 1D patterns were identified by comparison against a mineral reference database  
182 (PDF-4 database from ICDD).

183 EMPA spot analyses were carried out at the Department of Geosciences, University of Padova with a  
184 CAMECA SX50 instrument with 5 wavelength dispersive spectrometers. Data were acquired with a  
185 beam current of 20 nA, accelerating voltage of 20 kV, defocused beam diameter of 5  $\mu\text{m}$  and we  
186 used for each element an acquisition time of 10 seconds. Amelia plagioclase, olivine, orthoclase,  
187 diopside, sphalerite (blenda), synthetic  $\text{MnTiO}_3$  and  $\text{Cr}_2\text{O}_3$  standards were used for instrumental  
188 calibration. The Pouchou-Pichoir procedure (PAP), supplied by the manufacturer, was used for raw  
189 data reduction. Detection limits (wt%) are:  $\text{Na}_2\text{O}=0.04$ ;  $\text{MgO}=0.02$ ;  $\text{Al}_2\text{O}_3=0.02$ ;  $\text{SiO}_2=0.3$ ;  $\text{SO}_3=0.03$ ;  
190  $\text{K}_2\text{O}=0.02$ ;  $\text{CaO}=0.02$ ;  $\text{TiO}_2=0.02$ ;  $\text{Cr}_2\text{O}_3=0.04$ ;  $\text{MnO}=0.04$ ;  $\text{FeO}=0.04$ . We also used EMPA spot  
191 analyses to obtain the bulk composition of TAM5.29 by averaging a grid of 122 randomly spaced  
192 EMPA analyses in a specific representative area chosen on the basis of EDX maps, especially to avoid  
193 areas with major terrestrial alteration. Supporting standards-less EDS spot analyses were conducted  
194 on the FE-SEM at the University of Pisa.

195 Reflectance near-IR spectra were acquired at the laboratories at the IAPS-INAF (Istituto di Astrofisica  
196 e Planetologia Spaziali – Istituto Nazionale di Astrofisica). We used a microscope Micro-IR Hyperion  
197 2000 FTIR Vertex Bruker<sup>®</sup>. The spectra were acquired in the spectral range between 1.3- 22  $\mu\text{m}$   
198 (here we focused and report only the range between 1.5 – 4.2  $\mu\text{m}$ ), with a MCT detector. Infragold  
199 (Labsphere<sup>®</sup>) has been used to calibrate spectral reflectance. The spectra were acquired with a  
200 spectral resolution of  $2\text{ cm}^{-1}$  and an aperture on the sample of  $150 \times 150\ \mu\text{m}$ .

## 201 **3. Results**

### 202 **3.1 Petrography and mineral chemistry**

203 TAM5.29 is a  $\sim 300\ \mu\text{m} \times \sim 600\ \mu\text{m}$  sized particle with a partial magnetite rim only found along the  
204 fusion crust. The melt layer is discontinuous and the texture is similar to the fusion crust found on  
205 meteorites (Fig. 5A, Fig. S1). Voids are recognisable in the melt layer (Fig. 5A). High resolution FE-  
206 SEM imaging demonstrates that the particle is composed primarily of lath-shaped olivine crystals  
207 with widespread andradite inclusions surrounded by dark halos composed of intermixed pyroxene  
208 and jarosite, which also occur as alteration veins (Fig. 1A-B). These inclusions form lenses with an  
209 augen-like texture (Fig. 1A-B-E). A few of these lenses show a distinct asymmetrical shape (Fig. 1B).  
210 Olivine is euhedral to subhedral with dimensions from a few micrometres up to  $\sim 10\ \mu\text{m}$ . Andradite  
211 appears as sub-rounded crystals 5-10  $\mu\text{m}$  in size, while the diopside and jarosite surrounding these  
212 are anhedral and difficult to distinguish from one another in backscattered electron images since  
213 they are small crystals (few micrometres) finely mixed and with similar greyscales. EDS spot analyses  
214 (Table S1) reveal fayalitic olivines, with heterogeneous grain compositions, ranging from  $\text{Fa}_{42.5}$  to  
215 almost pure fayalite  $\text{Fa}_{92.3}$ . Rare crystals of forsterite are also present ( $\text{Fo}_{66-71}$ ). Similarly, pyroxenes  
216 have heterogeneous grain compositions ( $\text{Fs}_{1.8-60}\text{Wo}_{0.7-48}$ ). Owing to the small grain size, pores and  
217 limitations with the spatial resolution and the interaction volume of the electron microprobe, spot  
218 analyses on some phases suffer from beam overlap with adjacent hydrated minerals, oxides and  
219 sulphides, thus analyses shown in table S1 may reveal low weight totals (olivine 89 wt%; andradite

220 95 wt%). We therefore supported mineral identification with spatially resolved Raman and bulk  
221  $\mu$ XRD measurements.

222

223 Raman data revealed pyroxenes to be diopside (Fig. 2) with sparsely distributed isolated enstatite  
224 grains. In figure 2 a Raman map shows the typical microtexture of this particle. Furthermore, an  
225 aggregate of spinel crystals (mean  $\text{Cr}/(\text{Cr}+\text{Al})=0.003$  and  $\text{Fe}/(\text{Fe}+\text{Mg})=0.34$  based on EMPA analyses)  
226 are present (Fig. 1C-10, Table S1-S2), these co-occur with micron-sized Fe-oxides (mainly magnetite,  
227 spinel and ferrihydrite). Only minor FeNi alloys are found (Fig. 10 and Table S2) dispersed within the  
228 matrix (Fig. 1). Raman analysis together with EDS, EMPA and EBSD suggest that a considerable  
229 portion of the olivine has been altered resulting in a mixture of fine-grained hydrated Mg-Fe-  
230 sulphur-rich minerals and minor carbonates (likely Fe-carbonates) as suggested by the 3.9  $\mu\text{m}$  band  
231 in TAM5.29 IR spectra (Fig. 11), that we identify as iddingsite (see paragraph 3.3 for a more detailed  
232 description). Fine-grained phases are found as weathering films on fayalite and also scattered  
233 throughout the whole particle. Given their cryptocrystalline nature it is difficult to definitely resolve  
234 individual mineral phases, however, fibrous phyllosilicates are clearly resolved as dark haloes around  
235 andradite inclusions (Fig. 1A-B-F). Micro-XRD revealed that these fibrous phyllosilicates are mainly  
236 antigorite and saponite (Fig. 8).

237

238 The bulk composition of TAM5.29 shown in table 1 is similar to that of CCs, being within 2 orders of  
239 magnitude of CI values, and similar to other unmelted fine-grained Antarctic micrometeorites  
240 (UMM) (Fig. 3). However, TAM5.29 also demonstrates notable enrichment in Fe and depletion in Mg  
241 compared to the CCs, as also suggested by the 2.8  $\mu\text{m}$  band (Fig. 11) typical of Fe-rich hydrous  
242 phases (Takir et al, 2013). The aluminium content of TAM5.29 is similar to that of CV meteorites  
243 (Table 1 and Fig. 3) and is significantly higher than other CCs – consistent with the high  
244 concentrations of refractory elements found in CV and CK chondrites. Conversely Ca and Ti are  
245 depleted compared to CVs. The spider diagram in figure 3 shows a strong enrichment in K in  
246 TAM5.29 compared to both CVs and other UMM. If we consider bulk composition of only the matrix  
247 of CVs (Table 1 and Fig. 3) we see that FeO-MgO-CaO-TiO<sub>2</sub> values are closer to those of TAM5.29. On  
248 the contrary Al<sub>2</sub>O<sub>3</sub> and Cr<sub>2</sub>O<sub>3</sub> CVs matrix values are considerably different from the values of  
249 TAM5.29, which are closer to the average CVs (considering also chondrules). Slight depletion of Na<sub>2</sub>O  
250 in TAM5.29 compared to both CVs and CVs matrix is also found.

251

252 Carbon is also ubiquitous, Raman analyses (Fig. 2 and 6) identified characteristic “G” and “D” band  
253 peaks (located at  $\sim 1350$  and  $\sim 1590$   $\text{cm}^{-1}$  respectively) and associated with disordered carbonaceous  
254 phases (Bonal et al., 2006). Poorly Graphitized Carbon (PGC) occurs as tiny inclusions of 100-200 nm  
255 in fayalite and as thin films surrounding crystals (Krot et al., 1998; Abreu and Brearley, 2011). It is  
256 interesting to note that Raman spectra of the various mineral phases (fayalite, andradite, diopside,  
257 jarosite and fine grained material) also have peaks around 2680 and 2930  $\text{cm}^{-1}$ . These peaks are  
258 second order peaks of C but also attest to the presence of OH as well as S-H and C-H functional  
259 groups within organic molecules. Sulphur is detected in all EMPA and EDS analyses, reflecting either  
260 S in jarosite, S-bearing organics or S-rich phyllosilicates. IR reflectance spectra (Fig. 11) also show the  
261 presence of organic matter (i.e. CH compounds, aromatic and aliphatic hydrocarbons) with the 3.3  
262 and 3.4  $\mu\text{m}$  bands, the broad band between 3.6 and 3.8  $\mu\text{m}$  can be instead indicative of S-H  
263 compounds. IR spectra also indicate the presence of carbonates with the 3.9  $\mu\text{m}$  band (Fig. 11).

264 Furthermore, all EMPA and EDS analyses show low totals, indicating ubiquitous presence of OH and  
265 CH functional groups, in agreement with Raman observations.

266

267 Another important observation is a shock melt vein 136  $\mu\text{m}$  in length and approximately 5  $\mu\text{m}$  in  
268 thickness (Fig. 1D). This feature cross-cuts the fayalitic groundmass and either deforms or displaces  
269 the primary features (an andradite-diopside vein Fig. 1D). In addition, we observe a conjugate  
270 synthetic fracture and release band (Fig. 1D). The displacement of these features appears to show a  
271 dextral shear sense, although this interpretation remains uncertain owing to significant variations in  
272 the width of the Jarosite/diopside band on either side of the melt vein. The lower margin of the  
273 linear feature (as seen in Fig. 1D) shows an abrupt compositional contact with the host groundmass,  
274 while the upper margin is transitional over approximately 5  $\mu\text{m}$ . This feature is composed of a  
275 nanocrystalline or glassy matrix with a high porosity and hosting anhedral rounded small (<2  $\mu\text{m}$ )  
276 olivine crystallites and minor Fe-Ni oxides.

277

### 278 **3.2 Preferred orientation of olivine**

279 Figure 4 reports EBSD map data acquired on the same site shown in Fig. 1B where olivine crystals  
280 wrap around andradite-diopside/jarosite inclusions and form a foliation texture (Fig. 1B). To  
281 investigate whether the olivine crystals show a crystallographic preferred orientation (CPO), a  
282 detailed EBSD map (Fig. 4) was collected. In figure 4A, a phase map, olivine (red), andradite (green)  
283 and diopside (yellow) are shown. The dark part of the map is material that was not indexed. It is  
284 unlikely that the material between the olivine crystals has not been indexed because of potential  
285 polishing problems, since micro-Raman, SEM, EMPA and micro-XRD analysis shows the presence of  
286 fine grained alteration products (see below).

