- Sustainability of using vineyard pruning residues as an energy source:
 combustion performances and environmental impact
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18 Abstract

19 Open burning of agricultural waste is still a common practice as it is a rapid method for waste 20 disposal, although natural biomass, including agriculture residues, can be exploited as a renewable 21 energy source. We assessed the viability and sustainability of using vineyard pruning residues, as 22 wood chips, for energy conversion. Wood chips, obtained from vineyards in the Prosecco DOCG region (Italy), were characterized in terms of chemical composition, calorific value, ash content and 23 24 humidity. Combustion tests were performed in a medium-size biomass boiler (maximum power 500 25 kW) to assess the viability of the approach in terms of sustainable steady-state combustion. Primary 26 emissions of both macro- and micro-pollutants were measured to assess the environmental impact. An analytical method was purposely developed for the determination of polycyclic aromatic 27 28 hydrocarbons (PAHs), polychlorinated biphenyls (PCBs), polychlorinated dibenzo-p-dioxins (PCDDs) and polychlorinated dibenzofurans (PCDFs) in three matrices: fly ashes, condensate and 29 30 gas. The results showed that vineyard pruning residues can be used for energy conversion in

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31 medium- and large-scale biomass boilers. Primary emissions of measured pollutants were all below 32 limit values set by current European legislation except for particulate matter, for which current 33 available abatement technologies are required to contain emissions, thus making the use of vineyard pruning residues unsuitable for combustion in domestic appliances where such technologies are not 34 35 installed. Bottom ashes produced during combustion were also characterized to assess whether they 36 can be recycled in the vineyard as soil amendments/fertilizers. Copper content in combustion ashes 37 exceeded limit values for ashes to be used as fertilizers in agricultural fields for some European 38 countries but not for others, indicating that ashes may need to be disposed as waste.

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40 Keywords

41 Waste-derived fuel, Biomass, Vineyard residues, Wood boiler, Emissions, Persistent organic
42 pollutants (POPs)

43 **1 Introduction**

The most recent scenarios depict a rise in world energy demand by 30% between today and 2040, which is equivalent of adding another China and India to today's global demand [1]. It is therefore imperative to shift towards renewable energy sources and increase energy efficiency in order to limit global temperature rise and meet the goals set by the Paris climate agreement [2].

48 Natural biomass, including agriculture residues, is a renewable energy source [3], although this 49 definition does not consider the very different timescales when comparing carbon emissions during 50 burning and resequestration during a plant's lifetime [4]. Currently, a significant part of the energy 51 from agricultural by-products and wastes is lost: open burning is still a common practice as it is a rapid method for waste disposal [5–7] but with considerable detrimental impacts on air quality 52 53 [8,9]. Besides the underestimated risk of fire, this practice not only causes uncontrolled release of greenhouse gasses without recovering any energy, but it can be also the cause of uncontrolled 54 emissions of particulate matter (PM) and of persistent organic pollutants (POPs) in the 55 56 environment. polychlorinated dibenzo-p-dioxins and furans (PCDD/Fs), such as 57 polychlorobiphenyls (PCBs) and polycyclic aromatic hydrocarbons (PAHs) [7,10,11]. These 58 compounds are a cause of concern for human health and for the environment. Once emitted, these 59 POPs tend to accumulate in fatty tissues and in the food chain. They are also very persistent and 60 therefore subject to long-range atmospheric transport [12]. In fact, they have been detected even in 61 the most pristine environments on Earth [13].

Modern air quality regulations aim at reducing both intentional and unintentional release of micropollutants in the environment. In this connection, open burning of agricultural residues has been strictly limited or prohibited by local authorities (for instance in Italy since 2010 with the Legislative Decree D.Lgs. 205/2010 [14]). Otherwise, the controlled burning of waste-derived biomasses for energy use can be a viable and safer alternative [15] to uncontrolled burning in the field. Anyway, the sustainability of this resource requires that other environmental aspects are correctly addressed, in particular those regarding the management (collection, transport and processing) of the biomass to be used as fuel, the atmospheric emissions during controlled burning,and ash disposal.

In many countries, vineyard pruning residues (VPRs) represent a good example of agricultural waste that can become fuel. Recent studies have investigated the possibility to exploit this available biomass as an energy source, through the characterization of the biomass [16,17] and its use both as wood chips [18] and as pellets [19,20] in domestic appliances. However, due to the high ash content and humidity, maintaining an efficient steady-state combustion in domestic appliances has proven to be challenging [18,19] and the use of larger scale biomass boilers has been suggested as a better alternative [21].

78 In a densely inhabited area, the presence of large-scale thermal and power plants can be not well 79 accepted by local residents. This highlights a contradictory opinion, on one side biomass thermal or 80 power plants are considered fundamental to increase the share of renewable energy in the country, 81 on the other side large size plants, especially when low quality biofuel is used, face opposition by local residents [22]. As a consequence, the type, location and operation of the facilities play a 82 83 fundamental role on the social acceptance [23]. Besides this, the northern Italian Po valley 84 represents a European hot-spot concerning aerosol pollution [24], where local authorities often limit 85 the use of wood burning appliances in winter [25]. It follows the need to evaluate the environmental 86 drawbacks of using biofuels derived from poor quality residual biomass, such as VPRs, in medium-87 size boiler.

In this study, for the first time, we investigate the possibility of exploiting VPRs, as wood chips, for energy conversion in a medium-size biomass boiler (maximum power 500 kW) taking into account also the related environmental drawbacks. The study aims at assessing the viability and the sustainability of the steady-state combustion of VPRs, as wood chips, in a medium-size biomass boiler. The overall environmental impact of the combustion process is here evaluated in terms of quality of the VPR wood chips as fuel, primary emissions of both macro- and micro-pollutants and ashes produced from the combustion process.

95 2 Materials and methods

96 2.1 Characterization of the VPRs

97 The VPR wood chips were collected in a specific geographical area in northern Italy, the Prosecco 98 DOCG region, a hilly countryside area at the margin of the Po valley characterized by a significant 99 high density of vineyards. In the Prosecco DOCG district (6650 ha) the estimated production of 100 VPRs is of 25000 t/y. The sample used in this study was obtained by chipping VPRs using a 101 purpose-build shredder machine, the Cobra Collina[®] of Peruzzo S.r.l. (Curtarolo, Padova, Italy). 102 The main advantage in the use of purpose-build shredder machine lies in the versatility of the 103 machine and in the logistics of storage and handling that is better suited to the vineries in the area 104 even for limited storage spaces (the chipped VPRs occupy less space than the densified material 105 balers).