287

288 The three pole figures (Fig. 4B), lower hemisphere equal area projections, are obtained considering  
289 one point per olivine grain. This has been done to avoid a bias of the data distribution due to grain  
290 size effects. Pole figures in figure 4B show the pertinent crystallographic orientation of every single  
291 olivine crystal. In these pole figures a clear trend is recognized, with the [100] axis forming a  
292 maximum in the NW periphery, suggesting a NW-SE preferred orientation of this direction, hence  
293 orthogonally to the foliation trace. Conversely, the [010] and the [001] directions are dispersed along  
294 girdles consistent with the overall trend of the foliation trace. This implies that the [100] axis is a  
295 rotation axis, and that the differently elongated shape of the olivine crystals is the result of crystals  
296 having their [010] and [001] directions at systematically changing angles within the foliation plane.

297

### 298 **3.3 Fine-grained material**

299 With EDS, EMPA and Raman analyses widespread fine-grained cryptocrystalline phases are found in  
300 the TAM5.29 dust grain along with the previously described iddingsite. Raman analyses produce  
301 mixed spectra suggesting the presence of sulfides, probably related to the S-H bands as well as  
302 poorly crystalline hydrated phyllosilicates (Fig. 7). The Raman spatial resolution was 1  $\mu\text{m}$  (see  
303 methods), evidence that the fine-grained material has a grain size below 1  $\mu\text{m}$ . Micro-XRD data (Fig.  
304 9-10) are in agreement with this interpretation, revealing Fe-Ni sulfides, phyllosilicates and possibly  
305 oxides and hydroxides. Carbonates are not clearly detected by Raman and micro-XRD, but their  
306 presence is inferred by the 3.9  $\mu\text{m}$  band in the IR spectra (Fig. 11).

## 307 **4. Discussion**

### 308 **4.1 TAM5.29: micrometeorite or meteorite?**

309 TAM5.29 has a discontinuous melt layer with a texture similar to the fusion crust found on  
310 meteorites (Fig. 5A, Fig. S1). The presence of such a prominent melt layer is unusual for a  
311 micrometeorite exterior, which instead have a thin (~5-50  $\mu\text{m}$ ) double-layered igneous rim and  
312 magnetite rim (Genge, 2006). This could imply that TAM5.29 is a fragment of a larger meteorite.

313 Although meteorite fusion crusts are highly variable in composition, texture and thickness (Genge  
314 and Grady, 1998) they are commonly >1000  $\mu\text{m}$  (that is 10x thicker than the crust on TAM5.29, Fig.  
315 S1). In addition, the bulk composition of CV fusion crust is also markedly different from bulk  
316 composition of TAM5.29 (Table 1). As reported by Genge and Grady (1998), they have MgO= 25.65  
317 wt%; SiO<sub>2</sub>= 34.5 wt%; FeO= 30.46 wt%; CaO= 2.39 wt% and Al<sub>2</sub>O<sub>3</sub>= 3.34 wt%. On the contrary  
318 TAM5.29 has major elements bulk composition (Table 1) of MgO= 12.6 wt%; SiO<sub>2</sub>= 29.2 wt%; FeO=  
319 40.1 wt%; CaO= 1.41 wt% and Al<sub>2</sub>O<sub>3</sub>= 4.93 wt%. Furthermore, because the melt layer on TAM5.29 is  
320 discontinuous, this requires that the particle fragmented whilst on the Earth's surface, as evidenced  
321 by the lack of fusion features on the remainder of the particle perimeter. We estimate that the pre-  
322 atmospheric particle diameter – assuming a spherical shape (a simplistic approximation given fusion  
323 crusts can show protrusions) at 1300-2000  $\mu\text{m}$ , approaching the size limit for micrometeorites. Thus  
324 TAM5.29 may represent a different class of extraterrestrial material intermediate in size between  
325 micrometeorites (<2000  $\mu\text{m}$ ) and meteorites (~>1 cm). This size domain was previously described by  
326 Harvey and Maurette (1991), Kurat et al., (1994) and Folco and Rochette (2010) and termed  
327 “*minimeteorites*”.

328

329 The hypothesis that TAM5.29 was a small particle in space is further supported by its unique  
330 petrology (chondrule-free nature and mixed metamorphic history). Owing to their unique delivery  
331 mechanism (P-R drag, operating on grains <1 cm, Gonczi et al., 1982) cosmic dust sample a more  
332 diverse collection of asteroid parent bodies than their larger meteorite counterparts. Thus, because  
333 TAM5.29 represents a new lithology, this adds to our confidence that this sample is not simply a  
334 fragment of a larger meteorite broken off during atmospheric entry but instead existed as a grain of  
335 dust in interplanetary space derived from a parent body not currently delivering meteorites to Earth  
336 and otherwise unsampled.

337 Finally, the preferential fragmentation of a heterogenous material, with fractures following  
338 boundaries of chondrules and CAIs may explain their absence in this minimeteorite, thereby  
339 reflecting the unrepresentative sampling (of coarse-grained features within parent body), as  
340 previously argued by Genge et al. (2008).

341

### 342 **4.2 Terrestrial weathering**

343 Micrometeorites from the Transantarctic Mountains often show moderate to significant terrestrial  
344 weathering. In particular, jarosite is a common product of formed in subaerial Antarctic  
345 environments (van Ginneken et al., 2016). The EDX map in figure 5C-D reveals co-occurrences of S  
346 and K, the main components of jarosite, concentrated along the particle border and in the fractures  
347 within TAM5.29. The effect of terrestrial weathering is however limited being confined to the  
348 fractures and margins. The EDX map in figure 5C reveals that high Ca concentrations correspond to  
349 andradite inclusions, unaffected by replacement. In addition, Mg and Fe-element maps (Fig. 5B) are  
350 well delineated, suggesting that these elements have not been leached by terrestrial fluids. It  
351 appears that TAM5.29 has therefore suffered limited weathering, retaining the majority of its pre-



352 atmospheric mineralogy unaltered. Because the melt vein displaces a dark vein of diopside + jarosite  
353 (marked in Fig. 1D), we conclude that the shock event post-dates the metasomatic event. However,  
354 it remains unclear whether iddingsite is also intersected by the melt vein or if alteration that created  
355 the iddingsite overprints the shock vein. We favour the latter option assuming that iddingsite formed  
356 by fluids released by hydrous minerals as a consequence of the pressure exerted by the impact  
357 (because olivine in TAM micrometeorites typically weathers to jarosite and not iddingsite, [van  
358 Ginneken et al., 2012](#)). Later fractures instead cut the shock melt vein creating possible localized  
359 terrestrial alteration also on the shock melt vein.

360

#### 361 **4.3 Petrogenesis – Record of hydrothermal environment on the Parent Body**

362 The mineralogy displayed by TAM5.29, primarily composed of fayalitic olivine plus Ca-Fe rich  
363 pyroxenes, andradite, phyllosilicates and sulphides as well as the close matrix compositions  
364 demonstrate a clear affinity to the CV chondrite group. The mineralogy in TAM5.29 is, however,  
365 distinct from any currently reported CVs and also unique among reported micrometeorite studies.

366

367 TAM5.29's most common mineral – fayalite is associated with Fe-alkali-halogen metasomatism at  
368 temperatures <300°C ([Krot et al., 1995-1998](#); [Zolotov et al., 2006](#); [MacPhearson and Krot, 2014](#)). The  
369 preferred fayalite growing mechanism in CVs is during thermal metamorphism from an amorphous  
370 precursor phase in the presence of fluids ([Abreu and Brearley 2011](#)) and Ca-Fe-rich phases are  
371 derived from the aqueous alteration of CAIs and anorthitic-alibitic mesostasis ([Krot et al., 1995](#)).

372 Fe-rich olivine and diopside-hedenbergite assemblages are common in all CVs, while Ca-rich silicates  
373 (e.g. andradite) and Fe-Ni sulfides are almost absent in CV<sub>red</sub>, which instead contain kirschsteinite  
374 ([MacPhearson and Krot, 2014](#)). Nepheline and sodalite are typical of CV<sub>oxA</sub> and absent in CV<sub>oxB</sub>, which  
375 in turn contain phlogopite and saponite ([MacPhearson and Krot, 2014](#); [Krot et al., 1998](#)). TAM5.29 is  
376 therefore consistent with the CV<sub>oxB</sub> group's mineralogy, having several andradite inclusions plus  
377 phyllosilicates. This metasomatic alteration is similar to the terrestrial serpentinization-  
378 rodingitization process ([Python et al., 2007](#); [Bach and Klein, 2009](#)). During serpentinization at  
379 temperatures <385°C clinopyroxene replaces tremolite and, at lower temperatures, below 275°C,  
380 clinopyroxene is in turn replaced by andradite ([Python et al., 2007](#); [Bach and Klein, 2009](#)).

381 In summary, diopside and andradite grow at the expense of serpentine and, at lower temperatures  
382 (<275°C), andradite also replaces diopside ([Python et al., 2007](#); [Bach and Klein, 2009](#)). Tremolite and  
383 disordered biopyriboles have been found in Allende chondrules and their presence represents a  
384 peak metamorphic temperature below 340°C ([Brearley 1997](#)). In TAM5.29 tremolite is not found.  
385 Furthermore, serpentine and diopside are small anhedral crystals that form dark haloes around  
386 andradite that appear to consume them. This provides a temperature constraint on the formation of  
387 secondary phases in TAM5.29, between 275 and 250°C, in agreement with temperatures estimated  
388 by Krot et al. ([1995-1998](#)). Krot et al. ([1995-1998](#)) based the temperature range mainly on the CV  
389 tensile strength, liquid water can exist at a maximum temperature of 310°C. This limit of 310°C is  
390 also supported by textural observations of CVs, O-isotopic composition and thermodynamic analysis  
391 ([Krot et al., 1995-1998](#)).

392

393 The bulk composition of TAM5.29 supports the Fe-alkali-halogen metasomatism hypothesis.  
394 Significant enrichment of Fe and K are observed compared to average CVs bulk composition. The  
395 enrichment in Fe is indicative of highly oxidizing conditions. However, Fe value of CVs matrix is  
396 comparable with the concentration of Fe in TAM5.29. This is probably due to the higher

397 concentration of Fe-oxides (i.e. magnetite and alloys) in the matrix of CVs. Instead, K is easily  
398 leached from minerals by fluids. The high concentration of K is explained with jarosite formed by  
399 terrestrial alteration (even if bulk analyses were made in areas where K enrichment due to terrestrial  
400 alteration was as low as possible based on EDX maps). It is also possible that part of the K is  
401 representative of hydrothermal fluid circulation rich in alkali elements, even if this alone cannot  
402 account for the very high K abundance in TAM5.29. In fact, localized heating on C-type bodies (e.g.  
403 Ceres) have been proposed to be due to leaching and re-deposition near the surface of long-lived  
404 radioisotopes like  $^{40}\text{K}$ , which decays in  $^{40}\text{Ar}$  (Castillo-Rogez et al., 2008). The Ca and Ti depletion  
405 compared to CVs is instead due to the absence of refractory inclusions in TAM5.29. However Ca and  
406 Ti concentration of TAM5.29 is higher than that in the CVs matrix. This enrichment is due to the  
407 higher concentration of Ca-rich minerals and Ti-oxides in TAM5.29, compared to the CVs matrix,  
408 formed during metasomatism. In addition, Ca depletion relative to CVs can also be related to Ca  
409 leaching that formed carbonate veins (not sampled by TAM5.29) in the parent body, a feature  
410 commonly found in terrestrial serpentinization processes (Python et al., 2007). Mg depletion in  
411 TAM5.29 is a consequence of the oxidising conditions and Fe enrichment. Na is easily depleted  
412 during aqueous alteration and TAM5.29 doesn't have nepheline and sodalite, the main carriers of  
413 Na. Al in TAM5.29 is higher than CVs and CV matrix. This high Al content is tricky to explain but is  
414 likely given by Al in the phyllosilicates derived from leached CAIs.

415 However, under these environmental conditions fayalite is not stable with andradite and Ni-sulfides  
416 (Krot et al., 1998; Zolotov et al., 2006; MacPhearson and Krot, 2014). Andradite is stable at higher  
417 temperatures than fayalite and the two phases can only coexist at equilibrium at  $T < 100^\circ\text{C}$  (Krot et  
418 al., 1998; MacPhearson and Krot, 2014). Temperature  $< 100^\circ\text{C}$  have also been proposed for the  
419 formation of iddingsite in the Lafayette meteorite (Treiman et al., 1993). Raman spectra and EDS  
420 analyses of TAM5.29 also show that andradite retain a certain amount of water (Fig. 2 and Table S1),  
421 thus the OH in andradite could represent a  $(\text{SiO}_4)^{4-} \leftrightarrow (\text{O}_4\text{H}_4)^{4-}$  substitution reaction for which low  
422 temperature formations are also expected within a low-pressure post magmatic environment  
423 (Amthauer and Rossmann, 1998). However structural  $(\text{O}_4\text{H}_4)^{4-}$  is expected to have Raman peaks at  
424  $\sim 3560 \text{ cm}^{-1}$  (Amthauer and Rossmann, 1998). In contrast, in TAM5.29 andradite Raman spectra show  
425 OH peaks at lower wavelengths between  $\sim 2680$  and  $\sim 2930 \text{ cm}^{-1}$  implying the presence of non-  
426 structural water (in addition to S-H and C-H bonds).

427 In support of a low temperature formation, the bulk composition of TAM5.29 shows considerable  
428 enrichment in Fe compared to all other CCs (Table 1 and Fig. 3). The majority of this Fe is held within  
429 the silicate minerals (rather than reduced Fe-Ni metal) suggesting highly oxidizing conditions at low  
430 temperatures.

431 Serpentinization may also occur at lower temperatures but in this case lizardite is also expected,  
432 which is not detected in TAM5.29. Instead, in TAM5.29 the presence of antigorite can only be  
433 formed by serpentinization at higher temperatures, in particular at  $\sim 310^\circ\text{C}$  occurs the chrysotile  
434 breakdown to antigorite + brucite at low pressure (1 bar, which is also typical of minor bodies of the  
435 Solar System, Wunder et al., 2001). In contrast, fayalite can form over a wide range of temperatures  
436 from  $30^\circ\text{C}$  to  $300^\circ\text{C}$  with W/R (water to rock ratio) between 0.06 and 0.2 (Zolotov et al., 2006) and  
437 thus does not provide significant temperature restrictions.

438 Thus, the alteration assemblage in TAM5.29 must have formed across a range of temperatures, with  
439 two distinct alteration periods with distinct environmental conditions. An initial alteration regime  
440 generated the fayalite, andradite, antigorite and diopside at temperatures of  $\sim 275\text{-}250^\circ\text{C}$  under  
441 oxidising conditions. While, later, at a distinctly lower temperature ( $< 100^\circ\text{C}$ ), the iddingsite formed.

442 We conclude that fayalite-andradite-FeS-NiS assemblages derive from a retrograde metasomatism,  
443 inferred mainly by the andradite crystals consuming diopside and serpentine, at ~250°C, suggested  
444 by the mineral assemblage typical of serpentinization. Conversely, the formation of iddingsite  
445 appears to be related to an independent low-T event that will be discussed later in the discussion.

446

#### 447 **4.4 Origin of preferred orientation of olivine**

448 The CPOs seen in the EBSD map (Fig. 4) may at a first glance resemble the terrestrial “type D” fabric  
449 of olivines recognized in simple shear experiments by Bystricky et al. (2000). Olivine “Type D” fabric  
450 refers to high stress and water-poor conditions (Michibayashi et al., 2016), and Bystricky et al. (2000)  
451 associated this fabric to upper mantle conditions, for example in regions of extensional deformation  
452 in major detachment zones near the crust-mantle boundary, or in subduction zones where hot  
453 mantle convects past the upper side of cold slabs. Such deformation regimes are inconsistent with  
454 that recorded by TAM5.29. We believe that pole figures (Fig. 4), showing a clear maximum in the  
455 [100] direction, while the [010] and [001] directions are scattered in two girdles, are describing a  
456 kind of olivine fibre texture that relate to an axially symmetric shortening regime in which the (100)  
457 plane of olivine orients orthogonally to the maximum compression direction ( $\sigma_1$ ) and relate to  
458 compaction rather than to shearing.

459 It is also interesting to note that the elongated olivine in TAM5.29 are generally not associated to the  
460 elongated [100] direction, but are randomly associated to the [010] and [001] directions. Further  
461 examination reveals that the longest olivine crystals show clear alignment around andradite  
462 inclusions. These preferential CPOs concentrated locally around coarse-grained inclusions were  
463 previously described in CV chondrites where olivine crystals wrap around chondrules or dark  
464 inclusions (Watt et al., 2006; Forman et al. 2017). These areas, around chondrules in CVs and around  
465 andradite crystals in TAM5.29, were probably characterised by a higher porosity compared to the  
466 rest of the matrix and thus suffered a more significant compression and pore collapse during impact.  
467 Such a scenario results in heterogeneous strain distribution with significant heat production (~575  
468 °C) and locally high stress at the sites or pore collapse while the surrounding low-porosity matrix  
469 retains a lower degree of compaction and andradite inclusions records no compaction at all (Bland  
470 et al., 2014). The matrix therefore remains partially unaffected by the shock wave passage (Bland et  
471 al., 2014; Forman et al., 2017), while around andradite inclusions compaction is more enhanced with  
472 the creation of augen-like structures as predicted and observed in Allende by Bland et al. (2014). This  
473 process also produces a distinct heterogeneity in the CPOs (Forman et al. 2017), which is in  
474 agreement with our EBSD data.

475 Another intriguing fact is the dislocation of the diopside-jarosite vein in figure 1D. This vein has a  
476 dextral displacement. This kind of deformation is typically non-uniaxial and is in contrast to the  
477 theoretical uniaxial nature of the shock compaction. In support of the non-uniaxial deformation  
478 there's also the asymmetric andradite-diopside inclusion (Fig. 1B), proof of a shear stress. It seems  
479 thus that the preferred orientation of olivine has been created in the instant when the impact  
480 occurred - a moment in which stress is uniaxial. Subsequent to the initial stage of the impact, simple  
481 shear deformations form radial to the impact area (Kenkmann et al., 2014). In this second stage,  
482 deformation has been accommodated along the shock vein melt and around andradite, the two  
483 areas of TAM5.29 where weakness is higher. In particular, maximum strain occurred along the upper  
484 part of the shock vein melt where crystals were forced into the melt vein creating the transitional  
485 upper limit, the conjugate synthetic displacement and the releasing band (Fig. 1D).

486

487 Alternatively, non-impact processes may have generated the preferred orientation of olivine seen in  
488 TAM5.29. These include: sintering, sub-grain recrystallization, lithification and gravitational  
489 compaction (Forman et al., 2016-2017).

490 Sintering requires heating at temperatures higher than 360°C over long (>1 million years) timescales  
491 to achieve recrystallisation alignment (Gail et al., 2015). However, both the temperature and  
492 duration of heating are inconsistent with the formation conditions of TAM5.29, which require lower  
493 temperatures (~250°C) – as determined by the secondary mineral assemblage and shorter durations  
494 – as determined by Raman data and olivine crystal morphology. Likewise, the petrofabrics in  
495 TAM5.29 cannot have been formed by a plastic deformation processes such as sub-grain rotation,  
496 recrystallization and diffusion creep as these would result in significantly less elongation of olivine  
497 crystals and lower aspect ratios (Forman et al. 2017).

498 Petrofabrics formed by compression due to lithostatic forces cannot be ruled out, but it's less likely  
499 as also suggested by Watt et al. (2006), Format et al. (2016; 2017) and Bland et al. (2014). This is  
500 because the pressure on small protoplanets is negligible, especially on porous water-rich  
501 carbonaceous chondritic parent bodies. For example, at the centre of a 200 km diameter asteroid  
502 pressures are thought to reach a maximum of 1MPa (corresponding to a depth of a few tens of  
503 meters on Earth), far too low for lithostatic compaction (Weidenschilling and Cuzzi, 2006).  
504 Meanwhile, on larger Ceres-like bodies, pressure estimates vary between <0.2 GPa (Neumann et al.,  
505 2015) and 1220 MPa (Suttle et al., 2017). Even if sufficient pressures are possible, liberation would  
506 require an impact of an enormous magnitude, equivalent to the complete destruction of such a  
507 body. In both cases we would expect a brecciated texture of the resulting asteroid's chunks. In  
508 TAM5.29 brecciated texture is not observed. However TAM5.29 is a small fragment and does not  
509 necessarily sample an area with clasts boundaries. For this reason we cannot assume that TAM5.29  
510 is not part of a breccia and we cannot completely rule out the possibility of lithostatic compression.

511 Finally, a gravitational compaction model, such as that proposed for Allende by Watt et al. (2006)  
512 may be possible. Here, a muddy outer layer of a parent body affected by sedimentary processes  
513 operating under microgravity result in the alignment of olivine. However, gravity is very low even on  
514 a Ceres-like body (Ceres gravity is ~0.28 m/s<sup>2</sup>, around 1/35 Earth's gravity). The low gravity coupled  
515 with the chaotic [010] and [001] axis distribution of olivines in TAM5.29 (fig. 4), bring us to prefer an  
516 impact-induced compaction (Gattacceca et al., 2005; Bland et al., 2014; Forman et al., 2016-2017).  
517 Consequently, the process that most likely created the preferred orientation of olivines in TAM5.29  
518 is impact compaction, and this is further supported by the presence of the shock melt vein within  
519 the micrometeorite.

#### 520 **4.5 Carbon and Poorly Graphitized Carbon (PGC)**

521 Carbon, which is widely distributed in TAM5.29, is a powerful index of metamorphic grade. In  
522 particular the maturity of the organic matter is influenced by thermal metamorphism and can be  
523 used to establish petrologic types for individual meteorites (Bonval et al., 2006). The maturation  
524 grade of the organic matter can be determined by the study of the Raman D-band (~ 1350 cm<sup>-1</sup>) and  
525 the G-band (~1580 cm<sup>-1</sup>) peak parameters. In the least metamorphosed samples, the intensity of the  
526 G-band I<sub>G</sub> is higher than the intensity of the D-band I<sub>D</sub>, the opposite is seen in samples with higher  
527 metamorphic grade (Bonval et al., 2006). When the I<sub>D</sub>/I<sub>G</sub> ratio of the CV chondrites is compared to the  
528 FWHM (Full Width at Half Maximum) of the D-band (FWHD-D) two distinct groups are recognised  
529 (Fig. 6) (Bonval et al., 2006). The oxidised CVs have the highest I<sub>D</sub>/I<sub>G</sub> ratio (1.05 to 1.55) and lowest  
530 FWHM-D (~60 to ~100 cm<sup>-1</sup>) (lower right, Fig. 6), they are known to have experienced the highest  
531 metamorphic grade (Bonval et al., 2006). In particular, these data show that Allende suffered the  
532 highest thermal metamorphism (Bonval et al., 2006). An exception to this is the meteorite Kaba,

533 which is a  $CV_{oxB}$  that lies in the upper left group (Fig. 6) showing a minor metamorphic grade typical  
534 of the reduced CVs (Bonal et al., 2006). Analysis of the  $I_d$  and  $I_g$  bands from TAM5.29 (Fig. 7) shows  
535 an R1 ratio ( $I_d/I_g$ )>1 and in some other cases  $I_d/I_g \sim 1$  (but never <1), thus TAM5.29 does not clearly  
536 belong to either group. Instead, TAM5.29 values tend towards the most metamorphosed group  
537 ( $CV_{ox}$ ) although several values also lie in the less-metamorphosed group (Fig. 6). This is proof of the  
538 highly unequilibrated nature of this micrometeorite and suggests that TAM5.29 is transitional  
539 between the Kaba-like CVs and the more evolved Allende-like CVs. The reason for this unspecified  
540 petrologic type are, however, not clear. It is known that within the CV parent body many different  
541 environmental conditions existed from oxidising fluid-enriched locations to the reducing fluid-poor  
542 localities. The TAM5.29 metamorphic grade is thus representative of a new lithology of the CV  
543 parent body that experienced more oxidising conditions (resulting in significant Fe enrichment) with  
544 an incomplete thermal metamorphism terminated by the impact that also created the shock melt  
545 vein and preferred orientation of olivine.