106 After harvesting, the wood chips were transported and stored in the proximity of the farm in 107 covered piles for a period of 7 months. Covering the VPRs with an impermeable and breathable 108 sheet has the advantage of accelerating the seasoning and preserving the quality of the material. 109 Furthermore, with heaps with a contained height and a wide base, the preservation of the material is 110 further encouraged. At the end of this period, the wood chips were transported in big-bags (volume 111 of 1000 liters each) to the experimental biomass boiler site. Once stored at the experimental site, 112 three samples of 50 liters of wood chips were collected and transported to the biofuel laboratory to 113 be characterized in terms of physico-mechanical properties by the determination of moisture 114 content, ash content, high heating value and particle size [19]. The three samples refer to the initial 115 characterization done after 7-month seasoning in a covered pile, just before the first combustion 116 test. While additional characterization tests over time were not done, it is expected that physicalmechanical characteristics of the fuel had not changed significantly afterwards. The values of the 117 118 aforementioned parameters were assessed according with the recent in force ISO standards for solid 119 biofuel quality assessment [26]. The moisture content of the sample was measured following the

120 method described in the EN ISO 18134-1 [27]. Three samples were placed in a ventilated oven at 121 105±2°C for up to 24 hours until constant weight was attained. The ash content was measured according to the EN ISO 18122 [28], using three samples of grounded chips of minimum 1 g each. 122 123 The samples were weighed before and after their complete combustion that was carried out 124 according to the temperature program stated in the EN ISO 18122 [28]. The reported ash contents 125 are the average of the three repeats on a wet basis (as received). To measure the high heating value of the samples as received (HHVar) an IKA® C-200 calorimeter was used (three repeats; IKA®-126 127 Werke GmbH & Co. KG). The high heating value on a dry basis and ash free (HHVdaf) is 128 calculated as indicated in the standard EN ISO 18125 [29]. To measure the particle size distribution, 129 a volume of 8 liters of the wood chips have been sieved at moisture content below 20 w-% on a wet basis to prevent the fine particles from sticking together or losing moisture during the sieving 130 131 process. Five circular sieves (metal plates) with round perforated holes from 3.15 to 63 mm, in 132 accordance with ISO 3310-2 [30], have been used for particle size determination following the ISO 133 17827-1 [31]. Since in the general requirement for solid biofuel EN ISO 17225-1:2014 [32] the 134 VPRs are referred to as woody biomass and identified with origin number 1.1.7 "Segregated wood 135 from gardens, parks, roadside maintenance, vineyards, fruit orchards and driftwood from freshwater", the determination of particle size classes has been made in accordance with EN ISO 136 137 17225-4:2014 [33], where the origin 1.1.7 is a subset of the 1.1 "Forest, plantation and other virgin wood", and it refers to a B quality class. 138

139 **2.2** Combustion tests and measurements of gas and particulate emissions

Three combustion tests and measurement campaigns were done on the 9th August 2012, 21st December 2012 and 24th April 2013 for a duration of about 10 hours each. The combustion tests of the VPRs were carried out in a medium-size experimental biomass boiler (Uniconfort S.r.l., San Martino di Lupari, Padova, Italy), with a maximum power of 500 kW (Figure 1). The boiler is equipped with an automatic feeding system which slowly brings the wood chips to the furnace by means of a movable grate. After the furnace, exhaust gases are transported to the post-combustion chamber where they enter into contact with the heat-exchange unit. Combustion parameters (*e.g.*, furnace temperature, post-combustion chamber temperature, flue gas temperature, % O_2 in the flue gas) are monitored and self-adjusted in continuous to ensure maximum efficiency (O_2 11%). Combustion ashes are removed continuously from the grate through an automatic extractor.





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Figure 1. Picture and schematic representation of the wood boiler.

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During the campaigns, continuous measurements of the concentrations of O_2 , CO_2 , CO_2 , CO_3 , NO_x , SO_2 , C₃H₈, H₂S in the exhaust, together with combustion efficiency, combustion air temperature, and exhaust air temperature, were done using a MRU[®] VARIOplus (MRU Instruments, Inc.) at about 2 m down the flue from the heat-exchange unit (fume temperature 160-200°C).

Total suspended particles (TSP) were measured following the method EN 13284-1 [34], at a height of about 6 m (duct diameter 35 cm, fume temperature about 160-200°C). Isokinetic conditions (flue gas velocity of 3-5 m s⁻¹) were maintained throughout the sampling (~5 minutes, 1-2 repetitions per experiment) by using nozzles with different internal diameters, monitoring the exhaust gas velocity in the duct and by manually adjusting the aspiration flow from the pump (ZB1 timer, Zambelli S.r.l., Milan, Italy). Filters used for TSP collection were made of quartz fiber (QFF, AQFA type, Merck Millipore KGaA), and they were pre-treated at 400°C for 4 hours. Weighing of the filters was done, before and after sampling, after conditioning at a temperature of 20±1°C and relative humidity of 50±5% for at least 48 hours, as in previous studies [35]. The effect of weighing filters at 50% relative humidity rather than dry conditions is negligible in our experimental conditions because the hygroscopic growth factors at 50% relative humidity during hydration are small even for very hygroscopic particles like NaCl [36]. Flue gas velocity and volumetric flow were measured using the method UNI 10169 [37] in-between each sampling.

170 2.3 Determination of PAHs, PCBs and PCDD/Fs

171 2.3.1 Sampling

172 The sampling was carried out at a height of about 6 m (duct diameter 35 cm, fume temperature 160-173 200°C) in isokinetic regime using the "filter/condenser method" EN 1948-1 [38]. The sampler is 174 composed of a glass probe inserted in a thermostated tube (120°C), a thermostated box (120°C) 175 containing the filter holder fitted with a quartz fiber filter (QFF, AQFA type, Merck Millipore KGaA), a glass condenser maintained at 5°C using a chiller, a resin cartridge (~10 g of Amberlite[®] 176 XAD[®]-2, Supelco[®] or XAD[®]-2 Ultra-Clean Resin, Restek Co.) at room temperature, and a constant 177 178 flow pump. During sampling, gas velocity in the duct was continuously monitored using a Pitot tube. The isokinetic condition (flue gas velocity of 3-5 m s⁻¹) was maintained by using nozzles with 179 180 different internal diameters and by manually regulating the aspiration flow throughout the sampling 181 (~3 hours, 1-2 repetitions per experiment). This system allows the collection of three different 182 fractions: in the first step, the PM was collected on QFFs, then the vapors were condensed and 183 collected in a glass bottle placed under the condenser, and finally the gaseous products were 184 adsorbed in the resin cartridge. The filters used for PM collection were pre-treated at 400°C for 4 hours and spiked with $4x100 \,\mu\text{L}$ of 4 ¹³C-labelled PCB standards at 20 pg μL^{-1} (PCB SYRINGE 185 SPIKE, PCB-70/111/138/170, 1000 ng mL⁻¹, CIL – Cambridge Isotope Laboratories, Inc., diluted 186 2:100 in iso-octane, >99.3%, ROMIL-SpSTM) before sampling. 187