#### 546 **4.6 Origin of the fine-grained material**

547 Iddingsite is a common alteration feature that affects olivine in terrestrial rocks and is also found as  
548 a native minor component in chondritic meteorites. Iddingsite forms as a weathering film and  
549 represents a complex mixture of secondary hydrated silicates as well as carbonates, sulphates  
550 halides and oxides. Iddingsite compositions in TAM5.29 are difficult to interpret because of their  
551 very small grain size (<1  $\mu\text{m}$ ) and mixed phase composition, which in turn gives averaged data when  
552 analysed with Raman, EDS and EMPA.

553 Lee et al. (2015) described iddingsite in the Lafayette meteorite as an alteration sequence affecting  
554 olivine and augite concurrent with the formation of hydrous Fe-Mg-phylosilicates. These newly  
555 formed phyllosilicates are then partially replaced by siderite. During the growth of siderite Fe-oxides  
556 also begin to form (Abreu and Brearley 2011). The alteration sequence ends with saponite and other  
557 fibrous phyllosilicates replacing siderite. Tomeoka and Buseck (1985) described similar alteration  
558 features in the matrix of CM chondrites formed as an intergrowth of Fe-Ni-S-O phases and  
559 cronstedtite. Based on these findings we looked for possible constituent minerals of iddingsite  
560 within TAM5.29.

561  
562 Raman spectra of the fine-grained material of TAM5.29 (Fig. 8) show possible matches to  
563 mackinawite, cronstedtite, and chukanovite (from RRUFF database). The best match among these  
564 phases is the hydrated Fe-Ni sulfide mackinawite (the two peaks at  $209\text{ cm}^{-1}$  and  $279\text{ cm}^{-1}$  of  
565 TAM5.29 are also well matched by troilite), although this lacks characteristic peaks around  $525$  and  
566  $888\text{ cm}^{-1}$ . Mackinawite is a poorly crystalline precipitate formed by the reaction between  $\text{HS}^-$  and Fe  
567 (Lennie et al., 1997). In nature, mackinawite occurs as hydrothermal alteration product within  
568 serpentinized peridotites and has also been reported in meteorites. Conditions of formation of  
569 mackinawite are in agreement with the conditions of formation of TAM5.29, making this phase a  
570 plausible candidate.

571 Phyllosilicates are also present. Although cronstedtite can be ruled out due to a lack of diagnostic  
572 within the Raman spectra the  $\mu\text{XRD}$  data (Fig. 9) revealed the presence of saponite and antigorite in  
573 TAM5.29. Phlogopite and clay minerals may also be present but, since diffraction data could not be  
574 collected below  $16^\circ 2\theta$  and clay minerals have their main peaks in this region it is not possible to  
575 obtain further details about these phases. In addition, the  $\mu\text{XRD}$  data also suggested the presence of  
576 Fe-Ni sulfides (Fig. 9) (pentlandite in matrix olivine in Allende was reported by Brearley, 1999) as well  
577 as tentative evidence of Fe-carbonates.

578

579 Collectively, these data imply that the fine-grained material in TAM5.29 is a mixture of fibrous  
580 phyllosilicates (antigorite, saponite and possibly phlogopite-cronstedtite), Fe-Ni sulfides and possibly  
581 Fe-oxy-hydroxides with a possibility of rare carbonates inferred from the 3.9  $\mu\text{m}$  band in TAM5.29 IR  
582 spectra (Fig. 11). This mineralogy is in agreement with the final stage of alteration described by Lee  
583 et al. (2015) from the Lafayette meteorite, and demonstrating that TAM5.29 records a protracted  
584 episode of intense post impact aqueous alteration. Fluids involved in the formation of iddingsite may  
585 therefore derive from the partial dehydration of phyllosilicates (previously formed during  
586 metasomatism) liberated after the impact event. In this scenario iddingsite formation occurs after  
587 the interruption of metasomatism at lower temperatures and in agreement with iddingsite  
588 temperature of formation proposed by Treiman et al. (1993) ( $<100^\circ\text{C}$ ).

589

590 However, low-temperature aqueous alteration of olivine alone cannot explain the entire fine-  
591 grained mineral assemblage. Evidence also exists for the loss of CAIs. Greshake et al. (1996) reported  
592 within four CAIs, several crystals of periclase (MgO), rutile ( $\text{TiO}_2$ ), calcium oxide (CaO) and corundum  
593 ( $\text{Al}_2\text{O}_3$ ) inside and at grain boundaries of the constituent minerals of the inclusions. These oxides  
594 have dimensions of 50-200 nm (most of them under 100 nm), a grain size similar to the fine-grained  
595 material of TAM5.29 ( $<1 \mu\text{m}$ ). Since Ca-rich minerals (andradite and diopside) derive from alteration  
596 of CAIs and PRCs (Plagioclase-rich chondrules, Krot et al., 2002), it is possible that TAM5.29  
597 preserves some of these residual oxides as described by Greshake et al. (1996). CaO,  $\text{Al}_2\text{O}_3$  and  $\text{TiO}_2$   
598 in fact match some of the strongest peaks in the  $\mu\text{XRD}$  pattern of TAM5.29 (Fig. 9). So, in addition to  
599 the previously listed minerals, we believe that residual Ca-Al-Ti oxides are present as relicts of the  
600 primary parent body CAIs and PRCs. However, there are no evidences of residual CAIs in TAM5.29.  
601 We thus infer that the residual CAIs and PRCs minerals were not in situ alteration but were mobilised  
602 by fluids circulation. In fact hibonite and spinel are two of the most resistant CAI crystals to  
603 metasomatic alteration and in strongly altered CAIs Al-Ti-diopside is replaced by ilmenite and  
604 phyllosilicates (Krot et al., 1995).  $\text{Al}_2\text{O}_3$  may also be indicative of the presence of sericite, an  
605 alteration aggregate of fine-grained minerals such as illite, muscovite and palagonite (Al and K rich  
606 minerals) formed by hydrothermal fluids circulation. Palagonite and allophane on Earth are also  
607 alteration products volcanic glasses and water interaction. TAM5.29 had some glass content before  
608 the metasomatic event (that largely created fayalite, see paragraph 4.3). It is possible that part of it  
609 was converted into palagonite-allophane. Allophane can also be enriched in Fe and Ti (Gerard et al.,  
610 2007). The Ti enrichment can explain the  $\text{TiO}_2$  detection in TAM5.29. So, illite and palagonite-  
611 allophane can explain the presence of  $\text{Al}_2\text{O}_3$  and  $\text{TiO}_2$ , but not the presence of MgO and CaO oxides.  
612 Furthermore these minerals are also K-rich, which can explain part of the high K and Al bulk  
613 concentration of TAM5.29. The fine-grained material in TAM5.29 therefore requires two distinct  
614 alteration events and derived from two different processes: Fe-Ni sulfides and oxides are residues of  
615 the metasomatic event and iddingsite components (such as saponite) are derived from weathering  
616 at lower temperature ( $<100^\circ\text{C}$ ) in presence of fluids released from hydrous minerals by an impact.

617

## 618 **5. Conclusions**

619 We document an unambiguous and unique micrometeorite from the CV chondrite group (a member  
620 of the  $\text{CV}_{\text{ox}}$  family), thereby expanding our collective knowledge of micrometeorite parent body  
621 diversity.

- 622 ● Primary mineral phases of TAM5.29 are Fe-rich olivine, andradite and Ca-Fe-rich pyroxenes  
623 plus carbonaceous matter containing OH, S-H and C-H functional groups. Fayalite crystals  
624 grew during thermal metamorphism potentially from an amorphous precursor phase in  
625 presence of fluids.
- 626 ● The fine-grained material is derived by two distinct alteration events. The metasomatic  
627 process created: Ni-Fe sulfides (e.g. mackinawite), Mg-Ca-Al-Ti oxides partly derived by  
628 residual CAIs constituents mobilised by fluids and partly derived by illite-palagonite-  
629 allophane derived by aqueous alteration. Low temperature alteration created: Mg-Fe-  
630 phyllosilicates (saponite and possibly phlogopite), possibly minor Fe carbonates and FeO-OH.
- 631 ● TAM5.29 mineralogy lies in between the CV<sub>oxA</sub> and CV<sub>oxB</sub>. CV<sub>oxA</sub> are rich in andradite,  
632 magnetite and FeNiS like TAM5.29 but lacks of high abundances of hydrated minerals,  
633 common in TAM5.29. Conversely CV<sub>oxB</sub> are rich in hydrated phyllosilicates but contains  
634 almost pure fayalite not found in TAM5.29. TAM5.29 retains a mineralogical assemblage  
635 that might be a link between the CV<sub>oxA</sub> and CV<sub>oxB</sub>.
- 636 ● TAM5.29 retains a mineralogy dominated by thermal metamorphism products formed at  
637 ~275-250°C within the presence of Fe-alkali-halogens-rich fluids and under highly oxidizing  
638 conditions resulting in significant Fe enrichment.
- 639 ● This may represent a newly described alteration environment on the CV parent body, similar  
640 to the conditions recorded by Allende-Axtell-Mokoia-Kaba etc. but with differences. These  
641 differences are: higher oxidizing conditions, heterogeneous thermal metamorphism that  
642 shows different degrees of alteration within only one micrometeorite and a different  
643 secondary alteration history enabled by a particular impact history.
- 644 ● This is the proof of an even more heterogeneous CV parent body(-ies) thus adding a unique  
645 sample to the known CV lithologies.

646 In conclusion, the hypothesis of formation of the TAM5.29 micrometeorite may be divided in three  
647 main stages:

- 648 ● Stage one: metasomatism at ~275-250°C with Fe-alkali-halogens-rich fluids occurred on the  
649 parent body.
- 650 ● Stage two: the particle was involved in an impact that terminated the metamorphic event  
651 resulting in a strongly unequilibrated composition with cryptocrystalline and amorphous  
652 phases and generating a preferred orientation olivine petrofabric.
- 653 ● Stage three: characterised by the formation of iddingsite at lower temperatures, possibly  
654 from fluid released by hydrated minerals during the impact.

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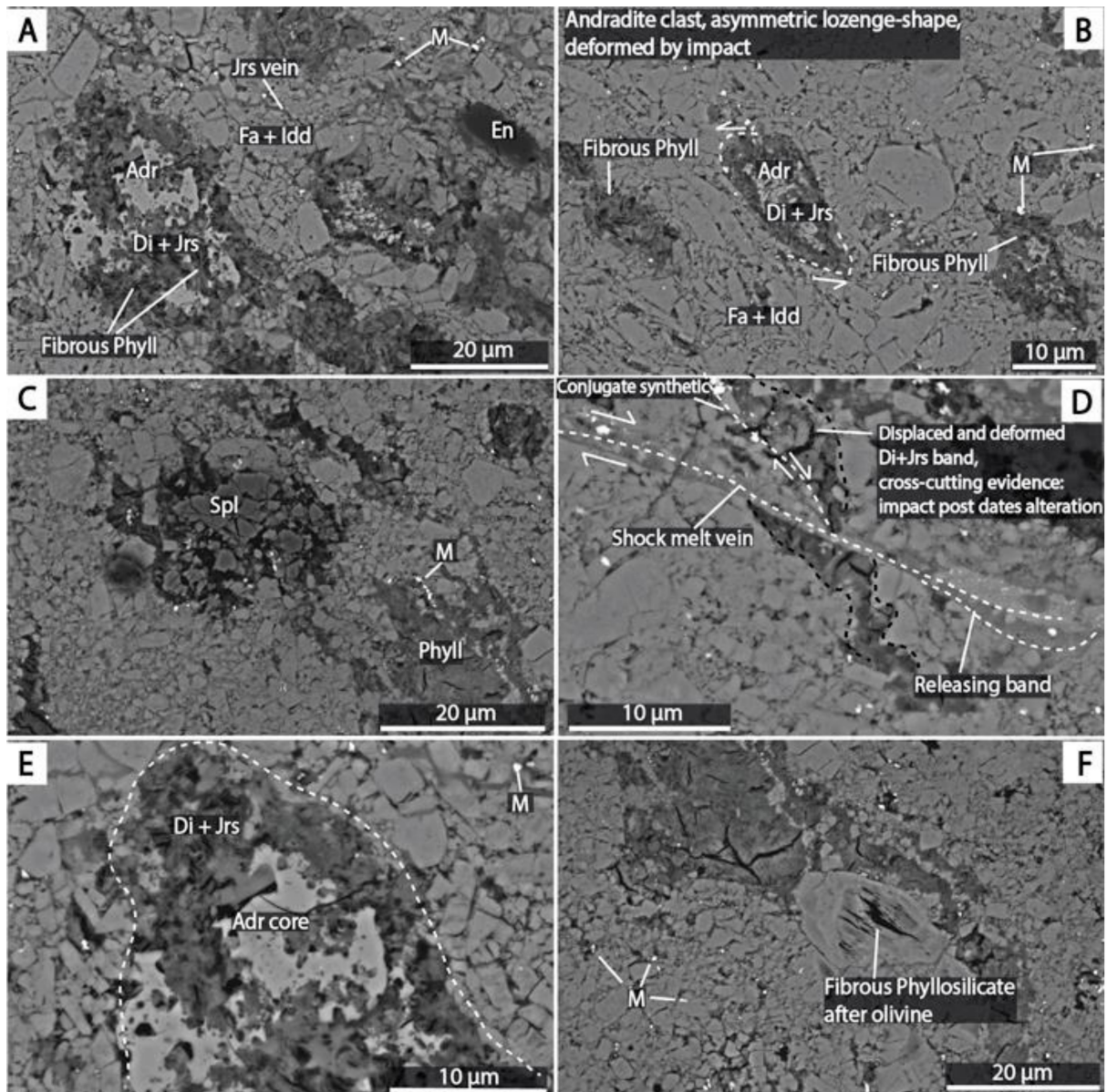
842 Table 1. Bulk composition obtained from 122 random EMPA analyses (oxide wt%) of TAM5.29  
 843 micrometeorite compared with bulk composition of carbonaceous chondrites and average  
 844 composition of unmelted fine-grained Antarctic micrometeorites (UMMs).

	TAM5.29		CV*	CM*	CO*	CI*	CK*	CR*	UMMs**	CV matrix
	Mean	$\sigma$	Mean	Mean	Mean	Mean	Mean	Mean	Mean	Mean
Na <sub>2</sub> O	0.26	0.18	0.46	0.48	0.47	1.13	0.42	0.31	0.44	0.46
MgO	12.6	3.23	23.3	20.9	23.9	19.1	24.4	17.4	16.8	18.8
Al <sub>2</sub> O <sub>3</sub>	4.93	8.34	4.02	2.55	2.98	2.02	2.98	2.31	4.28	2.49
SiO <sub>2</sub>	29.2	4.95	33.2	30.1	33.9	27.3	34.2	31.9	33.9	28.6
P <sub>2</sub> O <sub>5</sub>	n.d.	-	0.28	0.27	0.25	0.28	0.25	0.29	0.40	n.d.
SO <sub>3</sub>	2.17	2.22	3.96	8.6	2.34	17.9	2.9	2.7	1.9	3.29
K <sub>2</sub> O	0.28	0.65	0.04	0.06	0.06	0.11	0.04	0.04	0.13	0.02
CaO	1.41	2.58	3.50	1.96	2.26	1.71	2.39	1.91	0.72	0.85
TiO <sub>2</sub>	0.12	0.27	0.24	0.13	0.13	0.10	0.18	0.10	0.11	0.07
Cr <sub>2</sub> O <sub>3</sub>	0.53	0.42	0.50	0.47	0.52	0.42	0.53	0.53	0.49	0.36
MnO	0.15	0.05	0.19	0.24	0.23	0.28	0.18	0.22	0.23	0.23
FeO	40.1	6.61	28.9	29.5	31.3	28.5	30.3	30.4	29.6	37.6
NiO	n.d.	-	1.37	1.19	1.66	1.13	1.44	1.72	0.37	1.65
TOT.	91.8		100	96.5	99.9	100	100.2	89.9	89.4	94.5

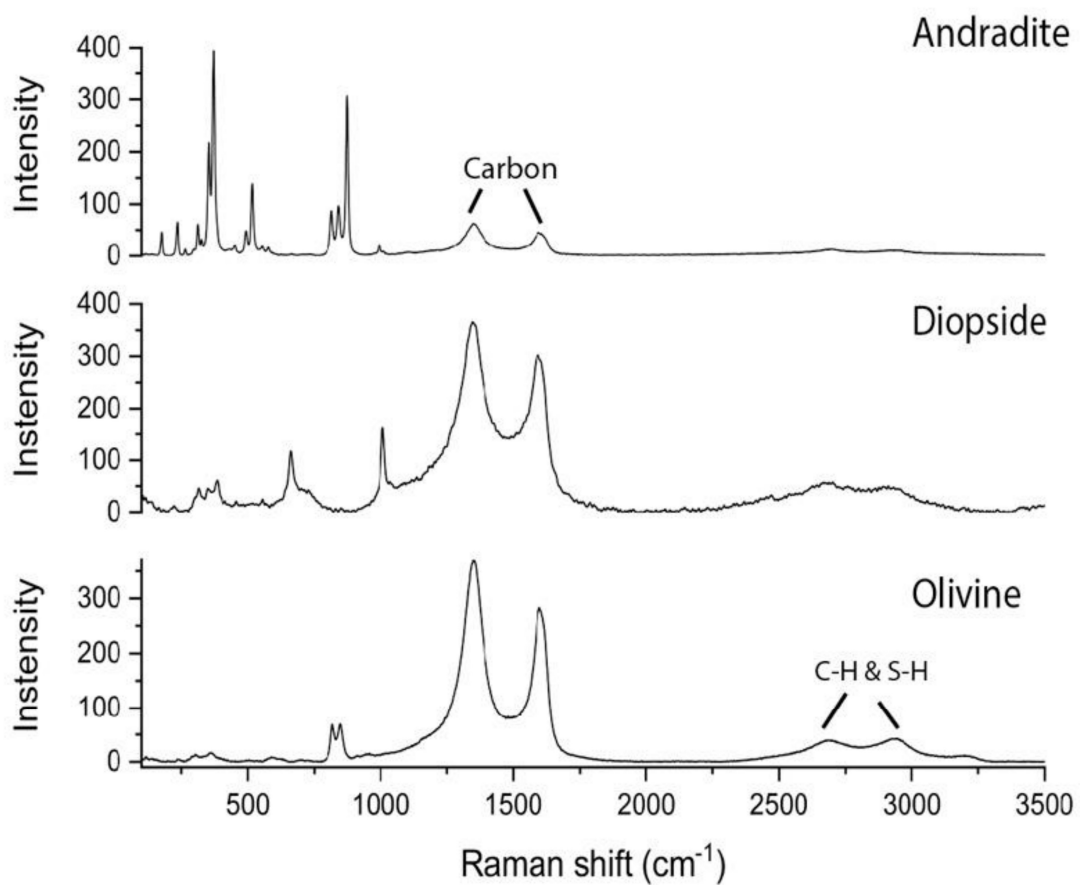
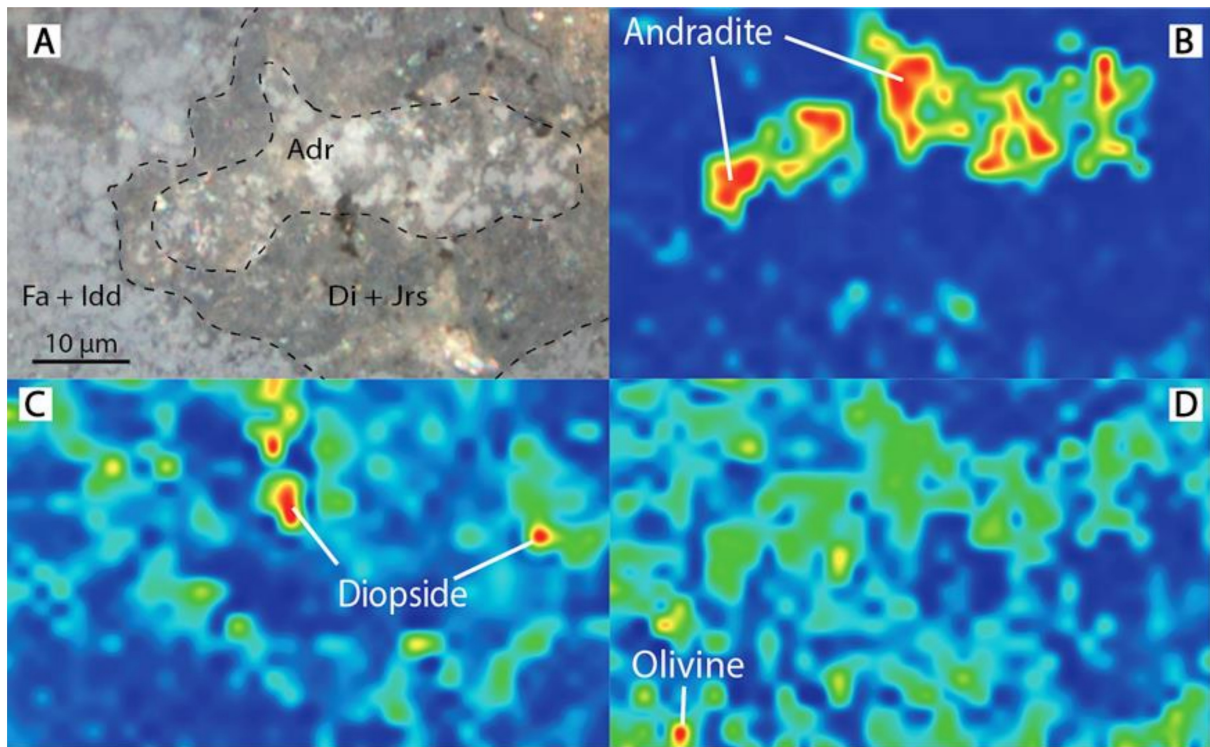
845 n.d.: Not Determined

846 \* from MetBase.