188 *2.3.2 Extraction*

189 Three different extraction procedures for the three different matrices (fly ash on the QFFs, condensate, gas adsorbed on the resins) were used, following the indications of the method EN 190 191 1948-2 [39]. Nevertheless, in order to enhance repeatability and reduce contamination risks, 192 automated instruments were used for analytes extraction, while toluene was replaced with low-193 boiling and less toxic extraction solvents in order to shorten the concentration step and avoid the 194 loss of the most volatile analytes (e.g., PAHs; see Figure S1 in the Supplementary Material). All 195 solvents used were Super Purity Solvents (SpS[™], Romil Ltd.). All the tools and glassware were washed with an aqueous 2% (v/v) Contrad[®] 2000 solution, dried with acetone and rinsed three 196 197 times with dichloromethane and three times with *n*-hexane.

For greater clarity, the steps of the proposed analytical method are summarized in the flow chart inFigure S2.

200 Prior to extraction, blanks and samples were spiked with a known amount of isotope-labelled 201 solution (3 ¹³C-labelled PAHs at 1 ng μ L⁻¹, 21 ¹³C-labelled PCBs at 40 pg μ L⁻¹ and 15 ¹³C-labelled 202 PCDD/Fs at 10-20 pg μ L⁻¹; CIL Inc.) as internal standards.

203 Concerning QFF samples, PAHs, PCBs and PCDD/Fs were extracted by means of a Pressurized Liquid Extractor (PLETM system, FMS, Fluid Management System Inc.) equipped with stainless 204 205 steel extraction cells, using *n*-hexane/dichloromethane (1:1, v/v). QFFs were broken in small pieces 206 and mixed with anhydrous sodium sulfate, diatomaceous earth and Ottawa sand (Sigma-Aldrich Co., Applied Separation Inc.). The operative conditions were previously optimized for enhanced 207 208 extraction efficiency and to minimize the amount of solvent required [40,41]. The extractions were 209 performed at 100°C and 100 bar, with 3 static cycles of 7 min each. An acid pre-treatment, before 210 extraction, is often done for analysis of organic compounds in fly ashes [42]. The attack with acid is 211 carried out to disrupt the mineral conglomerates that are formed in the combustion process. These 212 conglomerates (solid sphere, cenosphere, solid unshaped carbon particles) are formed at high temperature (1400°C) and are composed of Fe (Fe₂O₂), Al (Al₂O₃), Si (SiO₂), along with Ca, K, and 213

Mg [43–45]. The temperature reached in biomass boilers are much lower (in this work 500-700°C), for these reasons the acid pretreatment was not performed in this study.

216 The condensate was extracted through a liquid-liquid extraction. The samples (about 70-100 mL) 217 were put in a separating funnel with 10 mL of dichloromethane. The funnel was shaken for 15 218 minutes and the organic phase was collected in a glass bottle. The extraction was repeated three 219 times in total. The three aliquots of organic phase were then combined and anhydrified with sodium 220 sulfate (≥99.0%, anhydrous, ACS reagent, Sigma-Aldrich®). The procedural blank was prepared as 221 follow. A fourth aliquot of organic phase was used for extraction from the same sample and disposed. Successively, the sample was spiked again with ¹³C-labelled internal standards and the 222 223 extraction procedure was repeated another three times. The three aliquots of organic phase, constituting the procedural blank, were then combined and anhydrified. 224

The volatile compounds adsorbed onto the XAD[®]-2 resin were extracted using Accelerated Solvent Extraction (ASETM 200, Thermo Scientific, Dionex, Inc.). The resin samples were extracted using *n*-hexane/acetone (1:1, v/v) at a temperature of 100°C and a pressure of 120 bar using the procedure optimized by Yang *et al.* [46]. The extraction procedure, 5 minutes static extraction after a 5 min equilibration time, was repeated three times. After extraction, the cell was rinsed with the extraction solvent and flushed with N₂. The total extraction volume was about 70 mL. The same extraction procedure was also done on blank resin samples.

232 2.3.3 Clean-up

The clean-up procedure is the same for all extracts of the three matrices. The extracts were reduced to a volume of 500 µL under a gentle 5.0 (99.999%) nitrogen flow at 23°C (Turbovap[®] II, Caliper Life Science Inc.), recovered with 5 mL of *n*-hexane, and reduced again to 500 µL in the same conditions. Purification was performed by directly injecting the sample into a disposable neutral silica column in an automated system (Power-PrepTM, FMS Inc.) and by eluting it with 30 mL of *n*hexane and 30 mL of *n*-hexane/dichloromethane (1:1, v/v). The eluates were collected as one single fraction. Purified samples were reduced to 100 μ L under a gentle nitrogen flow at 23 °C and spiked with a known amount of the recovery standard solution (1 ¹³C-labelled PAHs at 1 ng μ L⁻¹, 2 ¹³Clabelled PCBs at 40 pg μ L⁻¹ and 2 ¹³C-labelled PCDD/Fs at 20 pg μ L⁻¹; CIL Inc.) according to the method fully described in Pizzini *et al.* [41,47].

243 2.3.4 Instrumental analysis

According to the method EN 1948-3 [48], analyses were performed by High-Resolution Gas Chromatography coupled both with Low and High-Resolution Mass Spectrometry (HRGC/LRMS and HRGC/HRMS respectively), while quantification was performed using internal standards and the isotopic dilution technique, using perfluorotributylamine (PFTBA; Sigma-Aldrich Co.) as the reference compound. 16 priority PAHs, 127 out of 209 possible PCB congeners and 17 2,3,7,8substituted PCDD/Fs were investigated. All the results were corrected using the instrumental response factor and for the procedural blanks.

Instrumental determination of PAHs and PCBs was carried out using a HRGC/LRMS (Agilent 7890A Series GC System, coupled with an Agilent 5975 inert Mass Selective Detector, Agilent Technologies Inc.) operating in electron impact mode (EI, 70 eV). Acquisition was performed in selected ion monitoring (SIM) mode.