847 \*\* from [Genge et al., 1997](#).

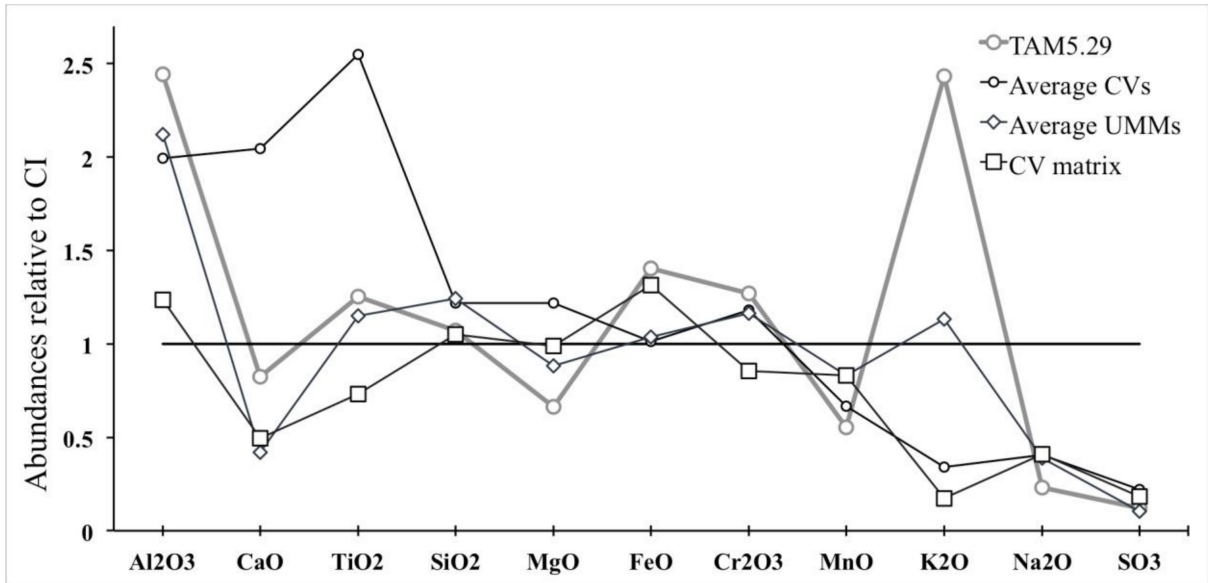


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 849 Fig.1. TAM5.29 FE-SEM BSE images. A-B-E) Typical mineral association of TAM5.29 with andradite surround by  
 850 diopside, jarosite and fibrous phyllosilicates in a matrix made of iron rich olivine altered in iddingsite. Many  
 851 metal grains can be seen. C) Isolated spinel crystals. D) shock melt vein that sharply cuts other crystals. F)  
 852 Fibrous phyllosilicates. Fa=fayalite, Di=diopside, Adr=andradite, Jrs=jarosite, En=enstatite, Phyll=phyllosilicate,  
 853 M=metal alloys-oxides, Spl=spinel, Idd=iddingsite.

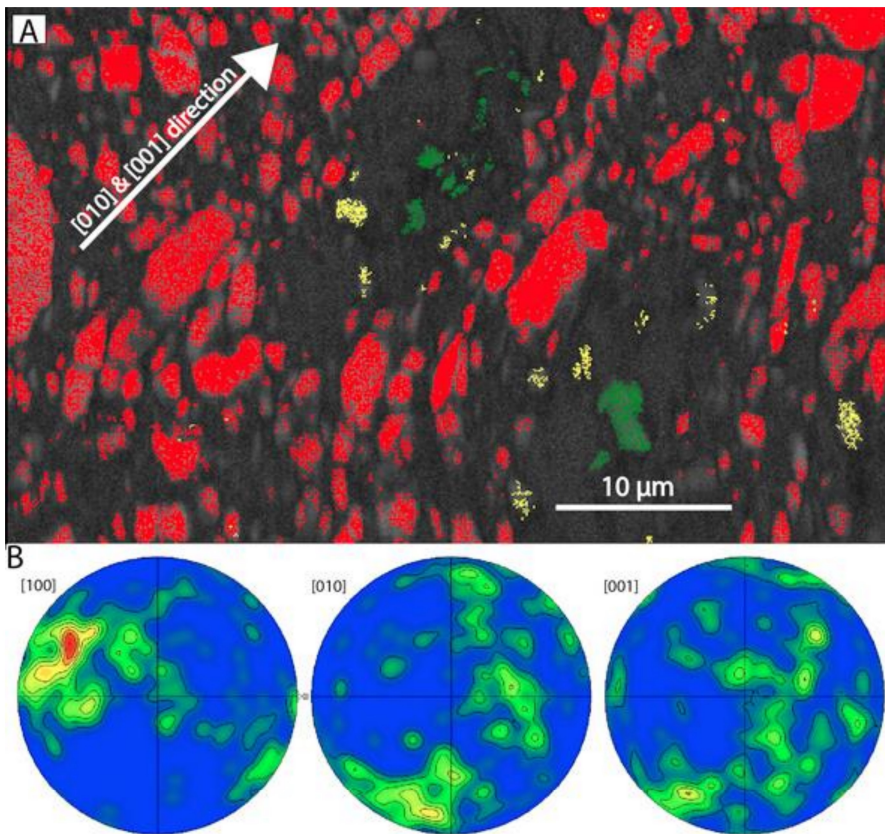


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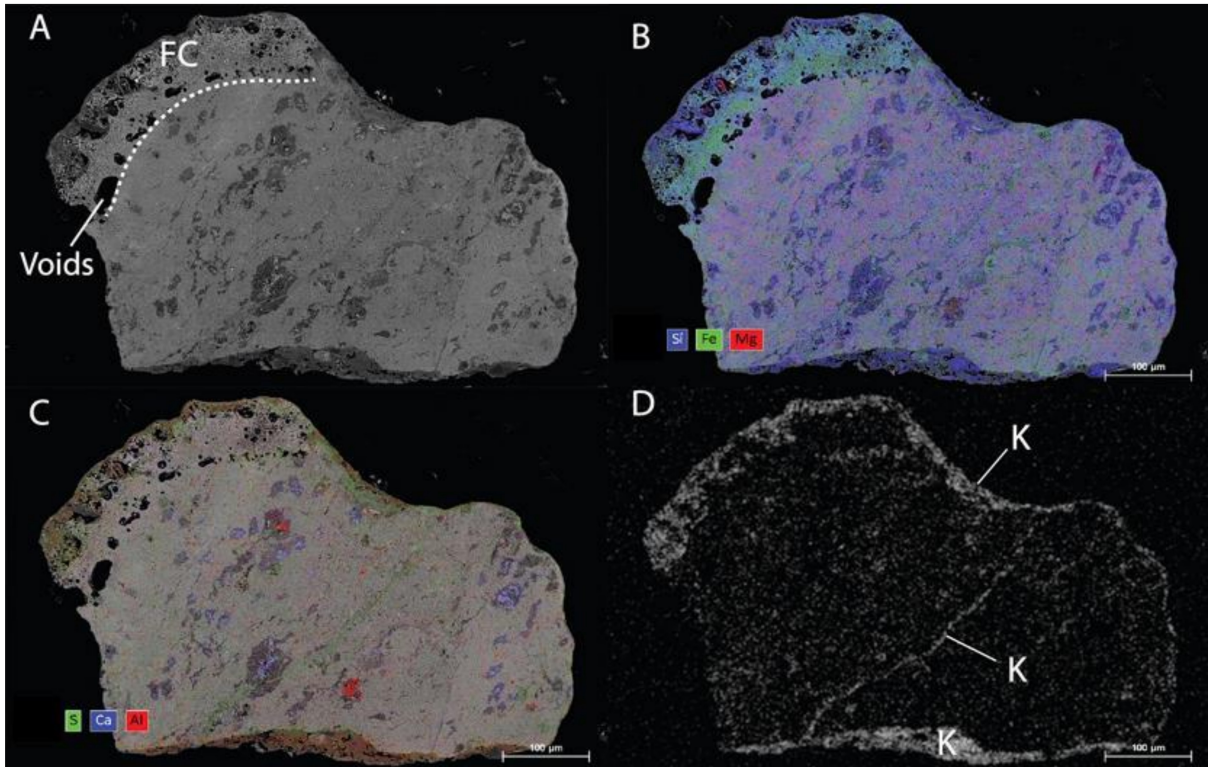
Fig. 2. Raman map showing a typical assemblage of TAM5.29 (A) consisting of andradite inclusions (B) surrounded by dark halos composed of intermixed pyroxene and jarosite (C), immersed in an olivine matrix (D). Abbreviations as in figure 1.



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 859 Fig. 3. Bulk composition of TAM5.29, average CVs (data from METDB database), average CV matrix and  
 860 average Unmelted fine-grained Antarctic micrometeorites (UMMs) (data from [Genge et al., 1997](#)) relative to  
 861 the CI chondrites composition.

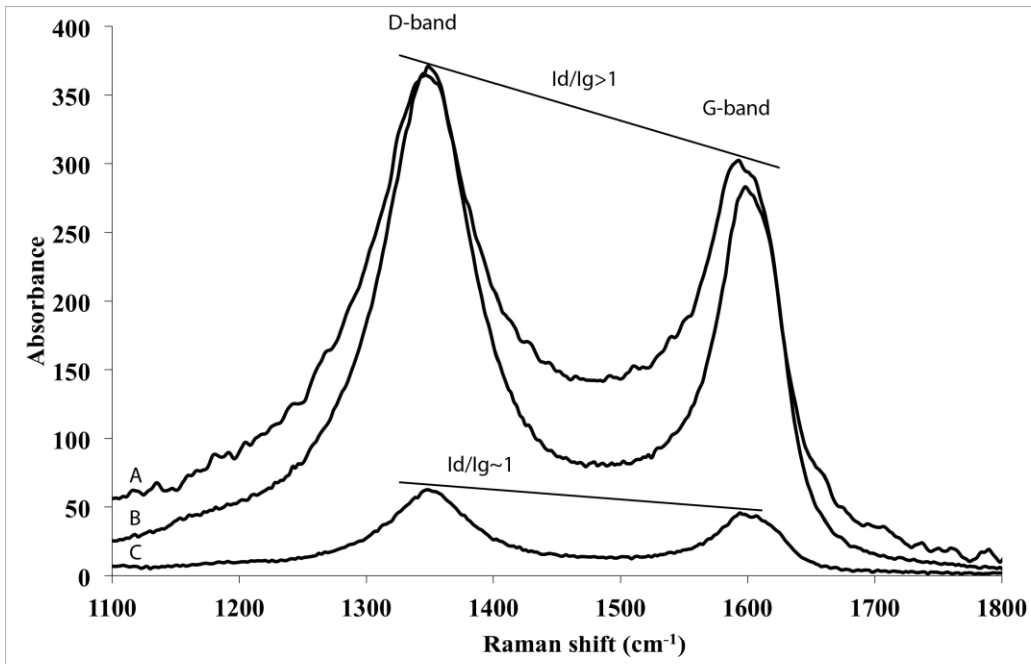


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 863 Fig. 4. A) EBSD phase map of the same area in fig. 1B with relative pole figures of olivine (panel B, one point  
 864 per grain). Red=olivine, green=andradite, yellow=diopside and black-grey area are fine grained undetected  
 865 material.



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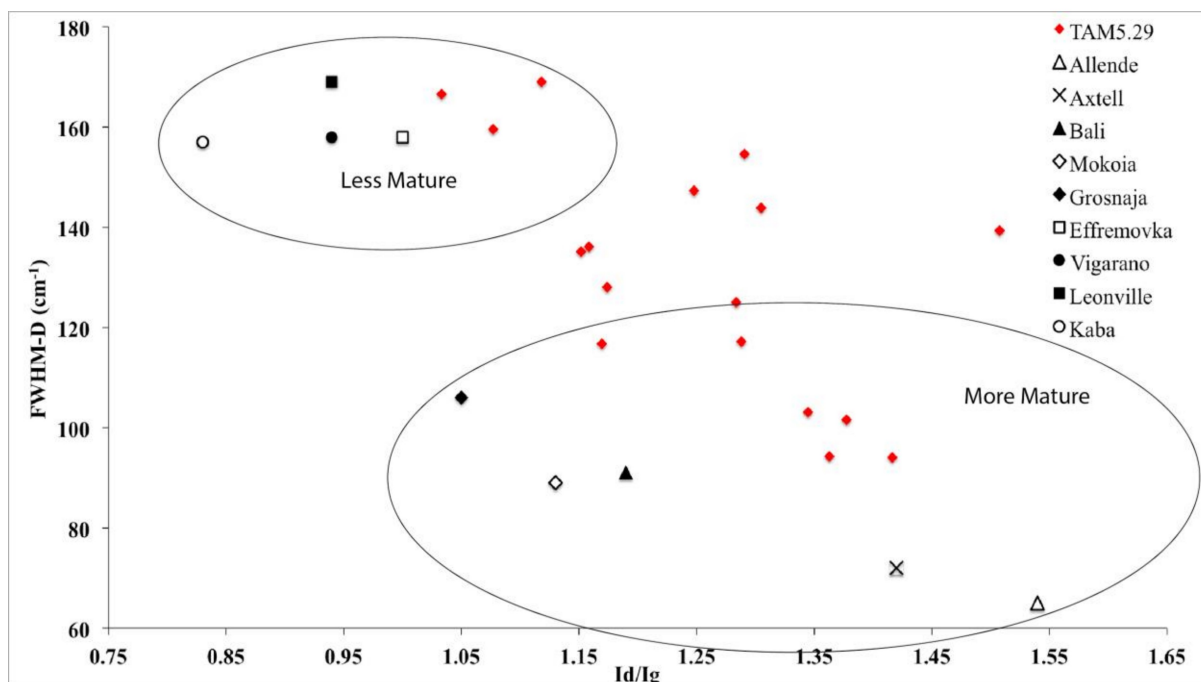
Fig. 5. EDX map of TAM5.29. A) Entire TAM5.29 particle. FC= Fusion Crust. B) Composite map showing Si-Fe-Mg abundances C) Composite map showing S-Ca-Al abundances. D) Map of K concentrations.



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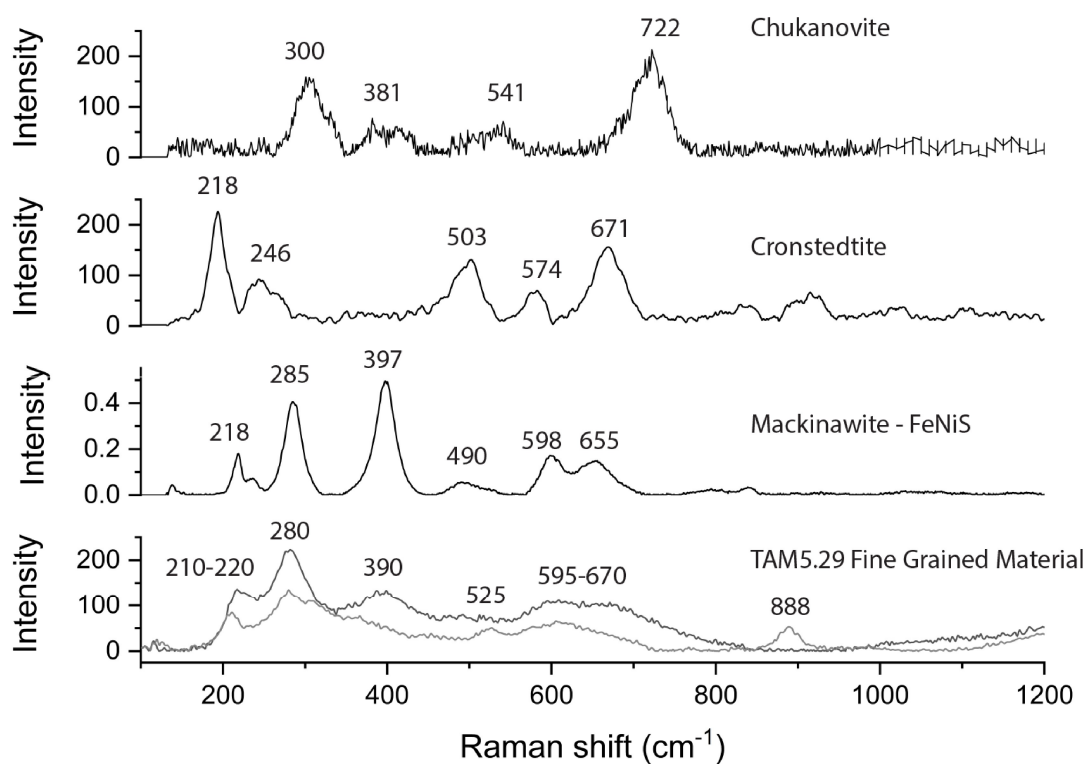
Fig. 6. D and G-bands of some representative analyses of TAM5.29. In case A and B D-band is higher than G-band giving a  $I_d/I_g$  ratio  $> 1$ , indicating high carbon maturity and high metamorphic grade. In case C  $I_d/I_g$  ratio is  $\sim 1$  related to a medium metamorphic grade.



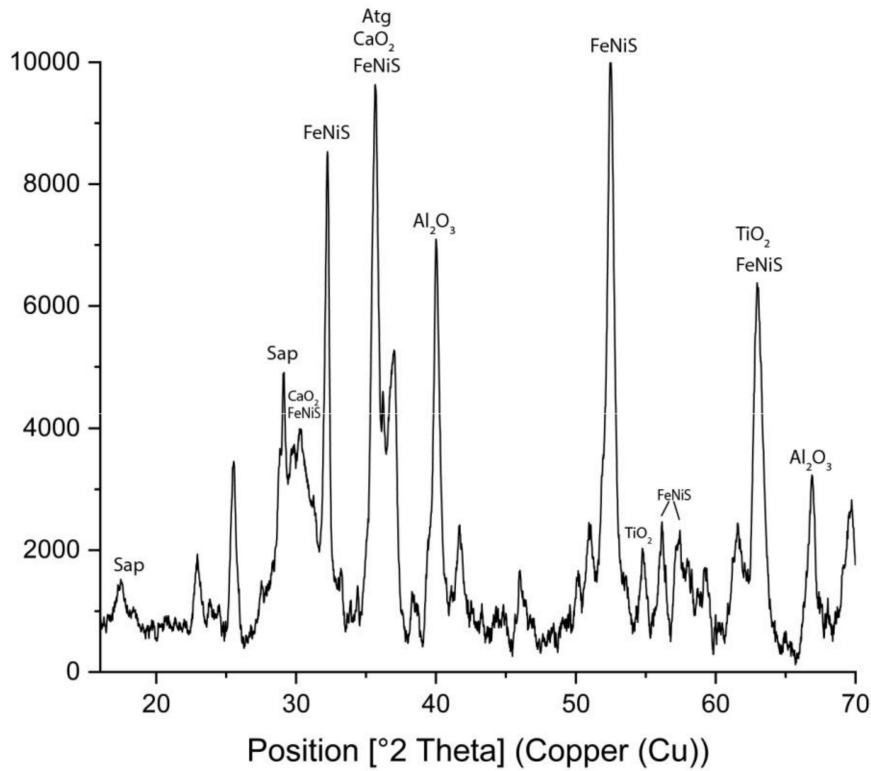


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 874 Fig. 7. Modified from Bonal et al. (2006). Id/Ig ratios are plotted versus FWHM of the D-band. The two groups  
 875 represent more metamorphosed (more mature) and less metamorphosed (less mature) CVs. Red dots  
 876 represent TAM5.29 spectral parameters.

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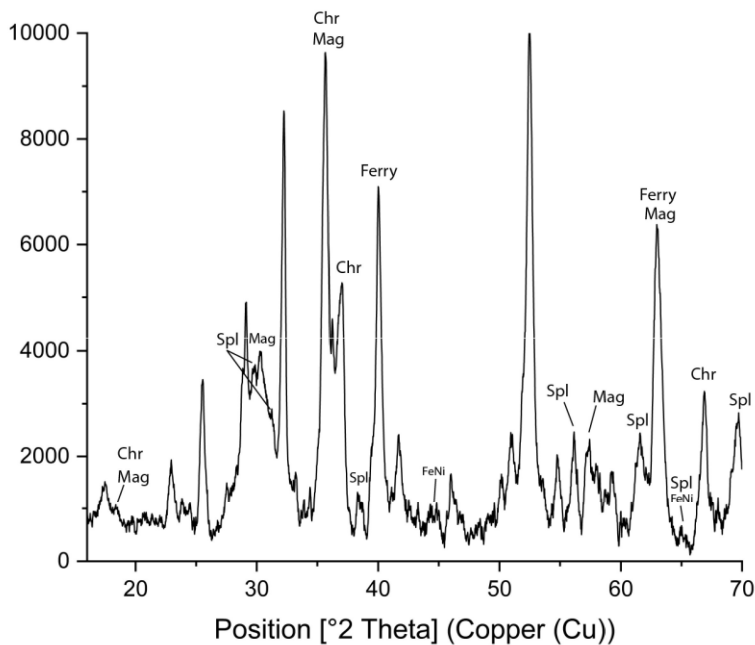


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 879 Fig. 8. Raman spectra of TAM5.29 fine-grained material compared to possible minerals that are contained  
 880 within it.



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Fig. 9. μXRD analysis of TAM5.29. Sap: saponite; Atg: antigorite; FeNiS: iron-nickel sulfides (millerite and mackinawite); CaO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub>-TiO<sub>2</sub>: oxides from CAIs that survived alteration. Sap=saponite, Atg=antigorite.