255 PCDDs and PCDFs were determined using a HRGC (Hewlett Packard - Agilent 6890 Series GC 256 System, Agilent Technologies Inc.) coupled with a HRMS (Thermo Finnigan MAT 95XP, Thermo 257 Fisher Scientific Inc.) operating in electron impact mode (EI, 45 eV, with $R \ge 10\ 000$). Acquisition 258 was performed using multiple ion detection (MID). Experiments were carried out monitoring the 259 two most intense peaks of the EI spectra for each analyte.

Both instruments were equipped with a 60 m HP5-MS column (0.25 mm I.D., 0.25 μm film
thickness, Agilent Technologies Inc.).

262 Operating conditions, reference and target ion masses are presented in detail in Tables S1-S6.

263 2.3.5 Data quality for the analysis of organic contaminants

Analytical bias and precision were estimated by analyzing different Standard Reference Materials (SRMs) as similar as possible to the matrices under investigation. The instrumental detection limits (IDLs) were evaluated as three times the standard deviation of the blank signal and ranged between 0.2 and 15 pg for PAHs, 0.2 and 6 pg for PCBs and 0.1 and 0.3 pg for PCDD/Fs (as injected amounts). Since preliminary tests on real samples showed that concentrations of PCDD/Fs were always below IDL, the analytical method for these classes of compounds was not validated further and following discussion refers to PAHs and PCBs only.

271 Initial tests on the QFFs were done on Fly Ash Reference Material EDF-5369 (CIL Inc.), which 272 derives from the filter of a municipal waste incinerator in northern Italy. Nevertheless, this kind of 273 matrix contains high levels of cenospheres, requiring an acid pretreatment that breaks up these 274 conglomerates before the analysis. In order to be more consistent with the analytical protocol used 275 in this work, it was therefore decided to use SRM 1649B (Urban Dust) from NIST (National 276 Institute of Standard and Technology) to check the analytical quality of the method. Indeed, this 277 SRM has a micro- and macroscopic structure more similar to the real samples, even if it does not 278 come from direct combustion sources.

279 About 0.05 g of SRM were deposited on a QFF and analyzed for PAHs and PCBs as previously 280 described. Precision, measured as relative standard deviation (RSD) of three replicated analyses of 281 the same SRM, ranged from 0.2 to 12% for PCBs, with an average value of 2%. As for PAHs, precision for single compounds ranged from 1 to 17% for high molecular weight (4-6 rings) 282 283 compounds, while for low molecular weight (2-3 rings) PAHs, RSD were 33% (with a total average 284 value of 12%). This high variability can be explained by their greater volatility and, hence, 285 sensitivity to the two steps of volume reduction that are required by the pre-analytical procedure. 286 The accuracy tests led to good results, with a relative error (E_R) in the range of 1-23% for PCBs, 287 while more than 60% of the certified analytes were accurate within 2 σ for PAHs.

288 Concerning condensate, since there is no available SRM for this matrix, it was necessary to spike 289 tap water, previously extracted according to the method described, with a known amount of the 290 surrogate standard solution of each class of analytes. About 100 mL of spiked water were analyzed 291 for PAHs and PCBs. Three repetitions were executed. The precision, expressed as repeatability, ranges between 4 and 10% for PCBs, with an average value of 6%. As for the PAHs, RSD ranges 292 293 between 2 and 21%, with an average value of 9%. The results for most of the PCB target 294 compounds were in good agreement with the certified/reference values, with the E_R in the range of 295 1-19%, while for PAHs E_R ranges from 1 to 44%, with the exception of Benzo[ghi]perylene, 296 systematically underestimated.

297 Concerning resin samples, in absence of a SRM to verify accuracy and precision of the analytical 298 procedure, a blank XAD[®] resin was spiked with a known amount of the surrogate standard solution 299 of each class of analytes and analyzed as described. Nevertheless, these kinds of resins contain high 300 background levels of PCBs causing an interference in the analysis. Therefore, the method was not 301 validated for this type of matrix and the concentration values obtained from its analyses are to be 302 considered as indicative values.

303 2.4 Analysis of combustion ashes

304 Combustion ashes were collected at the end of each test and analyzed according to the procedure 305 already described elsewhere [19,49]. Briefly, 0.3 g of ashes were treated with 3 g of nitric acid 306 (69%, PROLABO S.r.l., Milan, Italy) in a CEM Discover SP-D (CEM Corp., Matthews, NC, USA) 307 microwave digester, with a temperature ramp from 20°C to 200°C at 30 bar in 4 min, and 308 maintained at the final conditions for 2 min. After digestion, solutions were diluted in a 50 mL 309 volumetric flask with ultrapure water, produced by a Millipore Plus System (Milan, Italy, resistivity 310 18.2 MΩ cm⁻¹), and analyzed by ICP-MS (Agilent 7700x; Agilent Technologies, Santa Clara, CA, 311 USA).

312 **3 Results and discussion**

313 **3.1 VPR wood chips quality**

314 Physico-mechanical characteristics of the VPR wood chips are reported in Table 1. The moisture 315 content (MC), the calorific values (HHVdaf and HHVar) and the granulometric classifications 316 satisfy the requirement described in the EN ISO 17225-4:2014 [33] for type B wood chip samples 317 but also for A1 and A2 classes. In fact the MC of the three wood chip samples, obtained thanks to 318 the covering of the pile of VPRs with a breathable paper for 7 months, is $\leq 15\%$ (see "Table 2 — Specification of graded wood chips" of the EN ISO [33]), the calorific value of the wood chips is 319 320 not set by the standard (as minimum value to be stated), while the particle size distribution P31S 321 with 3.8% of fine fraction content is one of the classes included in the "Table 1 — Particle size of 322 graded wood chips" of the EN ISO standard [33]. Even if all of the above classification parameters 323 corresponded to the requirements for a class A graded wood chips, VPRs cannot be qualified as 324 type A due to their origin (not included in the list of those allowed for a type A samples) and their 325 ash content that is > 3% (Table 1), the limit set in the EN ISO standard [33] for type B wood chips. 326 Due to their ash content, VPRs cannot be burned in domestic appliances. In any case, previous 327 studies showed that maintaining an efficient steady-state combustion in domestic appliances is 328 challenging with VPRs both as wood chips and pellets [18,19]. Indeed, as demonstrated in a 329 previous study on the combustion of VPR pellets in a domestic stove, the high amount of ashes 330 produced during combustion can compromise the combustion itself because burning VPRs requires 331 a larger amount of injected air which in turn means a colder combustion chamber and higher 332 pollutant emissions [19].

Table 1. Average values (and standard deviations) of the physico-mechanical characteristics of chips from vineyard pruning residues (VPRs) according to in force standards (MC: Moisture Content; HHVdaf: High Heating Value on a dry basis and ash free; HHVar: High Heating Value as received; A: Ash content; P: particle size distribution; F: fine fraction). The three samples were collected from a covered pile after 7-month seasoning, before the first combustion test.

Wood chips Sample #	MC (%)	HHVdaf (MJ kg ⁻¹)	HHVar (MJ kg ⁻¹)	A (%)	P class
1	11.7 (0.04)	18.07 (0.04)	17.57 (0.04)	3.10 (0.09)	D215
2	11.5 (0.03) 11.6 (0.07)	18.04 (0.04) 18.01 (0.01)	17.59 (0.03)	3.07 (0.15)	(F: 3.8%)
3			17.58 (0.01)	3.08 (0.03)	

340 Furthermore, the concentrations of metals measured in the ashes (Table 2) produced during the combustion tests showed a very high amount of Cu, 274±72 mg kg⁻¹, above the limit value of 10 341 mg kg⁻¹ for combustion in domestic appliances [33]. In addition to Cu, other elements such as As, 342 343 Cr, Ni, Zn far exceed the limits imposed by the legislation for use in household appliances [33]. 344 Nevertheless, large scale combustion facilities are equipped with abatement technologies for fly 345 ashes, like cyclones [50] or bag house filters [51], and so, for this reason, the high metal content would not necessarily constitute a problem for atmospheric emissions. Concerning the possibility of 346 347 re-using bottom ashes as soil amendments, in Italy there are not guidelines for the application of 348 biomass ashes on agricultural fields. For this reason, trace element content has been compared with 349 limits set in other European countries [52-54], such as Austria, Germany, Denmark, Sweden, and 350 Finland (Table 2). For all elements, concentrations in VPR ashes were below all limits set by the 351 aforementioned countries with the exception of Cu for which the amount exceeded limit values of 352 some countries, namely Germany and Austria, but not others. Therefore, combustion ashes may 353 need to be disposed as waste rather than be re-utilized in the agricultural fields as fertilizer.

354

355Table 2. Measured concentrations (and standard deviations) of metals in the ashes collected from the wood356boilers at the end of each experiment and range of limit values set in different European countries by current357legislation (where available) for re-use of ashes as fertilizers in agricultural fields.

Measured concentrations Range of limit values

	(standard deviations)	in EU countries ^a
	(mg kg ⁻¹)	(mg kg ⁻¹)
Fe	$15.1 (4.7) \cdot 10^3$	
Al	8.3 (2.6) ·10 ³	
Mn	1000 (22)	
Cu	274 (72)	70-600
Zn	130 (70)	1000-7000
Cr	34.7 (8.7)	100-300
Ni	24.4 (5.7)	30-100
V	22.0 (8.2)	70-100
Pb	5.9 (1.1)	100-300
Co	4.9 (1.7)	
As	4.7 (1.9)	20-40
Mo	2.60 (0.87)	
Sn	1.22 (0.60)	
Cd	0.060 (0.068)	1.5-8
W	0.08 (0.10)	
^a Rang	e of limit values in Germany,	Austria, Denmark, Sweden

and Finland for ashes to be re-used in agricultural soils.

358 3.2 Combustion tests and characterization of the emissions

359 While VPRs cannot be burned in domestic appliances, combustion tests performed in this study 360 showed that VPRs can be used as an energy source in medium-size wood boilers (experimental 361 boiler used had a maximum power 500 kW). In normal operating conditions, with automatic 362 control/adjustment of combustion parameters, the combustion was stable throughout the entire duration of the experiments (up to 10 h). Typical conditions were characterized by a combustion 363 364 chamber temperature of 500-750°C, exhaust gas temperature (after the heat-exchange unit) of 180-240°C, O₂ and CO₂ concentrations in the exhaust gas of 12-13% and 8-9%, respectively, providing 365 a combustion efficiency of 86-89% with 12-20 kg h^{-1} of fuel consumption (Figure 2). 366



Figure 2. Evolution of the concentrations of macro-pollutants, air and fume temperatures during the combustion
 test performed on the 21st December 2012 as an example of typical boiler running conditions. Dashed red squares
 indicates a period of low-energy demand triggering a stand-by phase of the boiler and the shut-down period.

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373 Concerning macro-pollutants, their concentrations in the exhaust air were well below limit values of 374 current legislation (Italian Legislative Decree, D.Lgs. 152/2006 [55] and EU standards for 375 machineries [56]) for most of them, especially NO_x due to relatively low temperatures in the 376 combustion chamber, and SO₂ due to low S content in the fuel (Table 3). CO concentrations were well below the limit values for most of the duration of the tests (steady-state phase). CO 377 378 concentrations increased significantly only in non-optimal combustion conditions encountered 379 during start-up, shut-down and low energy demand phases (e.g., the central heating system of the factory was switched-off, therefore not dissipating the energy produced and triggering a stand-by 380 381 phase of the biomass boiler; Figure 2). Conversely, TSP concentrations were always above limit 382 values (Table 3). It is worthwhile to notice that the experimental boiler used was not equipped with

emission abatement devices, which are indeed mandatory to use in commercial wood boilers of similar sizes. Therefore, the problem of particle emissions would be dramatic only for small size/domestic boilers in which such devices are not deployed. In any case, such small size/domestic appliances are not fit for purpose, as evidenced also in previous studies, because of difficulties in maintaining stable combustion conditions due to excessive production of fly ashes with the fuel under study in both wood chip and pellet forms [18,19].

389 Compared with other types of woody biomasses, emissions of macro-pollutants were higher than 390 emissions from combustion of beech as log woods in biomass boilers for CO (780±410 vs. 15 mg MJ⁻¹ for VPRs and beech, respectively), NO_x (650±200 vs. 157 mg MJ⁻¹ for VPRs and beech, 391 respectively) and TSP (1200±800 vs. 18 mg MJ⁻¹ for VPRs and beech, respectively) but lower for 392 SO₂ (2.6±4.5 vs. 10 mg MJ⁻¹ for VPRs and beech, respectively) [57]. Emissions of macro-pollutants 393 were significantly higher than emissions from combustion of spruce in the form of wood chips (CO 394 75 mg MJ⁻¹, NO_x 127 mg MJ⁻¹, TSP 32 mg MJ⁻¹) and pellets (CO 17 mg MJ⁻¹, NO_x 81 mg MJ⁻¹, 395 TSP 14 mg MJ⁻¹) except for SO₂ (2.6±4.5 vs. 8 mg MJ⁻¹ for VPRs as wood chips and spruce as 396 397 pellets, respectively) [57].