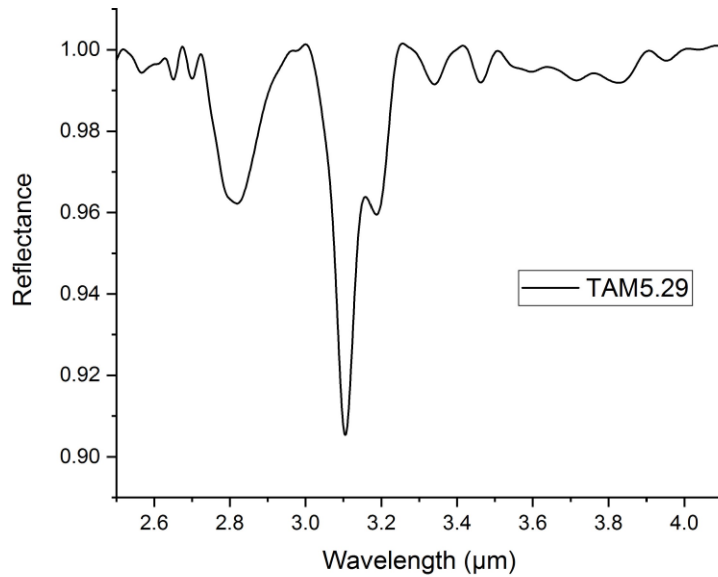


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Fig. 10. μXRD analysis of TAM5.29. Mg: magnetite; Sp: spinel; FeNi: kamacite; Chr: chromite; Ferry: ferrihydrite. Micron sized metal-oxide grains in TAM5.29, in agreement with EDS data in table S2, are mainly magnetite, spinel, chromite and hydrated Fe-oxides such as ferrihydrite. Minor and rare FeNi alloys may be also present. The fact that TAM5.29 mainly contains metal-oxides is also in agreement with μXRD data in fig. 9

891 where  $\text{CaO}_2\text{-Al}_2\text{O}_3\text{-TiO}_2$  are found. Furthermore this confirms the affinity of TAM5.29 with the CVox group.  
892 Mag=magnetite, Spl=spinel, Ferry=ferrihydrite, FeNi=metal alloys.

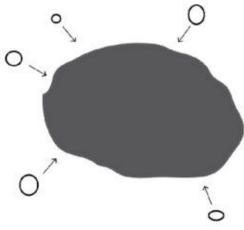
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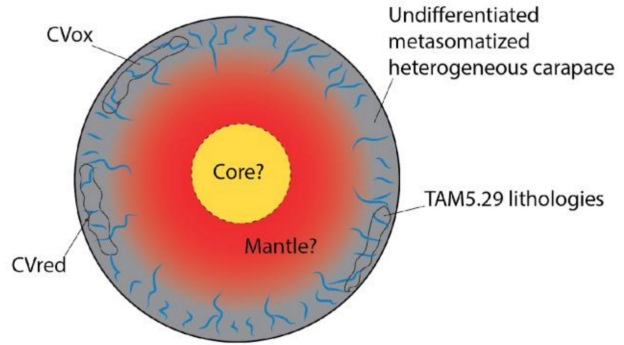
894 Fig. 11. IR reflectance spectra of TAM5.29. To be noticed the 2.8 μm band of Fe hydrous phyllosilicates, 3.3-3.4  
895 μm band given by organic matter (i.e. CH compounds, aromatic and aliphatic hydrocarbons), a broad band  
896 between 3.6 and 3.8 μm likely related to SH functional groups and the 3.9 μm band of carbonates (likely Fe-  
897 carbonates). The 3.1 μm band is a very discussed band and attribution to a mineral phase is difficult, its nature  
898 will be discussed in a dedicated paper.  
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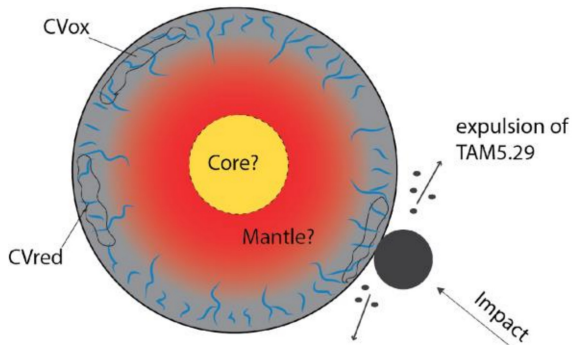
1) Accretion of CV parent Body



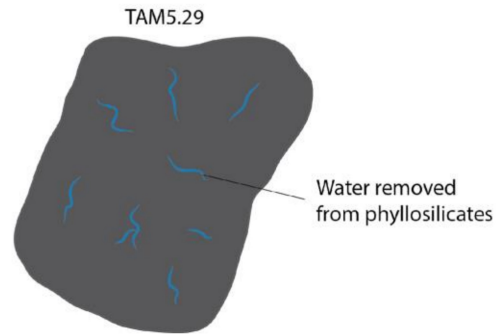
2) Differentiation and metasomatism



3) Impact and expulsion of TAM5.29-like lithologies - creation of olivine petrofabrics



4) Formation of iddingsite after removal of water from phyllosilicates due to impact



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Fig. 12. Sketch representing stages of formation of TAM5.29. 1) Accretion of the CV parent body. 2) Differentiation into a core, mantle and an undifferentiated chondritic crust. The outer layer, or carapace, is heterogeneous in composition and suffer metasomatic alteration, hydrothermal activity and impact gardening. Different areas with different composition and environment cause the formation of the different types of CVs (CV<sub>ox</sub>, CV<sub>red</sub> and TAM5.29-like lithologies). 3) An impact occurs in the area where TAM5.29-like lithologies are found and TAM5.29 is expelled from the CV parent body. In this stage preferred orientation of olivine is created and the thermal metamorphism is terminated. 4) TAM5.29 is already separated from the parent body and water removed from hydrated minerals creates iddingsite at low temperature.

929 **Supplementary Material**

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931 Table S1. Oxide (wt%) EDS analyses of olivine, pyroxene, andradite and spinel of the TAM5.29  
932 micrometeorite.

	Olivine		Pyroxene		Andradite		Spinel	
	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$
Na <sub>2</sub> O	0.11	0.08	0.49	0.33	0.12	0.11	0.23	0.10
MgO	15.2	2.50	21.9	15.1	0.76	0.77	17.1	1.54
Al <sub>2</sub> O <sub>3</sub>	1.27	1.46	3.27	3.00	0.65	0.54	65.2	1.65
SiO <sub>2</sub>	27.6	2.70	52.7	5.07	33.6	1.61	0.97	0.85
K <sub>2</sub> O	n.d.	0.96	0.44	0.58	0.05	0.05	0.03	0.02
CaO	0.35	0.09	7.55	9.15	28.3	9.06	0.16	0.19
TiO <sub>2</sub>	0.07	0.14	0.20	0.10	0.02	0.06	0.15	0.04
Cr <sub>2</sub> O <sub>3</sub>	0.13	0.13	0.75	0.68	0.04	0.06	0.28	0.04
MnO	0.11	2.78	0.15	-	4.58	11.6	0.09	0.06
FeO	44.3	0.12	12.8	11.8	27.0	1.81	15.6	2.06
NiO	0.19	-	n.d		n.d		0.06	0.06
Tot	89.3*		100.3		95.11*		99.82	
Fo/En	37.8	4.85	59.7	33.8				
Fa/Fs	62.0	4.85	24.0	24.6				
Wo			16.3	18.2				

933 n.d.: Not Determined

934 \* Low total due to contamination of near-by hydrous minerals and inclusions of organic  
935 matter.

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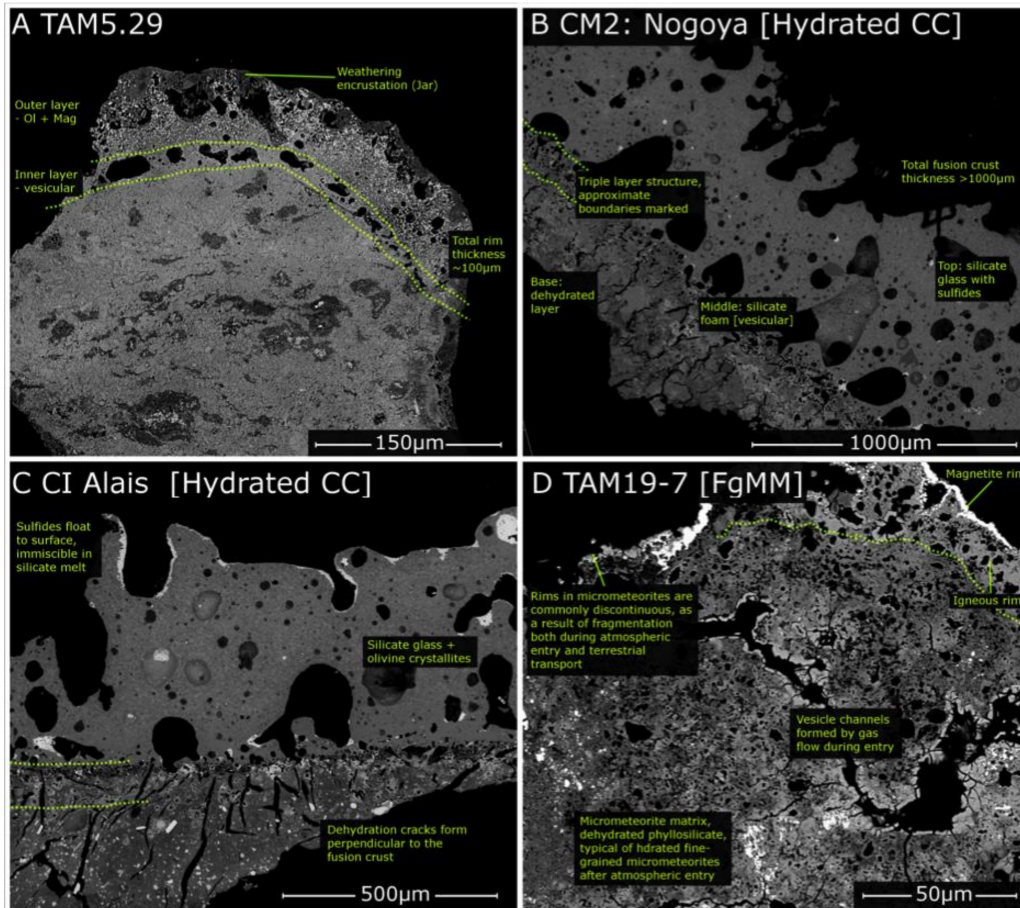
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961 Table S2. Oxide wt% EDS analyses of metal oxides of TAM5.29. Due to the small size of the  
 962 crystals (1-2  $\mu\text{m}$ ) analyses are contaminated by near-by phases giving high abundances of  $\text{SiO}_2$   
 963 and S and low totals due to OH and C contamination. All analyses have high concentrations of  
 964 Fe with the exception of Metal OX\_5 that have high Cr instead. In all the analyses are also  
 965 present minor quantities of Mg, Al and in Metal OX\_5 also Ti. Thus, the most likely phases are:  
 966 magnetite, chromite, spinel, ulvospinel etc.

	Metal OX_1	Metal OX_2	Metal OX_3	Metal OX_4	Metal OX_5
$\text{Na}_2\text{O}$	0.25	0.01	0.08	0.23	0.53
$\text{MgO}$	2.54	5.63	5.74	3.81	11.3
$\text{Al}_2\text{O}_3$	4.46	12.9	3.26	5.95	12.2
$\text{SiO}_2$	4.89	10.5	10.9	9.03	22.9
$\text{P}_2\text{O}_5$	n.d.	0.24	0.19	n.d.	0.72
S	0.32	2.83	1.46	1.59	7.92
$\text{K}_2\text{O}$	n.d.	0.24	n.d.	0.21	0.89
$\text{CaO}$	n.d.	0.20	0.14	0.02	0.65
$\text{TiO}_2$	n.d.	0.34	0.42	n.d.	2.50
$\text{Cr}_2\text{O}_3$	n.d.	0.76	0.30	0.21	24.5
$\text{MnO}$	n.d.	0.04	0.17	n.d.	n.d.
$\text{FeO}$	77.9	57.6	68.4	71.5	17.6
$\text{NiO}$	n.d.	0.11	0.01	0.01	n.d.
Tot	91.4	91.4	91.1	92.5	101.7

967 n.d.: Not Detected.

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Fig.S1. The melt layer on TAM5.29 developed during atmospheric entry and comparison against fusion crusts and igneous rims on micrometeorites.