398

Table 3. Average concentrations (and standard deviations) of macro-pollutants in the flue gas, compared with
 the limit values enforced in Italy, and emission factors (flue gas volumetric flow 700-1040 Nm³ h⁻¹) measured
 during the three measurement campaigns. LDL=Lower than Detection Limit.

	Concentra	ations in the flue gas		T :: 4		
	(standard deviations)			Limit values "		
	mg Nm ⁻³	mg Nm ⁻³ O ₂ 11%	g h ⁻¹	g kg ⁻¹	g MJ ⁻¹	mg Nm ⁻³ O ₂ 11%
CO ^b	200 (100)	270 (150)	200 (100)	12.3 (6.5)	0.78 (0.41)	350
NO	143 (16)	184.3 (1.5)	169 (51)	10.3 (3.1)	0.65 (0.20)	
NO_2	LDL °	LDL	LDL	LDL	LDL	
$NO_x {}^d$	220 (24)	283.0 (3.0)	259 (78)	15.8 (4.8)	1.00 (0.30)	500
SO_2	0.7 (1.2)	0.7 (1.2)	0.7 (1.2)	0.041 (0.071)	0.0026 (0.0045)	200
H_2S	LDL ^e	LDL	LDL	LDL	LDL	
C_3H_8	71 (53)	95 (76)	76 (51)	4.6 (3.1)	0.29 (0.20)	
TSP $^{\rm f}$	330 (160)	450 (270)	310 (220)	19 (13)	1.2 (0.8)	100

^a Italian Legislative Decree D.Lgs. 152/2006 for boilers of > 150 kW and ≤ 3 MW, more restrictive than the EU

standards for boilers of \leq 500 kW: CO 1200 mg Nm⁻³, TSP 150 mg Nm⁻³ [56]; ^b Average of 2nd and 3rd experiments; ^c LDL (NO₂) = 5 ppm; ^d Calculated based on the molecular mass of NO₂; ^e LDL(H₂S) = 5 ppm; ^f Average of 1st and 2nd experiments.

402

403 **3.3 Emissions of micro-pollutants**

404 In the perspective of a sustainable management of vineyard practices, the environmental impact of 405 using VPRs as energy source needs to be evaluated. After estimating the performances of the proposed analytical method, determination of PAHs, PCBs and PCDD/Fs was carried out on 406 407 samples coming from the three different combustion tests at the wood boiler (approximate sampling 408 volume of 1.5-3 Nm³). For each sample, three different matrices were collected: fly ash deposited on QFFs, condensed and flue gas adsorbed on XAD[®] resins. The distribution pattern of the different 409 410 analytes was also investigated. Finally, quality standards and limits set by the Italian regulations 411 were used to verify if and when emissions exceeded limit values for emissions in the atmosphere (Italian Legislative Decree, D.Lgs. 152/06 [55]), and the much more restrictive limit values for 412 413 emissions from waste incinerators (Italian Legislative Decree, D.Lgs. 46/2014 [58]).

The combustion tests have highlighted some critical issues that may arise when using VPRs for energy conversion, especially concerning the difficulties in selecting and cleaning the material used as fuel. Indeed, mixed together with the VPR wood chips, plastic residues were found deriving from vineyard practices such as transport bags and shelters (plastic pipes for the protection of the base of the vine). In the specific case of the study, it has been estimated that the plastic residues were below 0.5%. This fact may result in a supply of chlorine at high enough quantity to allow the occasional formation of PCDD/Fs during combustion.

421 Measured PAH and PCB concentrations for each sampled matrix are reported in Table 4 as 422 averages and in Table S7 as single values for each experiment at the wood boiler. The most 423 abundant contaminants in the analyzed samples were, as expected, PAHs followed by PCBs. 424 Conversely, PCDD/Fs were always below detection limits.

- 426 Table 4. Persistent Organic Pollutants (POPs) analyzed in three different combustion tests. Concentration levels
- 427 as averages (and standard deviations) of the three experiments for polycyclic aromatic hydrocarbons (PAHs; as
- 428 single compounds and total, ng Nm⁻³) and polychlorinated biphenyls (PCBs; as chlorination classes and total, pg
- 429 Nm⁻³) for each matrix analyzed. LDL = Lower than Detection Limit, n.a. = not analyzed.

POPs		Concentrations in the flue gas				Concentrations in the flue gas at O ₂ 11%			
		(standard deviations)			(standard deviations)				
			<u>C</u> ^b	<u>R</u> °	<u>Total</u>	<u>F</u>	<u>C</u>	<u>R</u>	<u>Total</u>
	Naphthalene	10.63	101 (65)	2010	2120	13.8	128	2630	2770
		(0.84)	101 (05)	(500)	(520)	(1.7)	(80)	(800)	(810)
	Acenaphthylene	2.1 (1.8)	86 (88)	1060	1150	2.8 (2.3)	110	1400	1500
				(830)	(920)		(120)	(1100)	(1200)
	Acenaphthene	0.35	20(17)	4.7	7.1 (3.7)	0.44	2.4	6.0	8.8 (3.7)
		(0.24)	2.0 (1.7)	(1.8)		(0.25)	(2.0)	(1.6)	
	Fluorene	0.44	87(83)	17.0	26 (14)	0.56	5	21.7	22 (17)
	Thuorene	(0.20)	0.7 (0.3)	(5.5)		(0.22)	11(11)	(6.3)	55 (17)
	Phenanthrene	17 (13)	440 (240)	780	1230	22 (17)	550	1020	1590
	Thenantinene	17 (13)		(540)	(730)	22 (17)	(290)	(720)	(960)
	Anthracene	1.74	24 (11)	45 (27)	71 (28)	2.2 (1.2)	30 (12)	59 (37)	91 (38)
		(0.91)	720 (600)	200	1200	110	040	510	1600
	Fluoranthene	80 (110)		(590)	(1300)	(150)	(790)	(770)	(1700)
	Pyrene	130	1000	530	(1300)	(130)	(7)0)	700	2200
		(170)	(1000)	(850)	(2000)	(220)	(1300)	(1100)	(2600)
PAHs (ng Nm ⁻	Benz[a]anthracene	(170) 13.5 (8.1)	30.2 (4.1)	0.65	44 (12)	(220)	39.4	0.85	58 (20)
				(0.73)		18 (11)	(8.9)	(0.95)	
³)	Chrysene	(0.1)	85 (18)	2.0			110	2.6	158 (60)
		34 (27)		(2.0)	121 (43)	46 (36)	(23)	(2.7)	
	Benzo[b]fluoranthe	19 (18)	25 (17)	0.70				0.89	58 (46)
				(0.64)	44 (35)	25 (23)	32 (22)	(0.84)	
	Benzo[k]fluoranthe	4.8 (4.2)	11.4 (3.6)	0.113	16.3 (5.9)	6.3 (5.5)	15.0	0.145	
				(0.071)			(6.2)	(0.092)	21.5 (8.9)
		9.6 (5.3)	10.4 (5.8)	2.21	22 (11)	12.7	13.7	2.83	00 (15)
	Benzo[a]pyrene			(0.56)		(7.2)	(7.8)	(0.48)	29 (15)
	Benzo[ghi]perylen	10.5	8.3 (6.5)	0.038		13.3	11.0	0.044	24 (14)
	e	(6.6)		(0.066)	(8.4)	(8.7)	(0.076)	24 (14)	
	Indeno[1,2,3-	(5 (70)	20 (16)	0.10	86 (88)	85 (95)	27 (22)	0.11	112 (115)
	cd]pyrene	65 (72)	20 (16)	(0.17)			27 (22)	(0.20)	
	Dibenz[a,h]anthrac	IDI	IDI	IDI	IDI	IDI	IDI	IDI	IDI
	ene	LDL	LDL	LDL	LDL	LDL	LDL	LDL	LDL
	Total (Σ_{16} PAHs)	400	2600	4800	<u>7800</u>	530	3300	6300	<u>10200</u>
		(430)	(2000)	(3300)	<u>(5700)</u>	(570)	(2600)	(4400)	<u>(7500)</u>
PCBs	Cl-1	70 (120)	2200	n.a.	2300	90 (150)	2600	n.a.	2700
(pg Nm ⁻			(2800)		(2800)		(3200)		(3100)
3)	Cl-2	300	3400	n.a.	3600	400	4100	n.a.	4500

-		(470)	(3800)		(3700)	(620)	(4400)		(4300)
	Cl-3	210	2600	n.a.	2800	250	3200	n.a.	3400
		(320)	(2100)		(2300)	(370)	(2500)		(2700)
	Cl-4	110	2800	n.a.	3000	150	3600	n.a.	3800
		(140)	(2900)		(3000)	(190)	(3700)		(3900)
	C1 5	52 (00)	4300	n.a.	4300	70 (120)	5400	n.a.	5500
	CI-5	33 (88)	(3900)		(4000)		(5100)		(5200)
	Cl-6	46 (70)	13000	n.a.	13000	60 (100)	17000	n.a.	17000
		40 (79)	(13000)		(13000)		(17000)		(17000)
	Cl-7	00 (150)	9800	n.a.	9800	110 (190)	12000	n.a.	12000
		90 (150)	(8400)		(8500)		(11000)		(11000)
	Cl-8	LDL	910 (640)	n.a.	910	LDL	1140	n.a.	1140
					(640)		(800)		(800)
	Cl-9	0.53) LDL	n.a.	0.53	$0 \in (1 \ 1)$	IDI		0.6(1.1)
		(0.91)			(0.91)	0.0 (1.1)	LDL	II.a.	0.0 (1.1)
	Cl-10	LDL	18 (17)	n.a.	18 (17)	LDL	22 (20)	n.a.	22 (20)
	$T_{-4-1}(\Sigma - DCD)$	880	39000	n.a.	<u>40000</u>	1100 (1200)	49000	n.a.	<u>50000</u>
	1 otal (Σ_{127} PCBs)	(900)	(32000)		<u>(32000)</u>		(40000)		<u>(42000)</u>

430 ^a Fly ash deposited on quartz fiber filters; ^b Condensate; ^c Flue gas adsorbed on XAD[®] resins.

432 3.3.1 PAHs

The detected concentrations of PAHs ranged widely, from 4.13 to 14.3 µg Nm⁻³ as sum of all PAHs 433 (with an average value of 7.81 µg Nm⁻³). Nevertheless, the first two sampling campaigns showed a 434 435 moderate variability, with a RSD of 13%. All the samples show a distinct pattern of PAH 436 compounds in their distribution, generally enriched in Pyrene, Phenanthrene, Fluoranthene and 437 Acenaphthylene. The predominant compound is Naphthalene, with values ranging from 1.62 to 2.66 μ g Nm⁻³ (with an average value of 2.12 μ g Nm⁻³), except for the third sampling campaign where 438 439 other heavier compounds dominated (*i.e.* Pyrene and Fluoranthene). Dibenz[a,h]anthracene was not 440 detected in any of the samples, whereas Benzo[ghi]perylene and Indeno[1,2,3-cd]pyrene were 441 detected only in the resin sample of the first sampling campaign.

As regards the estimation of carcinogenic potency and of degree of toxicity of these compounds, the 442 443 analyzed samples present a pattern enriched of PAHs with a low value of Toxicity Equivalence 444 Factor (TEF) [59]. Conversely, Benzo[a]pyrene and Dibenz[a,h]anthracene, that are generally 445 considered the most toxic and carcinogenic among the 16 U.S. Environmental Protection Agency 446 priority PAHs [60], showed levels of concentration respectively two order of magnitude lower (from 12.9 to 33.7 ng Nm⁻³) or always below the IDL. 447

The distribution pattern of compounds in the three matrices was rather similar and unvaried, as 448 449 shown in Figure 3. The highest levels of PAHs were detected in resin samples (with an average value of 4.84 µg Nm⁻³) and were mainly ascribable to Naphthalene and Acenaphthylene, which 450 451 jointly accounted on average for 70% of the total; slightly lower PAH concentrations were found in condensate (with an average value of 2.57 µg Nm⁻³), while the values for fly ash samples were 452 453 comprised between 0.104 and 0.901 µg Nm⁻³ as sum of all PAHs.





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458 When each single compound is considered, the distribution pattern remains relatively constant and showed a different matrix subdivision as the molecular weight of the analytes changes (Figure 4). 459 460 The lightest compounds (Naphthalene, Acenaphthylene, Acenaphthene, Fluorene, Phenanthrene and Anthracene) are more volatile and, hence, they were adsorbed on the resin. Conversely, fly ash 461 462 deposited on QFFs were generally enriched in high molecular weight compounds (4 to 6 aromatic rings). This is consistent with the physical-chemical properties of PAHs, in particular with the 463 decrease in vapor pressure with increasing molecular weight. 464



Figure 4. Average percentage distribution of concentration levels of single PAH compound analyzed in the three
different matrices for each sampling campaign. NAPH: Naphthalene; ACNY: Acenaphthylene; ACN:
Acenaphthene; FLU: Fluorene; PHE: Phenanthrene; ANT: Anthracene; FLT: Fluoranthene; PYR: Pyrene;
BAA: Benz[a]anthracene; CRY: Chrysene; BBF: Benzo[b]fluoranthene; BKF: Benzo[k]fluoranthene; BAP:
Benzo[a]pyrene; BPER: Benzo[ghi]perylene; IPYR: Indeno[1,2,3-cd]pyrene.

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The detected concentrations are significantly lower than the threshold values indicated by the legislation for large combustion plants (0.1 mg Nm⁻³ for single compound, 150 mg Nm⁻³ for Naphthalene, Italian Legislative Decree, D.Lgs. 152/06 [55]), and for the more restrictive ones related to waste incinerators (10 μ g Nm⁻³ as the sum of 11 compounds, Italian Legislative Decree, D.Lgs. 46/2014 [58]).

478 3.3.2 PCBs

Samples were analyzed for 127 of the 209 possible congeners of PCBs (namely for those related to the Aroclor commercial mixtures and the non-Aroclor PCB-11). Not all of them were found in all the samples analyzed: for instance, Octa- and Deca-chlorinated compounds were not detectable in any of the filter samples, while Nona-chlorinated compounds were always below the detection limits in condensates, as reported in Table 4. As previously explained, due to high background values of PCBs found in the commercial XAD[®] resin, it was not possible to determine the concentration of this class of analytes in the resin samples. As reported for PAHs, the levels of concentration of PCBs ranged widely, from 4.91 to 68.8 ng Nm⁻ ³ (average value of 40.1 ng Nm⁻³) with the second sampling campaign one order of magnitude lower than the others. The congeners composition is enriched in compounds with a high degree of chlorination: Hexa- and Hepta- represent, on average, 56% of the total distribution.

Also in this case, the distribution pattern of congeners in the two matrices under analysis is rather
constant. PCBs are preferentially distributed on the condensate (which represents 97.3-98.6% of the
PCB contamination), rather than on the filters where many homologue groups were below the
detection limits.

The obtained results are considerably lower than the threshold limit values for PCB emissions from both large combustion plants (0.5 mg Nm⁻³, Italian Legislative Decree, D.Lgs. 152/06) and waste incinerators (Italian Legislative Decree, D.Lgs. 46/2014).

497 3.3.3 PCDD/Fs

498 As anticipated, concentrations of PCDD/Fs were all below detection limits, therefore no particular 499 concern was evidenced. This fact could be ascribed to the low percentage of chlorine present in 500 VPR wood chips (about 8‰). A further explanation may derive from insufficient sampling volumes 501 for the determination of compounds usually present at very low concentrations like PCDD/Fs. 502 Nevertheless, we have still done an evaluation of the potential overrun of the limits imposed by the different legislations (10 µg Nm⁻³ for Italian Legislative Decree, D.Lgs. 152/06 and 0.1 ng Nm⁻³ for 503 504 Italian Legislative Decree, D.Lgs. 46/2014) comparing them with our highest instrumental detection limit value (30 pg Nm⁻³) and using International Toxicity Equivalence Factors (I-TEFs) defined by 505 506 Van den Berg et al. [61]. Even our highest instrumental detection limit values are significantly 507 lower than the limits imposed by the Italian legislation, indicating that the use of VPRs as biofuel 508 does not present risks of POPs contamination.

509 4 Conclusions

510 In this study, we assessed the viability and sustainability of using VPRs, as wood chips, for energy 511 conversion. Wood chips from VPRs were characterized as fuel in terms of chemical composition, 512 calorific value, ash content and humidity. Results showed that VPRs comply with type B standards,

513 however due to high ash content they are not suitable for burning in domestic appliances.

514 Combustion tests were performed in a medium-size biomass boiler (maximum power 500 kW) to 515 assess the viability of the approach in terms of sustainable steady-state combustion. We found that 516 VPRs can be used successfully in medium-size boilers and the combustion was stable for all the 517 duration of the experiments (~10 h) with a combustion efficiency of 86-89%.

518 Primary emissions of both macro- and micro-pollutants were measured to assess the environmental 519 impact. Primary emissions of macro-pollutants, such as CO₂, CO, NO_x, and SO₂, were below limit 520 values of current European legislation except for PM emissions that were above the limits in all 521 combustion tests. It is worth noting that medium- and large-size biomass boilers need to be 522 equipped with abatement technologies for particle emissions, not used in this series of experiments, 523 therefore this potential environmental drawback would only be relevant for small-size domestic 524 appliances that cannot anyhow be used with this type of fuel. Concerning micro-pollutants, both 525 PAH and PCB concentrations in the particulate, condensate and volatile fractions collected from the 526 flue were well below limit values set for large scale biomass boilers and also for incinerators, for 527 which limits are much more restrictive. PCDD/Fs were always below detection limits of our method 528 and therefore combustion of VPRs does not pose risks for production of POPs.

529 Bottom ashes produced during combustion were also characterized to assess whether they can be 530 recycled in the vineyard as soil amendments/fertilizers. Heavy metal content was compared with 531 limit values set in other European countries as Italy does not have limit values for ashes distributed 532 in the field. All trace metals were below limit values except for copper, whose concentration 533 exceeded limit values for some countries but not for others, therefore bottom ashes may need to be 534 disposed as waste rather than used as soil amendments in agricultural fields. The present study was based on a relatively homogeneous VPR sample with processing and storage conditions that respect the local good practices. The use of wetter wood chips, wood chips with a high percentage of fine particles or wood chips treated with chlorinated pesticides could influence negatively the quality of emissions, especially in small- and medium-sized installations such as the one used for this series of tests.

540 The analytical method developed for the determination of micro-pollutants, *i.e.* PAHs, PCBs and 541 PCDD/Fs, is fit-for-purpose to assess risks posed by emissions of compounds that are harmful for 542 public health. Further improvements of the analytical method may be done by testing the use of 543 polyurethane foam rather than XAD[®] resins for sampling the volatile components – because 544 background levels of PCBs in the XAD[®] resins were very high –, and by increasing sampling 545 volume to obtain a more accurate evaluation of PCDD/F concentrations.

546 While our study demonstrated the viability and sustainability of using VPRs for energy conversion, 547 more studies are needed to assess both the potential for aerosol formation in the atmosphere from 548 oxidation of emitted volatile organic compounds and whether additional abatement strategies would 549 be required.

550

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558 **Declaration of competing interests**

559 MS is an employee of Uniconfort S.r.l., producer of biomass boilers who fully collaborated in the 560 present study. The producer did not provide funding for the research activity; had no role in the 561 design of the study; in the collection, analyses or interpretation of data; in the writing of the 562 manuscript, or in the decision to publish the results.

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